

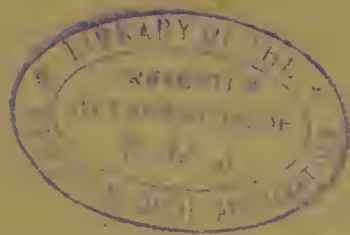
37/4



22102060764

Med

K15489



WORKS BY
CHARLES F. BOLDUAN, M.D.
PUBLISHED BY
JOHN WILEY & SONS

Immune Sera.

A concise exposition of the main facts and theories of infection and immunity. Fourth edition, rewritten. By Charles F. Bolduan, M.D. 12mo, ix + 226 pages. Cloth, \$1.50.

TRANSLATIONS.

The Suppression of Tuberculosis.

Together with Observations concerning Phthisiogenesis in Man and Animals, and Suggestions concerning the Hygiene of Cow Stables and the Production of Milk for Infant Feeding, with Special Reference to Tuberculosis. By Professor E. von Behring, University of Marburg. Authorized Translation by Charles F. Bolduan, M.D. 12mo, vi + 85 pages. Cloth, \$1.00.

Manual of Serum Diagnosis.

By Doctor O. Rostoski, University of Wurzburg. Authorized Translation by Charles F. Bolduan, M.D. 12mo, vi + 86 pages. Cloth, \$1.00.

Studies in Immunity.

By Professor Paul Ehrlich. Translated by Charles F. Bolduan, M.D. Second Edition, Revised and Enlarged. 8vo, xi + 712 pages. Cloth, \$6.00.

IMMUNE SERA

A CONCISE EXPOSITION OF THE
MAIN FACTS AND THEORIES OF

INFECTION AND IMMUNITY

BY

DR. CHARLES FREDERICK BOLDUAN

BACTERIOLOGIST, RESEARCH LABORATORY, DEPARTMENT OF HEALTH
CITY OF NEW YORK

FOURTH EDITION, REWRITTEN AND ENLARGED

FIRST THOUSAND

NEW YORK:

JOHN WILEY & SONS

LONDON: CHAPMAN & HALL, LIMITED

1911

COPYRIGHT, 1907, 1908, 1911,
BY CHARLES FREDERICK BOLDUAN

WELLCOME INSTITUTE LIBRARY	
Coll.	welMOMec
Call	
No.	

THE SCIENTIFIC PRESS
ROBERT DRUMMOND AND COMPANY
BROOKLYN, N. Y.

TO

HERMANN M. BIGGS, M.D., LL.D.

PIONEER IN SPECIFIC SERUM THERAPY AND SERUM
DIAGNOSIS IN AMERICA, FOUNDER OF THE
FIRST MUNICIPAL SERUM LABORATORY
IN THE WORLD, THIS VOLUME
IS DEDICATED AS A TOKEN
OF GREAT ADMIRATION
AND ESTEEM

PREFACE TO THE FOURTH EDITION

THE favorable reception accorded to the previous editions of this book, together with the fact that our knowledge of the subject has increased considerably in the past few years, has led the author to undertake a thorough revision of the work.

In its first edition, in 1904, this book dealt only with certain antibodies whose discovery had aroused a great deal of scientific interest, namely, hæmoly-sins, cytotoxins, and precipitins. To this was added, in subsequent editions, a discussion of anti-toxins, agglutinins, and opsonins. All these topics were naturally embraced under the title "Immune Sera." In the present (fourth) edition, while the old title has been retained, the scope of the subject matter has been greatly extended, so that now there is presented an exposition of the main facts of infection and immunity.

It is but natural that any discussion of the immunity reactions should center about the ingenious side-chain theory of Ehrlich, which has dominated the work in this field. Its heuristic value

has unquestionably been very great. At the same time it cannot be doubted that some of the deductions from the theory have led, here and there, to strained conceptions which apparently violate established biological facts. While presenting Ehrlich's views at length, therefore, the author has endeavored to bring out clearly just why and wherein certain other investigators differ. The aim of the book has been to present a broad, clear outline of the main facts and theories concerning infection and immunity, and while this may perhaps have led to the omission of some really excellent studies, it was felt best not to confuse the beginner with a mass of apparently contradictory observations.

CHARLES BOLDUAN.

NEW YORK, *April*, 1911.

CONTENTS

	PAGE
Antitoxins.....	1
HISTORICAL.....	1
PRESENT METHOD OF PRODUCING DIPHThERIA ANTI- TOXIN.....	2
PRODUCTION OF DIPHThERIA TOXIN.....	2
IMMUNIZING THE ANIMALS.....	3
COLLECTING THE SERUM.....	4
TESTING THE STRENGTH OF THE SERUM.....	5
EHRlich's THEORY FOR PRODUCTION.....	6
TOXINS, TOXOIDS.....	6
RECEPTORS.....	9
WEIGERT'S OVERPRODUCTION THEORY.....	10
EXPERIMENTAL EVIDENCE FOR EHRlich's THEORY....	13
ANTIGENS OR HAPTINS.....	16
NATURE OF ANTITOXINS IN GENERAL.....	17
TOXINS AND OTHER POISONOUS CELL DERIVATIVES IN GENERAL.....	20
RELATIONS BETWEEN TOXIN AND ANTITOXIN.....	22
"L ₀ " and "L ₊ ".....	23
PARTIAL SATURATION METHOD OF STUDYING TOXINS —TOXONS, TOXOIDS.....	23
EHRlich's "POISON SPECTRA".....	24
VIEWS OF ARRHENIUS, BORDET, AND OTHERS.....	28
Agglutinins.....	32
THE PHENOMENON.....	32
HÆMAGGLUTININS.....	34
ISOAGGLUTININS.....	35
PURPOSE OF AGGLUTINATION.....	36
HISTORICAL.....	37
PFAUNDLER'S REACTION (THREAD REACTION).....	38

	PAGE
NATURE OF AGGLUTININS AND OF THE AGGLUTINATION REACTION.....	39
AGGLUTINOIDS.....	39
GROUP AGGLUTININS:.....	43
ABSORPTION METHODS FOR DIFFERENTIATING BETWEEN A MIXED AND A SINGLE INFECTION.....	46
FORMATION OF AGGLUTININS ACCORDING TO THE SIDE- CHAIN THEORY, RECEPTORS OF FIRST, SECOND, AND THIRD ORDER.....	47
Bacteriolysins and Hæmolysins.....	50
HISTORICAL.....	50
FEHLIFFER'S PHENOMENON.....	51
HÆMOLYSIS.....	52
NATURE OF HÆMOLYTIC SERA.....	54
ANALOGY BETWEEN THE BACTERIOLYTIC AND HÆMO- LYTIC PROCESSES.....	57
EHRLICH AND MORGENROTH ON THE NATURE OF HÆMO- LYSIS.....	58
THEIR THREE CLASSIC EXPERIMENTS.....	59
NOMENCLATURE.....	63
ROLE OF THE IMMUNE BODY.....	64
ON WHAT THE SPECIFICITY DEPENDS.....	65
DIFFERENCE BETWEEN A SPECIFIC SERUM AND A NOR- MAL ONE.....	66
DIVERGING VIEWS OF EHRLICH AND BORDET.....	67
THE SIDE-CHAIN THEORY APPLIED TO THESE BODIES..	68
MULTIPLICITY OF COMPLEMENTS.....	70
THE BORDET-GENGOU PHENOMENON; NEISSER- SACHS BLOOD TEST.....	71
NORMAL SERUM, ITS HÆMOLYTIC AND BACTERIOLYTIC ACTION.....	73
ACTIVE AND INACTIVE NORMAL SERUM.....	75
ACTION NOT ENTIRELY SPECIFIC.....	77
MULTIPLICITY OF THE ACTIVE SUBSTANCES.....	78
DIFFERENCE BETWEEN A NORMAL AND A SPECIFIC IM- MUNE SERUM.....	79
NATURE OF THE IMMUNE BODY—PARTIAL IMMUNE BODIES OF EHRLICH.....	82

	PAGE
METCHNIKOFF'S VIEWS.....	84
SUPPORT FOR EHRLICH'S VIEW.....	85
ANTHÆMOLYSINS: THEIR NATURE—ANTI-COMPLEMENT OR ANTI-IMMUNE BODY.....	86
ANTI-COMPLEMENT.....	88
FLUCTUATIONS IN THE AMOUNT OF THE ACTIVE SUB- STANCES IN SERUM.....	91
SOURCE OF THE COMPLEMENTS—LEUCOCYTES AS A SOURCE—OTHER SOURCES.....	93
STRUCTURE OF THE COMPLEMENTS—COMPLEMENTOIDS...	94
ISOLYSINS—AUTOLYSINS—ANTI-ISOLYSINS.....	97
DEFLECTION OF COMPLEMENT.....	100
PRACTICAL VALUE OF ANTI-BACTERIAL SERA.....	104
Precipitins.....	107
DEFINITIONS.....	107
BACTERIAL PRECIPITINS.....	108
LACTOSERUM—OTHER SPECIFIC PRECIPITINS.....	108
SPECIFICITY OF THE PRECIPITINS.....	109
NATURE OF THE PRECIPITINS.....	111
PRACTICAL APPLICATION.....	112
THE WASSERMANN-UHLENHUTH BLOOD TEST.....	113
IMMUNIZING THE ANIMALS.....	114
COLLECTING THE SERUM.....	115
THE TEST.....	116
APPEARANCE OF THE REACTION.....	117
DELICACY OF THE PRECIPITIN TEST.....	118
OTHER APPLICATIONS OF THE PRECIPITIN TEST.....	118
ANTIPRECIPITINS—ISOPRECIPITINS.....	119
Cytotoxins.....	121
DEFINITION, LEUCOTOXIN—NATURE OF THE CYTOTOXIN ANTICYTOTOXIN.....	121
NEUROTOXIN.....	122
SERMOTOXIN.....	123
COMMON RECEPTORS.....	124
CYTOTOXIN FOR EPITHELIUM.....	124
CYTOTOXINS BY THE USE OF NUCLEOPROTEIDS...	125
Opsonins.....	127
HISTORICAL.....	127

	PAGE
BACTERIOTROPIC SUBSTANCES.....	129
OPSONINS DISTINCT ANTIBODIES.....	130
STRUCTURE OF OPSONINS.....	130
THE OPSONIC INDEX.....	131
TECHNIQUE.....	131
VALUE OF THE OPSONIC MEASUREMENTS.....	133
Snake Venoms and their Antisera.....	137
THE VENOMS.....	137
ANTIVENINS.....	139
Anaphylaxis.....	141
HISTORICAL.....	141
THE PHENOMENON.....	142
SERUM RASHES.....	143
THEORIES OF ANAPHYLAXIS.....	144
ALLERGY.....	147
SUPPOSED RELATION TO PRECIPITIN ACTION.....	148
PATHOLOGY OF ANAPHYLACTIC SHOCK.....	150
RELATION OF ANAPHYLAXIS TO SERUM THERAPY.....	151
Infection and Immunity.....	154
INFECTION.....	154
THE INFECTING AGENT.....	154
ANAPHYLATOXIN.....	157
RESISTANCE AGAINST INFECTION.....	158
NATURAL IMMUNITY.....	159
ACQUIRED IMMUNITY.....	160
MECHANISM OF IMMUNITY.....	162
RELATION OF ANAPHYLAXIS TO IMMUNITY.....	163
IMMUNITY REACTION OF THE PART OF BACTERIA.....	165
ATREPSY.....	166
Bacterial Vaccines.....	169
HISTORICAL.....	169
METHODS OF ACTIVE IMMUNIZATION.....	171
TREATMENT WITH VACCINES.....	173
THE VACCINES.....	174
DOSES.....	175
RESULTS.....	176

	PAGE
Leucocyte Extracts in the Treatment of Infections.....	177
THEORY.....	177
PREPARATION OF THE EXTRACTS.....	177
APPLICATION AND RESULTS.....	173
Principles Underlying Treatment of Syphilis with Salvarsan.....	179
PARASITOTROPISM AND ORGANOTROPISM.....	179
CHEMORECEPTORS.....	181
Appendices.....	185
A. THE WASSERMANN TEST FOR SYPHILIS.....	185
B. NOGUCHI'S MODIFIED WASSERMANN REACTION.....	200
C. BLOOD EXAMINATION PREPARATORY TO TRANSFUSION.....	204
D. THE CONGLUTINATION REACTION.....	207
THE MUCH-HOLZMANN COBRA VENOM REACTION...	208
THE MEIOSTAGMIN REACTION.....	208
WEIL'S COBRA VENOM TEST IN SYPHILIS.....	210
ANTITRYPSIN DETERMINATIONS.....	212

IMMUNE SERA

ANTITOXINS

Historical. — The researches of Buchner¹ in 1889 had shown that the serum of animals artificially immunized against a certain bacterium possessed marked bactericidal properties for that particular organism. In studying immunity on animals which had been successfully immunized against diphtheria infection, Behring,² working in Koch's laboratory was struck by the fact that in these animals living virulent diphtheria bacilli were often demonstrable in the scab at the site of injection several weeks after the infection, and furthermore that the blood serum of the animals did not possess bactericidal properties. In a study published in 1890 Behring showed that the serum of rabbits artificially immunized against diphtheria was able to confer a specific immunity against diphtheria infections in other animals. He also demonstrated that such a serum could be used therapeutically to cure an infection already in progress. Such a serum

¹ Buchner, *Centralblatt Bacteriologie*, Vol. v. 1889. *Archiv. f. Hygiene*, Vol. x. 1890.

² Behring & Kitasato, *Deutsche med. Wochenschrift*, No. 49. 1890.

was not bactericidal, and retained its therapeutic power for a considerable time. He believed that the action of the serum was effected by a neutralization of the bacterial toxin by an "antitoxic serum constituent." The action was strictly specific, an antitoxic serum obtained after a diphtheria infection protected only against diphtheria; one derived from a tetanus animal, only against tetanus. Subsequently Behring and Knorr showed that immunization could be effected with bacterial-free filtrates of tetanus cultures and that the serum thus produced protected not only against tetanus infection but against poisoning by the toxic products of the bacilli. After considerable experimental work Behring and his collaborators devised an effective method of immunizing sheep and certain other animals against diphtheria and against tetanus and so produced antitoxic sera in considerable amounts.

The following account taken from Park shows the present methods of producing diphtheria antitoxin.

Production of the Diphtheria Toxin. — A strong diphtheria toxin should be obtained by taking a very virulent culture and growing it in broth which is about 8 cc. normal soda solution per liter above the neutral point to litmus. The culture fluid should be in comparatively thin layers and in large-necked Erlenmeyer flasks, so as to allow of a free access of air; the temperature should be about 35° to 36° C. The culture, after a weeks growth, is removed from the incubator,

and having been tested for purity by microscopic and culture tests is rendered sterile by the addition of 10 per cent of a 5 per cent solution of carbolic acid. After 48 hours the dead bacilli have settled on the bottom of the jar and the clear fluid is filtered through ordinary sterile filter paper and stored in full bottles in a cold place until needed. Its strength is then tested by giving a series of guinea pigs carefully measured amounts. Less than 0.01 cc. when injected hypodermically should kill a 250 gram guinea pig.

Immunizing the Animals. — The horses used should be young, vigorous, of fair size, and absolutely healthy. Vicious habits, such as kicking, etc., make no difference, except, of course, to those who handle the animals. The horses are severally injected with an amount of toxin sufficient to kill five thousand guinea pigs of 250 grams weight (about 20 cc. of strong toxin). After from three to five days, so soon as the fever reaction has subsided, a second subcutaneous injection of a slightly larger dose is given. With the first three injections of toxin 10,000 units of antitoxin are given. If antitoxin is not mixed with the first doses of toxin only one-tenth of the doses advised is to be given. At intervals of from five to eight days increasing injections of pure toxin are made until at the end of two months from ten to twenty times the original amount is given. There is absolutely no way of judging which horses will produce the highest grades of antitoxin. Very roughly those horses which are extremely sensitive, and those which react hardly at all are the poorest, but even here there are exceptions. The only way, therefore, is at the end of six weeks or two months to bleed the horses and test their serum. If only high grade serum is wanted all the horses that give less

than 150 units per cc. are discarded. If moderate grades only are desired, all that yield 100 units may be retained. The retained horses receive steadily increasing doses, the rapidity of the increase and the interval of time between the doses (three days to one week) depending somewhat on the reaction following the injection, an elevation of temperature of more than 3° F. being undesirable. At the end of three months the antitoxic serum of all the horses should contain over 300 units and in about 10 per cent as much as 800 units per cc. Very few horses ever give over 1000 units, and none so far has given as much as 2000 units per cc. The very best horses, if pushed to their limit continue to furnish blood of gradually decreasing strength. If every nine months an interval of three months' freedom from inoculations is given, the best horses furnish high grade serum during their periods of treatment for from two to four years.

Collecting the Serum. — In order to obtain the serum the blood is withdrawn from the jugular vein by means of a sharp-pointed canula which is plunged through the vein wall, a slit having been made in the skin. The blood is carried by a sterile rubber tube attached to the canula, into large Erlenmeyer flasks and allowed to clot, the flasks, however being placed in a slanting position before clotting has commenced. The serum is drawn off after 4 days by means of sterile glass and rubber tubing, and is stored in large flasks in a refrigerator. From this as needed small vials are filled. The vials and their stoppers, as indeed all the utensils used for holding the serum, must be absolutely sterile and every possible precaution must be taken to avoid contamination of the serum. An antiseptic may be added as a preservative, but is not necessary. Diph-

theria antitoxin, when stored in vials and kept in a cool place away from light and air contains within 10 per cent of its original strength for at least two months; after that it can be used by allowing for a maximum deterioration of 3 per cent for each month.

Testing the Strength of the Antitoxin. — This is carried out as follows: Six guinea pigs are injected with mixtures of toxin and antitoxin. In each of the mixtures there is 100 times the amount of a toxin (similar to that adopted as the standard) which will kill a 250 grams on an average in 96 hours. In each of the mixtures the amount of antitoxin varies; for instance, No. 1 would contain 0.002 cc. serum; No. 2, 0.003 cc.; No. 3, 0.004 cc.; No. 4, 0.005 cc., etc. If at the end of the fourth day Nos. 1, 2 and 3 were dead and Nos. 4, 5 and 6 were alive we would consider the serum to contain 200 units of antitoxin for each cubic centimeter. When we mix only ten fatal doses of toxin with one-tenth of the amount of antitoxin used with 100 fatal doses, the guinea pig must remain well. The mixed toxin and antitoxin must remain together for fifteen minutes before injecting.

Behring's publication was followed in the next two years by considerable work along these lines, valuable contributions being made by Aronson,¹ Roux, and Martin,² Wernicke,³ Knorr⁴ and others. The statements of Behring as to the strict specificity of the antitoxins were fully confirmed. Certain

¹ Berliner med. Gesellschaft, Sitzung, Dec. 21, 1892. Also Berliner Klin. Wochenschrift, 1893 and 1894.

² Roux and Martin, Annal. Pasteur 1894.

³ Behring and Wernicke, Zeitsch. Hygiene, 1892. Vol. xi.

Behring and Knorr, Zeitsch. f. Hygiene, 1893. Vol. xii.

observations by Buchner¹ and by Roux and Martin threw doubt, however, on the correctness of Behrings view that the toxin was neutralized by the specific serum just as a base was neutralized by an acid. It was claimed, for example, that the specific serum acted mainly on the body cells causing them to become non-susceptible to the poison in question. Various theories were formulated to account for the production of the antitoxins, their specificity, etc., but of them all only one has at all maintained itself. This, is the so-called side-chain theory, which was formulated by Ehrlich² in 1897.

Ehrlich's Side-Chain Theory. — Originally the side-chain theory was applied by Ehrlich only to the production of the specific *antitoxins*, i.e., substances in the blood, which act not only on the living bacteria, but also and especially on their dissolved toxins. Later on he extended it so as to apply also to the formation of specific bactericidal and hæmolytic substances in the serum of animals treated with living bacteria or with animal cells.

Toxins — Toxoids — Special Function of the Side Chains. — The basis of the theory is the fact that poison and counter-poison, toxin and antitoxin, combine directly in any given quantity. This combination always occurs in definite proportions

¹ Buchner, Münchener med. Wochenschrift, 1894.

² Ehrlich, Klinisches Jahrbuch, 1897.

following the laws of chemical combination; and, still following those laws, is slower at lower temperatures than at higher, stronger in concentrated than in dilute form. Ehrlich could further show that each poison for which by the process of immunizing one can develop a counter-poison possesses two groups which are concerned in the combination with the counter-poison or antitoxin. One of these, the so-called *haptophore group*, is the combining group proper; the other, the *toxophore group*, is the carrier of the poison. A poison molecule, therefore, might lose the one, the toxophore, and still be capable by means of its haptophore group of combining with antitoxin. Such a modified poison, which because of the loss of the toxophore group can hardly be called a poison, but which still possesses the power to combine with antitoxin, Ehrlich calls a *toxoid*. Toxoids may be produced spontaneously in old poisons through decomposition of the poison molecule, or they may be produced artificially by causing certain destructive agents such as heat or chemicals to act on bacterial poisons. The toxophore group is a very delicate one and much more readily decomposed than the combining (haptophore) group. Ehrlich reasoned that in order for a poison to be toxic to an organism, i.e., in order that the toxophore group be able to act destructively on a cell, it is necessary for the haptophore group of the poison to combine with

the cell. "In every living cell," Ehrlich says, "there must exist a dominating body [Leistungs Kern] and a number of other chemical groups or side chains. These groups have the greatest variety of function, but especially those of nutrition and assimilation."

The side chains, then, according to this author,

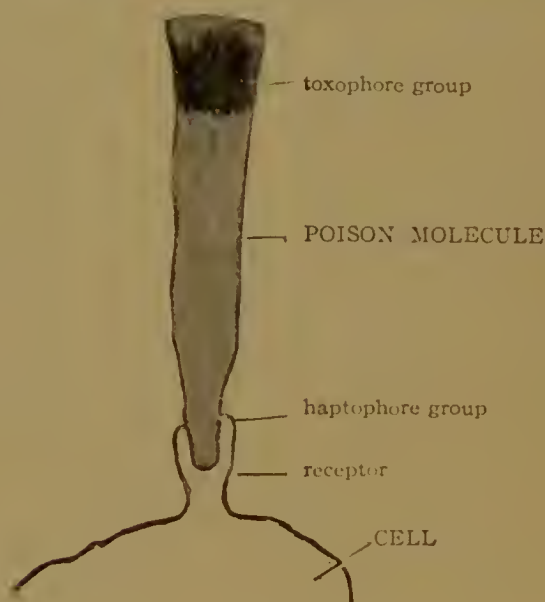


FIG. 1

are able to combine with the greatest variety of foreign substances and convert these into nourishment suitable to the requirements of the active central body. They are comparable to the pseudopodia of the lower animals, which engulf food particles and assimilate the same for the immediate use of the organism. In order that any substance

may combine with these side chains it is necessary that certain very definite relations exist between the combining group of the substance and that of the side chain. Using the well-known simile of Emil Fischer, the relation must be like that of lock and key, i.e., the two groups must fit accurately. Hence not every substance will fit all the side chains of an organism. It will combine only with those for which it possesses a fitting group.

Receptors — Weigert's Overproduction Theory. — This doctrine of the chemistry of the organism's metabolism Ehrlich applied to the action of toxins and antitoxins. "The toxin," he said, "can act only when its haptophore group happens to fit to one of the side chains," or *receptors*, as he now prefers to call them. As a result of this combination, the toxophore group is able to act on the cell and injure it. If we take as an example tetanus, in which all the symptoms are due to the central nervous system, the side-chain theory assumes that the haptophore group of the tetanus poison fits exactly and is combined with the side chain or receptors of the central nervous system. Other experiments, which we will not reproduce here, have shown us unquestionably that the action of the antitoxins depends on the fact that this combines with the haptophore group of the poison and so satisfies the latter's affinity. Ehrlich, therefore, concluded that the antitoxin is nothing else than

the side chains or receptors which are given off by the cells and thrust into the circulation. The way in which these side chains or receptors are thrust off as a result of the immunizing process, Ehrlich explains by means of *Weigert's Overproduction Theory*.

At the meeting of German Naturalists and Physicians held at Frankfurt in 1896, Weigert¹ in discussing regeneration, advanced an hypothesis the essential features of which are that physiological structure and function depend upon the equilibrium of the tissues maintained by virtue of mutual restraint between their component cells; that destruction of a single integer or group of integers of a tissue or a cell removes a corresponding amount of restraint at the point injured, and therefore destroys equilibrium and permits of the abnormal exhibition of bioplastic energies on the part of the remaining uninjured components, which activity may be viewed as a compensating hyperplasia; that hyperplasia is not, therefore, the *direct* result of external irritation, and cannot be, since the action of the irritant is destructive and is confined to the cells or integers of cells that it destroys, but occurs rather indirectly as a function of the surrounding uninjured tissues that have been excited to bioplastic activity through the removal of the restraint

¹ Weigert, *Verhandlungen der Ges. deutscher Naturforscher und Aerzte*, 1896

hitherto exerted by the cells destroyed by the irritant; and, finally, when such bioplastic activity is called into play there is always hypercompensation — i.e. there is more plastic material generated than is necessary to compensate for the loss.

Ehrlich points out that owing to the combination of the toxin with the side chain of a cell, these side chains are practically lost to the cell; that the latter or its fellows now produces new side chains to replace this loss, but that this production always goes so far as to make a surplus of side chains; that these side chains are thrown off by the cell as unnecessary ballast, and then circulate in the blood as antitoxin. The same substances, therefore, which when part of the cell combine with the haptophore group of the toxin, enabling that to act on the cell, when circulating free in the blood combine with and satisfy this haptophore group of the toxin, and prevent the poison from combining with and damaging the cells of the organism.

It does not follow from Ehrlich's theory that the antitoxin is produced by the same set of cells whose injury by the toxin gives rise to the particular clinical symptoms. Thus we might believe that although in tetanus the cells of the central nervous system give rise to the characteristic symptoms, cells entirely apart from these, e.g., in the bone marrow, might be the main source of the antitoxin. The

fact that we appreciate symptoms from only one organ is, obviously, no proof that other tissues have been unaffected.

It may be well here to call attention to another rather common misconception regarding the production of antitoxin, namely that the body cells have to become educated, so to speak, to produce the antitoxin. This, it is believed, is effected by giving gradually increasing doses of toxin. As a matter of fact the reason for this gradual increase in the dose injected is quite different. The object in view is the administration of an enormously large dose of toxin, one that will engage the receptors of many cells. The previous injections have brought about some production of antitoxin and this partially neutralizes some of the toxin injected, making it possible to give a larger dose than before. If one gives at the outset a large amount of toxin, partially neutralized by antitoxin, one will produce an amount of antitoxin equal to that ordinarily obtained in response to the same quantity of unaltered toxin given as the tenth or twentieth injection of a series. Park and Atkinson for example, injected a fresh horse with one litre of a toxin neutralized $1\frac{1}{2}$ times for guinea pigs. At the end of a week the horse had produced a serum containing 60 units per cc. When the toxin was neutralized 6 fold no antitoxin whatever was produced.

Experimental Evidence for Ehrlich's Theory. — According to Ehrlich, then, the formation of specific antibodies must proceed in three stages:

1. The binding of the haptophore group to the receptor.
2. The increased production of the receptors following this binding.
3. The thrusting-off of these increased receptors into the blood.

So far as the first point is concerned Wassermann¹ showed that with tetanus, in which, as is well known, all the symptoms are referable to the central nervous system, tetanus toxin was bound by central nervous system substance in vitro. A mixture of tetanus poison and normal central nervous system was innocuous to animals, showing that certain substances present in the central nervous system combine with and thus satisfy the affinity of the haptophore group of the poison. This of course prevents the latter from combining with any cells of the organism. Organs other than the central nervous system do not possess this property of combining with tetanus poison, just as the central nervous system is, on the contrary, incapable of combining with diphtheria poison, which clinically does not show any pronounced affinity for the central nervous system.

Wassermann² also believes recently to have given

¹ Wassermann and Takaki, *Berliner Klin. Wochenschr.*, 1898.

² Wassermann, *New York Medical Journal*, 1904.

experimental proof of the second and third points, the increased production of the receptors and their thrusting off. For this purpose he employed a tetanus poison which he had kept for about eight years, and which was originally very poisonous. In the course of years, however, owing to the damaging action of light, of oxidation, etc., it had become so weak that it was no longer toxic at all. Injections of one cc. into a guinea pig produced no tetanus. Nevertheless the haptophore group remained intact, as could readily be proved, for this non-poisonous tetanus toxin was still able to bind tetanus antitoxin, i.e. thrust-off receptors. On injecting rabbits with this non-poisonous tetanus toxoid in increasing doses, and then examining the blood serum of the animal he found not a trace of tetanus antitoxin. This absence could have either of two causes: It might be that the toxoid no longer produced any physiological effect whatever in the organism; or although it still caused an increase in the receptors, these increased receptors remained in the organs (sessile) and were not thrust off into the blood. In order to decide this question Wassermann first determined the exact quantity of fresh tetanus toxin which constituted a fatal dose for guinea pigs. He reasoned that if he injected first the toxoid, and shortly after, say in one or two hours, the fresh toxin, he should in such an animal have to increase the fatal dose,

i.e. more tetanus toxin should be required to kill this animal than a normal one, because owing to the previous toxoid injection part of the cells susceptible to tetanus toxin would already have been occupied. Provided Ehrlich's theory were correct, so that this binding of the toxoid really occurred, the conditions should be entirely different when, instead of injecting the toxin shortly after the toxoid, he waited somewhat longer, one to three days, and then injected the fresh tetanus toxin. In that case Weigert's law should come into play and the receptors have commenced to increase in number, i.e. the organ should now possess more sensitive groups than before. This would manifest itself in such fashion that in contrast to the first experiment the fatal dose of fresh tetanus toxin could now be decreased; in other words a small dose would now tetanize the animal in a shorter time.

As a matter of fact Wassermann's experiments yielded exactly the results deduced theoretically. He injected a guinea pig with some of the non-poisonous toxoid and then, an hour later, with tetanus toxin. He found that much more toxin was required to kill this animal than a normal guinea pig of equal size. When, on the contrary, he waited one to three days, it was found that then a dose of tetanus toxin which would not even tetanize a normal guinea pig was sufficient to kill this one.

It will be seen that in the above experiments the completely non-poisonous toxoid, although it effected an increased production of receptors, did not cause their thrusting-off. The serum of the rabbit treated with toxoid contained no antitoxin whatever. Wassermann concludes from this and other experiments that the thrusting-off cannot be a function of the haptophore group, and that something additional is required. This "something," he claims is a function of the toxophore group. It may be stated that Von Dungern has also published experiments (with majaplastin) pointing to the existence of the second stage, the stage of sessile receptors.

Antigens or Haptins. — It has been found that it is impossible to produce any immunity against all poisons, e.g. strychnine or morphine. According to Ehrlich these simpler chemical molecules do not enter into a true chemical combination with the tissues, but form rather a kind of solid solution, a loose combination with the cells, so that they can again be abstracted from these cells by all kinds of solvents, e.g. by shaking out with ether or chloroform. The point can perhaps be likened to the difference between saccharin and sugar. Both substances taste sweet, but despite this similarity in their physiological action they behave very differently toward the cells of the organism. Saccharin simply passes through the organism without

entering into a firm combination, i.e. without being assimilated, and is therefore no food. Its sweetening action is a mere contact effect on the cells sensitive to taste. Sugar, on the contrary, is actually bound by the cells, assimilated and burnt, and so is a true food. Until recently it was believed that the simpler chemical substances could not excite the production of antibodies. Ford and Abel¹ have however been able to show that toad stool poison, a true toxin, against which an antitoxin can be produced is chemically a glucoside.

As we shall subsequently see it is possible to immunize the animal body against a large number of substances, including not only such cell products as ferments, toxins and venoms, but also cells of the greatest variety, bacteria, dissolved proteids, etc. All these substances, therefore, must possess haptophore groups able to combine with the side chains or receptors in the animal body. Collectively, we speak of such substances as *antigens* or *haptins*.

Nature of Antitoxins in General. — But little is known concerning the constitution of antitoxins, for we do not know them apart from serum or serum constituents. It seems probable that they are proteid in character, but this has not been positively decided. It has been found that like the globulins they are quite resistant to the action of trypsin, but are acted on by pepsin-hydrochloric

¹ Ford and Abel. Journal of Biological Chemistry, Vol. ii, 1907

acid. In general they withstand a fair degree of heat, certainly far more than the toxins. Antitoxins are to be regarded as inactive substances, effecting merely a blocking of the haptophore group of the corresponding toxin. They do not act on the toxins destructively. This is indicated by experiments of Wassermann on pyocyanus toxin, and of Calmette and Morgenroth¹ on snake venom, which showed that in the toxin-antitoxin combination, the toxin could again manifest itself after the antitoxin had been destroyed. The antitoxins therefore are not ferment-like substances. As far back as 1897 attempts were made to determine the chemical nature of the antitoxins. In that year Belfanti and Carbone² found that the antitoxin was precipitated with the globulins of the serum by means of magnesium sulphate. Dieudonné³ had previously shown that the proteids thrown out of solution by acetic and carbonic acids contained none of the antitoxin. In 1901 Atkinson⁴ showed that the globulins increase markedly in the serum of horses as the antitoxic strength increases. The most recent work on this subject is that of Gibson,⁵ who shows that if the ammonium sulphate precipi-

¹ Morgenroth, Berlin. klin. Wochenschr. 1905.

² Belfanti and Carbone, Centralblatt Bacteriologie (Ref.), Vol. xxiii. 1898.

³ Dieudonné, Arbeiten a.d. kaiserl. Gesundheitsamte. Vol. xiii, 1897.

⁴ Atkinson, Jour. Exper. Medicine, Vol. i, 1901.

⁵ Gibson, Journ. Biological Chemistry, Vol. I, 1906.

state (globulins, nucleo-proteids, etc.) is treated with saturated sodium chloride solution, practically all the antitoxic fraction passes into solution. Gibson's was the first really practicable method of concentrating the antitoxin. By means of it solutions of antitoxic globulin could easily be made to contain 1500 units per cc. Continuing Gibson's work, Banzhaf discovered that if the antitoxic serum or plasma was heated to 57° for 18 hours, there was a change of a considerable portion of the soluble globulins (soluble in NaCl solution) into insoluble globulins. The antitoxin remained unchanged. This procedure, therefore, permits of a still greater elimination of the non-antitoxic proteids.

Gibson has recently studied the possibility of differentiating other antibodies by means of their precipitation characteristics. He believes that a differentiation of the antibodies into those precipitated with the pseudo globulins and with the euglobulin fractions, according to the Hofmeister classification, is based on a misconception of the application of ammonium sulphate in separating proteids by their precipitation characters. While there seem to be some differences in the distribution of the antibodies in individual specific sera in comparative experiments, this is not so absolute as maintained by Pick¹ and others. Gibson's work on the fractionating of poly agglutina-

¹ Pick, *Beitrage z. chem. Physiol. u. Pathol.*, Vol. 1, 1901.

tive serum shows that no separation of the several antibodies developed in an individual serum is possible. In the case of antitoxic sera both Gibson and Ledingham find that in goat serum the antitoxin is not invariably associated with the euglobulin fraction as maintained by Pick, but shows the same solubilities as that in horse serum.

Toxins and other Poisonous Cell Derivatives, in General. — Soon after bacteriology had demonstrated the etiological connection between bacteria and disease, the conviction gained ground that it was less the actual destruction wrought by the bacteria directly, than the injury produced by their chemical products that gave rise to the lesions in the infectious diseases. Brieger, especially, was one of the first to direct attention to the probable existence of specific poisons in the bacteria. He isolated a number of well defined chemical substances called ptomaines, most of which were highly toxic. Subsequent study, however, showed that these were not the specific bacterial poisons. The latter, the true toxins are something quite different as we shall see in a moment. Still later other substances were isolated from bacteria, and these were termed toxalbumins. We now know that some of these were identical with the true toxins, but that others were entirely unrelated.

What then are the true toxins? A number of pathogenic bacteria, when grown in pure culture,

produce *dissolved* poisons in the culture fluid. These poisons are neither ptomaines nor proteid substances; their chemical nature is still absolutely unknown. They are extremely sensitive to external influences, especially against heat, and in many ways are very analogous to ferments. Physiologically the toxins are extremely poisonous, far beyond that of any of the ordinary well known poisons, and this poisonous action manifests itself only after a certain latent period known as the *period of incubation*. Finally one of the fundamental properties of the toxins is their ability to excite, in the organism attacked, antitoxins directed specifically against them, so that for every true toxin there is a corresponding antitoxin.

In addition to these bacterial toxins we know of other poisonous substances possessing similar characteristics. Among these are the "zoötoxins," — snake venoms, spider and toad poisons, the toxin of eel blood, and the "phytotoxins," — ricin, croton, abrin, etc. It may be mentioned that some of these are of somewhat more complex constitution than the ordinary bacterial toxins. Ricin, for example, appears to possess one haptophore group but two ergophore groups, a toxic and an agglutinating one. In the case of the snake venoms it is not yet definitely known whether they are haptins of the first order or of the second. (See page 49.)

The Relations Existing between Toxin and Antitoxin.—The exact nature of the toxin-antitoxin reaction has long been the subject of study and has given rise to considerable discussion. For obvious reasons most of the work has been done with diphtheria and tetanus toxins and their antitoxins. In order to give the reader some conception of the diverging views of various authorities we shall devote a few pages to a brief study of the diphtheria toxin-antitoxin reaction.

During the earlier years of toxin-antitoxin investigations the filtered or sterilized bouillon, in which the diphtheria bacillus had grown and produced its "toxin," was supposed to require for its neutralization an amount of antitoxin directly proportional to its toxicity as tested in guinea pigs. Thus, if from one bouillon culture ten fatal doses of "toxin" were required to neutralize a certain quantity of antitoxin, it was believed that ten fatal doses from every culture, without regard to the way in which it had been produced or preserved, would also neutralize the same amount of antitoxin. Upon this belief was founded the Behring-Ehrlich definition of an antitoxin unit.¹

The results of tests by different experimenters of the same antitoxic serum, but with different diph-

¹ This unit was "ten times the amount of antitoxic serum necessary to just protect a 250 gramme guinea pig against ten fatal doses of the toxin"

theria toxins, proved this opinion to be incorrect. Ehrlich¹ deserves the credit for first clearly perceiving and calling attention to this fact. He obtained from various sources twelve toxins and compared their neutralizing value upon antitoxin; these tests gave interesting and important information. The following table gives the results in four of his toxins and well illustrates the point in question:

Serial Number.	Estimated minimal fatal dose for 250 gm. guinea pigs.	Smallest number of fatal doses of toxic bouillon required to kill a 250 gm. guinea pig within 5 days when mixed with one antitoxin unit. ("L _† Ehrlich.")	Fatal doses required to "completely neutralize" one antitoxin unit as determined by the health of the guinea pig remaining unaffected. ("L ₀ Ehrlich.")	L _† minus L ₀ in fatal doses.	Remarks.
A	0.009 cc.	39.4	33.4	6	Old; deteriorated from 0.003 to 0.009.
B	0.0165 cc.	76.3	54.4	22	Fresh toxin, preserved with tricresol.
C	0.039 cc.	123.	108.	15	A number of fresh cultures, grown at 37° C. four and eight days.
D	0.0025 cc.	100	50	50	Tested immediately after its withdrawal.

It was natural to suppose, as the early investigators did, that a just neutral mixture of toxin and

¹ Ehrlich, Die Werthbemessung des Diphtherieheilserums. Klinisches Jahrbuch, 1897.

antitoxin, would require the addition of but one fatal dose of toxin in order to regularly kill the test animal. In the above table, however, we see that this difference ranges from six to fifty fatal doses.

Partial Saturation Method — Toxons, Toxoids. — Ehrlich obtained considerable additional information by means of his "partial saturation" method. Certain experiments had led him to believe that the original antitoxin on which he had based his "unit" determinations, while able to neutralize 100 fatal doses (per unit) really represented 200 "binding units," and that the toxic bouillon really contained several kinds of poisonous substances able to combine with antitoxin.

He now believes that the diphtheria bacilli excrete at least two such poisons, "toxins" and "toxons;" that these very quickly decompose to a greater or less extent forming various "toxoids."

In the case of a hypothetically pure toxin Ehrlich believes that one antitoxic unit would correspond to 200 fatal doses or 200 binding units. If the entire amount of antitoxin, i.e. $\frac{2}{100}$ is added to the amount of toxin in question, the result will be just complete neutralization. If the toxin is entirely pure, $\frac{1}{100}$ of the antitoxin unit would neutralize all but $\frac{1}{100}$ of the initial toxicity and $\frac{1}{200}$, or $\frac{1}{100}$ or $\frac{75}{100}$, etc. of the antitoxin added would permit corresponding degrees of toxicity to be demonstrated through animal inoculations. It was found, however, that neutralization according to this simple scale did not

take place. The results were complicated and Ehrlich found it convenient to express them graphically in the form of the so-called "toxin spectra." Without

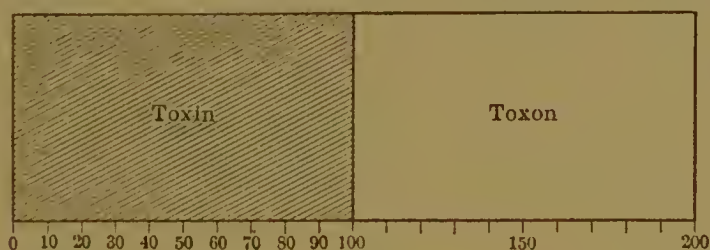


FIG. 2.

going much deeper into the subject the point may be illustrated by the appended diagrams or "spectra."

Fig. 2 shows the simplest conceivable diphtheria poison. In this case the following values would be obtained.

x^{cc} poison (100 fatal doses) + $\frac{2}{3}\%$ antitoxin units = 0, i.e. absolutely neutral.

x^{cc} poison + $\frac{1}{3}\%$ = Free toxon.

x^{cc} poison + $\frac{1}{6}\%$ = Free toxon.

That is to say, if the proportion of antitoxin added was $\frac{1}{6}\%$ of the amount required for complete neutralization, it would be found that the poison thus uncombined was much less, and *differently* toxic than a corresponding amount of the original toxin. It was found that these fractions possessed a rather constant though low degree of toxicity with characteristic action. This consisted in the production of some local œdema, followed by a long incubation period, and finally the development of cachexia and paralysis. Ehrlich believes

that this action is due to a separate poison excreted by the diphtheria bacillus which he calls a *toxon*.

If we continue with the above poison we shall obtain these values:

$$x^{cc} \text{ poison} + \frac{99}{200} = \text{Toxin action (1 fatal dose).}$$

$$x^{cc} \text{ poison} + \frac{70}{200} = 30 \text{ fatal doses.}$$

$$x^{cc} \text{ poison} + \frac{10}{200} = 90 \text{ fatal doses, etc.}$$

That is to say, if we add only $\frac{99}{200}$ units antitoxin, i.e. $\frac{1}{200}$ unit less than in the $\frac{100}{200}$ mixture, we find that one fatal dose is set free. This relation would exist right to the end. The fact that in this experiment the toxins are liberated after the toxons, shows that the toxons have less affinity for the antitoxin than have the toxins.

As a matter of fact, however, conditions are probably never as simple as this. In the process of toxin formation a double action is always going on — that of toxin and toxon production, and that of their decomposition. As was pointed out on a previous page the poisons quickly change into non-poisonous toxoids, and these substances are still able to bind antitoxin.

This is shown in the following "spectrum."

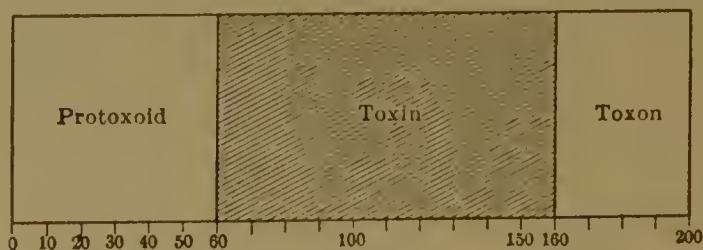


FIG. 3.

Here we would obtain the following figures:

x^{cc} poison + $\frac{3}{200}$ antitoxin unit = 0, i.e. absolutely neutral.

x^{cc} poison + $\frac{1}{200}$ = Toxon free.

x^{cc} poison + $\frac{1}{200}$ = Toxon free.

x^{cc} poison + $\frac{1}{200}$ = Toxin free (1 fatal dose.)

x^{cc} poison + $\frac{1}{200}$ = Toxin free (60 fatal doses.)

x^{cc} poison + $\frac{6}{200}$ = Toxin free (100 fatal doses.)

Now we come to the non-poisonous "prototoxoids":

x^{cc} + $\frac{5}{200}$ = Toxin free (100 fatal doses.)

x^{cc} + $\frac{3}{200}$ = Toxin free (100 fatal doses.)

x^{cc} + $\frac{1}{200}$ = Toxin free (100 fatal doses.)

We see here that after we have reduced the antitoxin to $\frac{6}{200}$ no further increase of toxicity is brought about by any further reductions. Ehrlich calls these toxoids "prototoxoids" because they have such a high affinity for the antitoxin. But there are apparently still other toxoids, as is shown by the following spectrum:

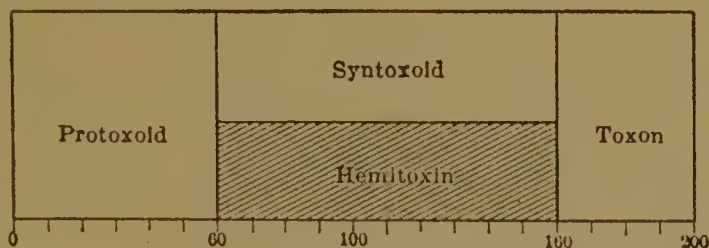


FIG. 4.

Here we would obtain values as follows:

x^{cc} poison + $\frac{3}{200}$ = 0, i.e. absolutely neutral.

x^{cc} poison + $\frac{1}{200}$ = Toxon.

x^{cc} poison + $\frac{1}{200}$ = Toxin free (1 fatal dose).

x^{cc} poison + $\frac{1}{100}$ = Toxin free (2 fatal doses.)

x^{cc} poison + $\frac{1}{50}$ = Toxin free (30 fatal doses.)

Here we find that in the middle part of the "spectrum" we encounter a zone in which each $\frac{1}{200}$ antitoxin unit neutralizes one fatal dose. Ehrlich believes that this part of the mixture consists of equal parts of syntoxoid and toxin—that is to say, he believes there are also toxoids which have the same degree of affinity for antitoxin that this toxin has. He speaks of these as "syn-toxoids."

By following out this conception of the toxin-antitoxin combination, Ehrlich comes to the conclusion that diphtheria poison is a very complex substance, containing toxin, toxon, and perhaps still other primary secretion products in addition to the various secondary modifications of these, toxoids, toxonoids, etc. It is difficult to escape the feeling that the existence of some of these hypothetical substances is more apparent than real.

Views of Arrhenius, Bordet and Others.—Bordet and others refuse to accept Ehrlich's views and the whole matter is still unsettled. Thus the existence or non-existence of toxons has excited a great deal of discussion among investigators.

The great Swedish chemist, Arrhenius, has given much attention to the toxins; and has applied the principles of physical chemistry to the toxin-

antitoxin reaction. It is, of course, well known that a solution of a compound such as sodium chloride represents not only NaCl in solution, but also sodium ions and chlorine ions. There is a certain amount of dissociation going on hand in hand with a combination of the two components. The degree of this varies with the temperature and the dilution of the substances. Arrhenius believes that the same process goes on with the toxin-antitoxin combination and that such more or less dissociated compounds give rise to the effects Ehrlich ascribes to the toxon. There is, however, no direct evidence that the combination of toxin-antitoxin is reversible. It is true that Morgenroth has been able to dissociate the two components of a neutral mixture of cobra venom and its antitoxin. But even here we are not dealing with a reversible reaction, for it requires certain manipulations to disrupt the neutral combination. In their work on the toxin of symptomatic anthrax, Grassberger and Schattenfroh found that different mixtures were obtained, depending on whether they mixed the toxin and antitoxin after diluting them, or diluted the toxin-antitoxin mixture. This fact is not in favor of Arrhenius' theory, for according to that, the same state of equilibrium should exist in both instances owing to reversibility, and the same fraction of the toxin of necessity remain free.

Bordet ¹ believes that the neutralization of toxin by antitoxin is an adsorption phenomenon, and compares it with the process of dyeing. The molecules of the toxin would "stain" more or less deeply by the antitoxin molecule, and the complexes that result in the various instances would be less toxic in proportion as they contained more antitoxin and less toxin. If a large piece of filter paper is placed in a certain volume of sufficiently diluted dye, it takes a uniform shade of intensity; if, on the other hand, the same sized piece of paper is cut in pieces and added in fragments, the first pieces are stained deeply, and the last find no color left. In the same way, on adding toxin to antitoxin in divided doses, the last portions of the poison cannot be neutralized, as the first are supersaturated with antitoxin. When the entire mixture is made at once, on the contrary, the antitoxin is spread all over the toxin molecules and a complex is obtained which contains an even proportion of the antidote, and which, consequently, is not as fatal as even a small dose of free toxin. The action which Ehrlich therefore ascribes to toxons, Bordet refers to toxin which is partially saturated with antitoxin. Bordet also cites the researches of Grassberger and Schattenfroh on the toxin of symptomatic anthrax. The toxic fluid which these

¹ Bordet-Gay, *Collected Studies in Immunity*, Wiley & Sons, 1909.

authors employed contains only a single poison: there is no reason for assuming the existence of toxoids, inasmuch as the toxic power of the poison is constantly parallel to its neutralizing power for antitoxin. On mixing a certain dose of the toxin either with little or with much antitoxin, complexes of toxin-antitoxin were obtained which varied in their reaction to heat. Moreover, these authors found that their poison absorbs much more antitoxin than is necessary to destroy its entire toxicity, and forms a stable compound with it. Bordet's conception of the toxin-antitoxin reaction thus seems to be very simple. The main difficulty which it encounters is the strict specificity of the combination. However, recent investigations make it probable that the affinity of adsorption is similar to a true chemical affinity, in that both are elective. It is possible, therefore, that the existence of strict specificity may still be found entirely compatible with the adsorption theory.

AGGLUTININS

The Agglutination Phenomenon. — We have just seen that pathogenic bacteria may be divided into those which produce extracellular toxins in culture media, and those which do not. Against the former the organism defends itself by the production of antitoxins; against the latter it produces a variety of antibodies: — bacteriolysins, agglutinins, precipitins, opsonins and possibly others.

The agglutinins can be observed either in a test-tube or in a microscopical preparation. For example, if typhoid or cholera immune sera are added respectively to a 24-hour culture of typhoid or cholera bacilli, and the mixture placed in a thermostat, the following phenomenon will be noticed: The bacteria which previously clouded the bouillon uniformly, clump together into little masses, settle to the sides of the test-tube and gradually fall to the bottom until the fluid is almost entirely clear. In a control test, on the contrary, to which no active serum is added, the fluid remains uniformly cloudy. The reaction is completed in twenty-four hours at the most. If the reaction is observed in a hanging drop, it is seen that the addition of the active serum first produces an increased motility of the

bacteria which lasts a short time and is followed by a gradual formation of clumps. One gets the impression that the bacteria are dying together. Frequently one sees bacteria which have recently joined a group make violent motions as though they were attempting to tear themselves away; then they gradually lose their motility completely. Even the larger groups of bacteria may exhibit movement as a whole. After not more than one or two hours the reaction is completed; in place of the bacteria moving quickly across the field, one sees one or several groups of absolutely immobile bacilli. Now and then in a number of preparations one sees a few separate bacteria still moving about among the groups. If the reaction is feeble, either because the immune serum has been strongly diluted or because it contains very little agglutinin, the groups are small and one finds comparatively many isolated and perhaps also moving bacteria. It is essential each time to make a control test of the same bacterial culture without the addition of serum. Under some circumstances the reaction proceeds with extraordinary rapidity so that the bacilli are clumped almost immediately. By the time the microscopical slide has been prepared and brought into view nothing is to be seen of any moving or isolated bacteria, and only by means of the control test is it possible to tell whether the culture possessed normal motility.

We are not yet informed as to the nature of these phenomena. A number of theories have been advanced, into which, however, we cannot here enter.

In some cases the agglutinins are active even in very high dilutions. Thus in typhoid patients and typhoid convalescents a distinct agglutination has been observed in dilutions of 1:5000, and this action persisted for years, though not, of course, in the same degree. Even normal blood-serum, when undiluted, often produces agglutination. But the above specific agglutinins, which do not exist beforehand, being formed only in consequence of an infection, are characterized by this, that the agglutination occurs even when the serum is diluted (at least 1:30 to 1:50), and, furthermore, that after this dilution the action is still specific, i.e. cholera immune serum agglutinates only cholera bacilli, typhoid immune serum only typhoid bacilli, etc. This specificity, however, as will be shown later, is not always absolute.

Agglutinins can also be developed against red blood cells and against certain protozoa (trypanosomes). We speak of the former as *hæmagglutinins*. Analogous to the hæmolytic action or *normal* serum on the red cells of certain other species, we find that *normal* serum is able to *agglutinate* the red cells of many species and bacteria. For example, normal goat serum agglutinates the red cells of man, pigeon, and rabbit;

normal rabbit serum agglutinates typhoid and cholera bacilli.

Of practical interest is the fact that normal serum may agglutinate the red blood cells of another individual of the same species. Following Ehrlich's nomenclature, we speak of this as *isoagglutination*. The subject has been studied by a number of investigators, and mostly in human blood. According to the extensive investigations of Moss, isoagglutinins occur in the serum of about 90% of adult human beings. Landsteiner divided the individuals into three groups, namely:

Group 1.—The *corpuscles* are not agglutinated by sera of the other two groups, while the *sera* agglutinate the corpuscles of both groups.

Group 2.—The *corpuscles* are agglutinated by the sera of the other two groups, while the *sera* agglutinate the corpuscles of Group 3, but not of Group 1.

Group 3.—The *corpuscles* are agglutinated by the other two sera, and the *sera* agglutinate the corpuscles of Group 2, but not of Group 1.

An examination of this grouping shows that in no case is there an agglutination of erythrocytes by their own serum, in other words these are isoagglutinins but not autoagglutinins. A somewhat different classification was made by Jansky, and independently of him also by Moss. Both these authors find it necessary to establish four groups in order to embrace all the cases met with.

Gay calls attention to the fact that the clumping of erythrocytes by serum is not necessarily due to the presence of an agglutinin at all, but may be due to variations in the molecular concentration of the serum constituents or of the constituents of the blood cells. Whatever finally proves to be the correct explanation, it is obvious that the occurrence of isoagglutination in human blood might

possibly prove disastrous in human homologous transfusion. For this reason it is now common practice to always precede homologous transfusion by an examination of the blood of both donor and recipient. A brief outline of these tests is given in the appendix.

Purpose of Agglutination.—It is not yet clear what the purpose, if any, of the agglutinating function is. Gruber, the first to thoroughly study and appreciate the bacterial agglutinins, assumes that the process injures the affected cell, preparing it for solution and destruction. After numerous experiments I have not been able to convince myself of any damaging influence of the agglutinins on the affected cell, be this blood cell or bacterium, and the observations of other authors confirm this opinion. Agglutinated bacteria are capable of living and of reproduction, and agglutinated red blood cells are no more fragile or easier to destroy than normal, non-agglutinated cells. Neither can anything be discovered microscopically which would indicate any injury to their structure.

One thing is certain: that the agglutinins are in no way related to the lysins found in serum, and so of course are not identical with these. The simultaneous occurrence in a serum of immune bodies, interbodies, complements, and agglutinins is an entirely independent phenomenon which is in no way regular. There are sera which dissolve certain cells without agglutinating them, and others which agglutinate cells without dissolving them.

Historical. — Serum diagnosis by means of the agglutinins was introduced chiefly through the labors of Gruber and Widal. The studies undertaken by Gruber and his pupil Durham began as early as 1894. At the Congress for Internal Medicine in 1896¹ Gruber first announced that he had discovered the reaction in typhoid convalescents, and asked that his observations be verified if possible. Soon after this Pfeiffer and his co-workers published a study which confirmed Gruber's results.² The significance of the reaction as a diagnostic help was unquestionably first pointed out by Widal,³ who showed that the reaction appears at a relatively early period of the disease, and may therefore be employed as a diagnostic measure. We must not omit to state that Grünbaum⁴ in March, 1896, several months before Widal's publication, had also grasped the significance of the reaction as a diagnostic measure. Owing to insufficient clinical material his publication did not appear until some time after Widal's. Hence, in acknowledgment of the labors of the two authors most concerned in the discovery and introduction of this reaction, we now speak

¹ Transactions of the Congress, edited by E. von Leyden and R. Pfeiffer, Wiesbaden, 1896.

² Pfeiffer and Kolle, *Deutsche med. Wochenschrift*, 1896, No. 12.

³ Widal, *Bulletin de la soc. méd. des hôp.*, June 26, 1896.

⁴ Grünbaum, *Lancet*, Sept. 19, 1896; *Muench. med. Wochenschrift*, 1897, No. 13; Blood and the identification of bacterial species, *Science Progress*, Vol. I, No. 5, 1897.

of it as the "Gruber-Widal reaction," whereas in the beginning only the term "Widal reaction" was used.

The manner in which the reaction proceeds in microscopical preparations as well as when macroscopically observed has been described above (page 32). Nowadays the microscopic method is given the preference¹ because in many cases it is distinct when the macroscopic reaction fails; and further because the former yields distinct results within an hour at the most, whereas in many cases twenty-four hours are required for the macroscopic test.

Pfaundler's Reaction (Thread Reaction). — It may be well at this point to call attention to a peculiar reaction described by Pfaundler² in 1896. This author showed that certain bacteria, though they might not be agglutinated by a given serum, would often, when they were grown therein, develop in the form of long threads more or less interlaced. This occurred only in the specific serum and was absent in the normal serum. Most authorities regard the thread reaction as a manifestation of agglutinins. According to Metchnikoff this reaction sometimes gives more information concerning a serum than does the ordinary agglutination test.

¹ This applies to typhoid; in other diseases the macroscopic method is sometimes preferable.

² Pfaundler, *Centralblatt Bacteriologie*, Vol. xix, 1896.

Nature of the Agglutinins, and of the Agglutination Reaction.—The agglutinins are fairly resistant substances which withstand heating to 60°C ., and lose their power only on heating to 65°C . It is possible, therefore, to make a serum bacteriolytically inactive by heating to 55°C ., and still preserve its agglutinating power. It has been found that agglutinins when heated may keep the property of uniting with bacteria, although they lose the property of agglutinating them. To explain this fact, Ehrlich supposes that agglutinins possess two groups, a haptophore group, effecting the specific union with the cell, and an ergophore group, which effects the clumping. He supposes further that under the conditions mentioned the agglutinin loses its agglutinating group but keeps its combining group. Such a modified agglutinin Ehrlich calls an *agglutinoid*, just as toxins which have lost their toxophore groups are called toxoids. The nature of agglutinoid, however, is still very obscure. In fact, as we shall presently see, the opponents of the Ehrlich school refuse to believe in the existence of agglutinoids. It has occasionally been observed that agglutination is absent in concentrated serum, and present in dilute serum. This zone of no agglutination, preceding that of agglutination, is often spoken of as the *pro zone* and was first described by Eisenberg and Volk. According to Ehrlich it is due to the presence in the serum of agglutinoids. These are assumed to possess higher affinity for the bacteria than do the agglutinins

and so prevent the latter from acting on the bacteria. Since, however, the agglutinins are usually far more abundant than the agglutinoids, dilution of the serum dilutes the latter to practically nothing, thus allowing the agglutinins to combine with the bacteria.

Ehrlich's conception of the structure of the agglutinin molecule and his views on the nature of the agglutination reaction have been sharply combated.

Elser very properly points out that not enough attention has been paid to the effect of heat on serum, and that alterations in the physical characters of the serum may be sufficient to account for phenomena heretofore ascribed to chemical changes. Among other things he cites the effect of heat on horse serum; heating produces a marked increase in the viscosity of the serum. It is obvious, therefore, when heated sera are used in agglutination experiments, that this purely physical characteristic exerts a profound influence on the result of the reaction. Differences in the behavior of an agglutinating serum before and after heating must therefore be interpreted with great caution, and must not at once be taken to indicate the chemical alteration of the agglutinin complex.

Bordet, for example, cites an interesting experiment of Gengou. An aqueous solution of agar, so diluted as to be only slightly viscous at room temperature, agglutinates barium sulphate suspended in water. Heating such a solution destroys this property without affecting the adsorbing property; under these conditions it produces the

opposite effect, namely, disseminates the particles of barium and gives a milky appearance to the fluid. Can we, says Bordet, claim that by heating this solution we have caused it to lose its agglutinating group? Bordet agrees with Porges, who believes that the hypothesis of such a group in the antibody molecule has no foundation. Porges found, on studying the effect of heat on the agglutinating power of the albuminous substances of serum for mastic emulsions, that he could obtain results entirely similar to those that have been noted for agglutinins. Bordet insists that we have no right to localize the cause of agglutination in a molecule of the antibody rather than in one of the antigen. The hypothesis of a functional group in the molecule of the agglutinin, he says, is all the more doubtful, inasmuch as it is not the only substance which can render bacteria sensitive to the flocculating action of salts. Bacteria that have adsorbed iron, uranium, or aluminium compounds are subsequently flocculable by salts, and silicic acid is similar in its action. According to Bordet, the essential phenomenon with agglutination, as with other active substances in sera, is its union with the antigen; as far as the agglutination itself, which follows this union, is concerned, it is only a secondary phenomenon on which we cannot depend in considering agglutinins as functionally different in molecular structure from the other antibodies.

The influence of salts upon agglutination is in

a sense comparable to their action upon the precipitins. Joos found that antityphoid serum did not agglutinate typhoid bacilli in the absence of salts. For agglutination to take place he considers it as necessary as the agglutinin and agglutinable substance. He believes that salts play an *active* part in the process. Bordet, on the other hand, believes that the absence of salts offers only a physical impediment to agglutination. Friedberger does not consider that the salts act chemically, for he found that agglutination took place in the presence of grape sugar, asparagin, etc., in the place of salts.

In view of the fact that the protoplasm of the body and the albuminous constituents of serum have a close relationship to, or really are, colloids, investigators have studied certain reactions which occur among the colloids with the expectation that these would throw some light on the reactions of protoplasm and of serums.¹ Colloids diffuse very slowly and exert little or no osmotic pressure, supposedly because of the large size of the particles. They do not conduct electricity, but the particles react to the electric current by alterations in the direction of their motion (i.e., toward the positive or the negative pole) and, moreover, carry electric charges themselves. The features of colloids which bring them into relation with the subject in hand are their coagulable nature in certain instances and the fact that their particles may be agglutinated

¹ This subject is well presented in: Pauli-Fischer, *Physical Chemistry in the Service of Medicine*. Wiley & Sons, N. Y.

or precipitated by the addition of minute amounts of salts (electrolytes). This of course is entirely analogous to the need of salts in the agglutination of bacteria by sera. In the latter reaction the agglutinins carry a positive, the bacteria a negative charge. The resulting combination, therefore, does not precipitate from the menstruum supposedly because there is still sufficient difference in the electric potential. When salts are present the kations so alter the electric conditions of the colloidal particles, i.e., of the agglutinin-bacterium combination, that their surface tension is increased. In order to overcome this the particles get together, presenting in a clump less surface tension than if they remained as individual particles. Some experiments by Field indicate that the *pro zone* may be explained on the assumption that the bacteria and agglutinins behave as colloids. It has already been stated that the union of agglutinin and bacterium does not precipitate because, possibly, there is still sufficient electric potential; the combination carries a negative charge. Field believes that with very large amounts of agglutinin (as in the *pro zone*) the bacteria load themselves with so much agglutinin that the combination now carries a considerable positive charge. The surface tension therefore is not sufficient to cause a clumping to occur. Naturally, the presence of salts does not alter the condition, as the kations also carry a positive charge.

Group Agglutinins.—For some time after their

discovery the agglutinins were regarded as strictly specific, i.e., a serum derived, for example, from a typhoid infection would agglutinate only typhoid bacilli and no others. After a time, however, it was found that such a serum would frequently agglutinate somewhat related organisms, though not, usually, to so high a degree. In other words, while *agglutinins* may be nearly, if not quite, specific in their action, a serum which produces agglutination may be far from being so.

The following examples will illustrate the point. In a case of infection with paratyphoid bacilli, type B, the bacilli of the infecting type B were agglutinated 1:5700; typhoid bacilli, however, only 1:120, while paratyphoid bacilli type A were not agglutinated at all. In a case of typhoid infection an agglutination with a dilution of 1:40 was obtained for paratyphoid type B, while typhoid bacilli were agglutinated in a dilution of 1:300 and over. As a rule the agglutination with the infecting agent is by far the strongest, i.e. it proceeds even in high dilutions, whereas other bacteria require a stronger concentration.

This phenomenon is known as *group agglutination*. The bacteria which are agglutinated by one and the same serum need not necessarily be related, although usually this is the case. Conversely, microorganisms which, because of their morphological or other biological characteristics, are regarded as entirely identical or nearly so, are sharply

differentiated by means of their agglutination. Because of this lack of absolute specificity the serum diagnosis of infection or the identification of bacteria by means of agglutination tests, has value only when very carefully tested. We have said above that while agglutinins are specific, a serum which produces agglutination may be far from being so. The reason for this is that the serum may contain several agglutinins. In fact, when immunizing an animal with a particular bacterium both specific and group agglutinins are produced. This will perhaps be made clearer by reference to the following diagram. We assume that the typhoid

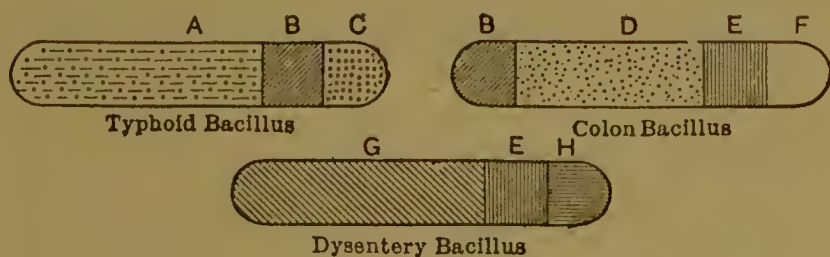


FIG. 5.

bacillus possesses considerable protoplasm A, which is specific for the typhoid bacillus; that it possesses also certain protoplasm B, which is common to it, and to the colon bacillus; and some protoplasm C, common perhaps to some other bacterium. In the case of the colon bacillus, protoplasm D is specific, i.e., possessed only by this bacillus, while B is common to it and the typhoid bacillus, and E common to colon and dysentery bacilli. By immunization with the typhoid bacillus we would obtain

a serum containing agglutinins against protoplasm *A*, *B*, and *C*. By virtue of this the serum would exert some agglutinating power also on colon bacilli.

Absorption Method for Differentiating between a Mixed and a Single Infection and for Identifying Bacteria.—

In 1902 Castellani called attention to a procedure which consists in saturating the diluted immune serum with successive quantities of the bacteria most strongly agglutinated until the agglutinating power for these is zero. After centrifuging, the clear fluid is tested on the second variety of bacteria, and from this one learns whether mixed or single infection was present. According to Castellani, if the serum of an animal immunized against a certain microorganism is saturated with that organism, the serum will lose its agglutinating power not only for that organism, but also for all other varieties that it formerly acted on. Saturated with the others, its action upon the first is reduced little or none at all. The serum of an animal immunized against two microorganisms *A* and *B*, loses its agglutination when saturated with *A*, only for *A*. Saturated with *A* and *B* it loses agglutinating power for both.

The absorption test is extensively used in the identification of bacteria, but it must be used with caution, as its interpretation is open to error. Referring to the figure illustrating specific and group agglutinins, let us assume we have obtained a specific typhoid serum by immunization with typhoid bacilli. By virtue of the common agglutinin, this

serum will act also on colon bacilli. On extracting such a serum with typhoid bacilli, all the agglutinating power would be lost, that for typhoid bacilli as well as that for colon. On extracting the serum with colon bacilli, we would remove the agglutinating power for these bacilli, but leave the specific agglutinating power for typhoid bacilli. If we extracted the serum with a culture suspected to be typhoid bacilli, and found after extraction that the serum no longer agglutinated known typhoid bacilli, we could conclude that the suspected culture was also one of typhoid bacilli.

Formation of the Agglutinins According to the Side-Chain Theory—Receptors of First, Second and Third Order.—Ehrlich's theory as outlined in the preceding chapter offers a ready explanation for the development of these bodies. Certain peculiarities of the agglutinins require merely a slight elaboration of detail in order to be clearly understood. According to Ehrlich the prime function of the side chains of a cell is to provide for the nutrition of the cell. Obviously the simplest mechanism for this purpose will be a side chain which merely anchors the food molecule, leaving the digestion entirely to the cell proper. This type of receptor suffices for comparatively small molecules such as those of the toxins, for these are, after all, but the products of cellular activity. When the protoplasm of the bacterial cell itself, however, is to serve as food for the animal cell the latter needs more than a mere

anchoring group, it needs also an active group which can in some way act on the huge food particle and make it more readily assimilable. Such receptors then possess two groups, a haptophore group and another functional group acting on the food particle thus anchored. Ehrlich calls these his "receptors of the second order," and places in this class the agglutinins and the precipitins. The same action can perhaps be more economically brought about by having these receptors, in addition to their specific haptophore group, possess the means by which the action of a ferment-like substance can be brought to bear on the anchored food particle. Such a receptor would then possess two haptophore groups, one for the food particle, the other for the ferment-like substance. These are Ehrlich's "receptors of the third order" and will be discussed in the next chapter. Confining ourselves for the present to the agglutinins we find that the existence of the two groups (haptophore and agglutinating) has experimental confirmation. We have seen that an agglutinin may be changed by the action, for instance, of acids, so that it will no longer possess any agglutinating action, but will still combine with the bacteria. Once the agglutinating power is lost it cannot be restored, in which respect the agglutinins differ from the bacteriolysins.

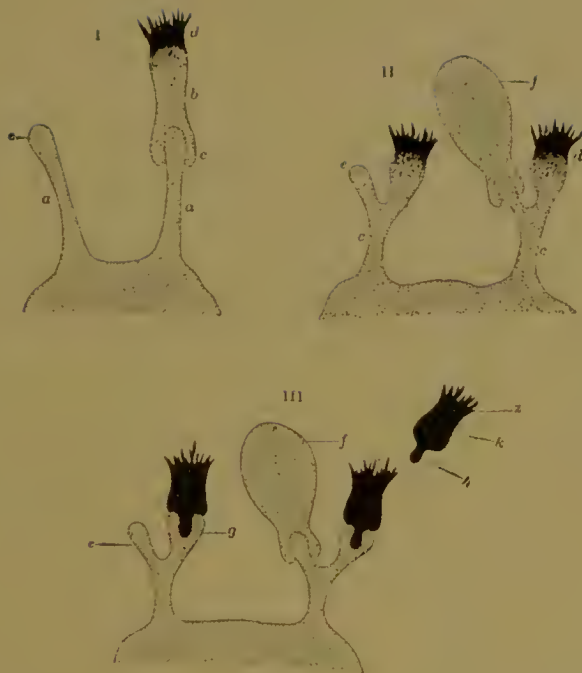


FIG. 6. — THE VARIOUS TYPES OF RECEPTORS ACCORDING TO EHRLICH.

- I. *Receptors of the First Order.* — This type is pictured in *a*. The portion *e* represents the haptophore group, whilst *b* represents a toxin molecule, which possesses a haptophore group *c* and a toxophore group *d*. This represents the union of toxin and antitoxin, or ferment and antiferment, the union between antibody and the toxin or ferment being direct.
- II. *Receptors of the Second Order* are pictured in *c*. Here *e* represents the haptophore group, and *d* the zymophore group of the receptor, *f* being the food molecule with which this receptor combines. Such receptors are possessed by agglutinins and precipitins. It is to be noted that the zymophore group is an integral part of the receptor.
- III. *Receptors of the Third Order* are pictured in III, *e* being the haptophore group and *g* the complementophile group of the receptor. The complement *k* possesses a haptophore group *h* and zymotoxic group *z*; whilst *f* represents the food molecule which has become linked to the receptor. Such receptors are found in hæmolysins, bacteriolysins, and other cytolytins, the union with these cellular elements being effected by the amboceptor (a thrust-off receptor of this order). It is to be noted that the digesting body, the complement, is distinct from the receptor, a point in which these receptors therefore differ from those of the preceding order.

BACTERIOLYSINS AND HÆMOLYSINS

Historical. — As far back as 1874, Gscheidlen and Traube¹ demonstrated that considerable quantities of septic material could be injected into the circulation of warm-blooded animals without apparently any effect on the animal. Very little was thought of this observation at the time, and it is not until more than ten years later that we find a similar observation made by Fodor.² In 1888 Nuttall³ showed that normal blood serum possessed marked germicidal properties, and his observations stimulated a number of workers who undertook to determine the conditions most favorable to the exhibition of this phenomenon, and further to decide upon the constituent of the serum to which this property was due or whether it was a function of the serum as a whole. In 1889 Buchner⁴ published a series of experiments and showed that an exposure of 55° C. robs the serum of its bactericidal property. He concluded that the active element in the process was a living albumen and

¹ Gscheidlen and Traube. *Schlez. Gesellschaft. f. Vaterland. Cultur, Med. Sect.*, 1874.

² Fodor, *Deutsche med. Wochenschr.*, 1886.

³ Nuttall, *Zeitschr. f. Hygiene*, Vol. iv, 1888.

⁴ Buchner, *Centralblatt Bacteriologie*, Vol. v, 1889. *Archiv. f. Hygiene*, Vol. x, 1890.

suggested for it the name "alexin." He found that it was possible to greatly increase the bactericidal action, (i.e. the quantity of "alexin") for a particular bacterium by immunizing an animal with that bacterium.

Pfeiffer's Phenomenon. — An enormous advance in the study of immunity was made in the discovery of Pfeiffer's phenomenon in 1894, and it is to Pfeiffer's splendid observations¹ that we owe the first and most important insight into the mode of action of the bacteriolytic immune sera. A normal guinea pig is able to kill and dissolve a number of living cholera spirilla if these are injected intraperitoneally. If in such an animal we gradually increase the dose injected, it will be possible after a time to inject at one dose an amount of cholera spirilla that represents many times an ordinary fatal dose. If from this animal we now withdraw serum and inject it into another animal, we find that this serum, even in such small amounts as the fractional part of a centigram or even of a milligram, is able to protect the second animal against living cholera spirilla. Under the influence of these small amounts of serum of the treated animal, the organism of the untreated animal is able to dissolve large amounts of cholera spirilla, amounts which would otherwise be invariably fatal. This process, as R. Pfeiffer showed, is a specific one, i.e.,

¹ R. Pfeiffer, *Zeitschr. Hygiene*, Vol. xviii, 1894.

the serum of the guinea pig treated with cholera spirilla transmits an increased solvent power only for cholera spirilla, but not for any other species of bacteria. The active substance of such a bacteriolytic immune serum Pfeiffer called a *specific bactericide*. If we allow some of this specific cholera immune serum to remain for some time outside of the body, e.g. in a bottle, and then test it for solvent properties against cholera spirilla, not in a living body but in a test-tube, we shall find that its power is almost nil. If we add to this serum in the test-tube some fresh peritoneal exudate or some other body fluid, such as serum of a normal, untreated guinea pig, as Metchnikoff first did, we find that this serum has now acquired the power to rapidly dissolve cholera spirilla even in a test-tube. Bordet,¹ in 1895, showed that in order for the specific immune serum to dissolve spirilla in a test-tube, it is unnecessary to add fresh normal serum or peritoneal fluid; but that immune serum freshly drawn from the vein is able even under these circumstances to dissolve the spirilla.

Hæmolysis.—In his experiments with the bacteriolysis of cholera spirilla, Bordet used as an immune serum the serum of a goat that had been immunized against cholera spirilla, and as alexin, fresh normal guinea-pig serum. It often happened that the latter contained a certain number of red

¹ Bordet, *Annal. Inst. Pasteur*, 1895.

blood cells and he found that these were agglutinated and would be agglutinated even when mixed with normal goat serum. Knowing, as he did, that immunization against bacteria increases the agglutinating property toward a given organism over that in the normal animal, it was natural that he should experiment to see whether similar results could be obtained with red blood cells. Accordingly he injected guinea pigs several times with 5 cc. defibrinated rabbit's blood, and found that not only did this guinea-pig serum acquire agglutinating properties, but also the property to dissolve rapidly and intensely, in a test-tube, the red blood cells of a rabbit. The serum of a normal guinea pig was incapable of doing this, or did it in only a slight degree. Bordet could further show that this action is a specific one, i.e., the serum of animals treated with rabbit blood acquires this dissolving property only for the red cells of rabbits, not for those of any other species of animal. For the latter, such a serum is no more strongly solvent than the serum of a normal animal. The same property that Bordet had demonstrated in the serum of guinea pigs treated with rabbit blood could now be shown for the sera of all animals treated with blood cells of a different species. We can formulate this as follows: The serum of animals, species *A*, after these have been injected either subcutaneously, intraperitoneally, or intravenously with erythrocytes

of species *B*, acquires an increased solvent action for erythrocytes of species *B*, and only for this species.¹ It is therefore a specific action. We call this *hæmolysis*, and the substances which effect the solution of the red cells, *hæmolysins* or *hæmotoxins*.

At about the same time, and independently of Bordet, similar experiments with similar results were published by Landsteiner² and v. Dungern.³ As a result of this work, the acquired toxicity of horse serum, found by Belfanti and Carbone when they treated horses with red cells of rabbits, was explained. The serum of the horses so treated had become *hæmolytic* for rabbit blood, and therefore caused a solution or destruction of the red cells in the living body just as it did in a test-tube.

Nature of Hæmolytic Sera.—In a subsequent study Bordet⁴ was able to show that the solvent power of the specific hæmolysins depended on the combined action of two constituents of the specific serum. When the fresh hæmolytic serum was heated for half an hour to 55° C., it lost its power. If to this *inactive serum* a very small amount of the serum of a normal guinea pig was added (a serum which of course was not hæmolytic for rabbit red cells), the full hæmolytic power was

¹ We shall point out a few exceptions later on.

² Landsteiner, *Centralblatt Bacteriol.*, Vol. xxv, 1899.

³ Von Dungern, *Munch. med. Wochenschrift*, 1898.

⁴ Bordet, *Annal. Inst. Pasteur*, Vol. xii, 1898.

restored to this inactive serum. In other words, it had been *reactivated* by this addition.

This experiment permits of only one conclusion, namely, that the hæmolytic action of the specific hæmolytic serum depends on two substances. One of these is able to withstand heating to $55^{\circ}\text{C}.$, and is contained only in the specific serum. The other is destroyed by heating to $55^{\circ}\text{C}.$, and is contained not only in the specific hæmolytic serum, but also in the serum of normal untreated animals.

Buchner, we have seen, applied the term *alexins* to the constituents of normal serum which were actively destructive to corpuscular elements, bacteria, and other cells with which they came in contact. This term was retained by Bordet to designate that constituent of normal serum which did not withstand heating to $55^{\circ}\text{C}.$, and which was one of the factors in the hæmolytic process. The other substance, which was found only in the specific serum and which withstood heating to $55^{\circ}\text{C}.$, he termed *substance sensibilatrice*.

According to Bordet, therefore, the substances required for hæmolysis are the substance sensibilatrice of the specific hæmolytic serum and the alexin which exists even in normal serum. The action of these two substances Bordet explains by assuming that the red cell is not vulnerable to the alexin; just as, for example, there are certain substances that will not take a dye without the previous

action of a mordant. The substance sensibilatrice plays the rôle of mordant. It makes the blood cells vulnerable to the alexin, so that the latter can attack the cells and dissolve them. The alexin he regards as a sort of ferment body with digestive powers.

Bordet says further, that the substance sensibilatrice sensitizes the blood cells not only for the alexin derived from the serum of the same species as that from which it (the substance sensibilatrice) is derived, but sensitizes such cells also for the alexins of normal sera of other species. For example, in the foregoing experiment of Bordet, the substance sensibilatrice derived from the guinea pig by treatment with rabbit blood sensitizes the red blood cells of rabbits not only for the alexin of normal guinea pig blood, but also for the alexins of other normal sera. In another experiment this author showed that rabbit red cells sensitized with an inactive specific hæmolytic serum derived from a guinea pig would dissolve rapidly on the addition of normal rabbit blood. Here, then, the rabbit red cells, sensitized (according to Bordet) by the substance sensibilatrice of the guinea pig, dissolve on the addition of the alexin of their own serum.

Résumé.—Reviewing the important facts we have learned, we find them to be as follows: By means of the treatment of one species of animal with the erythrocytes of a different one, the serum of the

first species acquires an uncommonly increased power to dissolve and to agglutinate the erythrocytes of the second species. This increased hæmolytic power shows itself not only *in vivo*, so that an animal so treated is able to cause red cells injected into it to rapidly dissolve and disappear, but it shows itself also *in vitro* when the serum of this animal is used. The process consists in the combined action of two substances, that which is excited in response to the injection, the substance sensibilatrice, and the alexin of normal serum.

Analogy between the Bacteriolytic and Hæmolytic Processes.—If we now recall the main points in cholera immunity the close analogy between this and the subject of hæmolysis is apparent. Just as, when immunizing an organism against cholera bacilli the organism responds with an increased solvent power for those bacteria, so does the organism respond when it is treated, *i.e.* immunized, with red cells of another species, by increasing the solvent power of its serum for those particular cells. Furthermore, just as the hæmolytic process was seen to depend on the combined action of two substances, one developed in the hæmolytic serum, the other already present in normal serum, so also in the bactericidal process just studied there are two factors. It is easy to understand, therefore, what formerly was not at all clear, why a specific bactericidal serum against cholera, typhoid, or

other infectious disease should not act in a test-tube unless there had first been added some normal serum (according to Metchnikoff), or there had been employed a perfectly fresh serum (according to Bordet): simply because in either of these ways the alexin necessary to co-operate with the substance sensibilatrice is introduced. This alexin no longer exists in the immune serum, if this be not perfectly fresh, for we have seen that it decomposes either on warming, or spontaneously on standing. A bactericidal serum, therefore, that has stood for some time is incapable of dissolving bacteria. It is possible, however, to make an old inactive serum again capable of dissolving bacteria *in vitro* by adding a little fresh alexin, according to the suggestion of Metchnikoff. In other words, it is thus reactivated. Another obscure point was cleared up by these studies: why a specific bactericidal serum which is inactive *in vitro* should be intensely active in the living body. This is because in the living body the serum finds the alexin necessary for its working, which is not the case in the test-tube unless fresh normal serum be added. We see from all this that even the first experiments in hæmolysis have served to clear up a number of practical points in an important branch of bacteriology.

Ehrlich and Morgenroth on the Nature of Hæmolysis. — In continuing the study of hæmolysins we

must note particularly the researches of Ehrlich and Morgenroth.¹ These authors asked themselves the following questions: (1) What relation does the hæmolytic serum or its two active components bear to the cell to be dissolved? (2) On what does the specificity of this hæmolytic process depend? Ehrlich was led to these researches particularly by his so-called Side-chain Theory, which we shall examine in a moment.

He made his experiments with a hæmolytic serum that had been derived from a goat treated with the red cells of a sheep. This serum, therefore, was hæmolytic specifically for sheep blood cells; i.e., it had increased solvent properties exclusively for sheep blood cells.

Basing his reasoning on his side-chain theory, Ehrlich argued as follows: "If the hæmolysin is able to exert a specific solvent action on sheep blood cells, then either of its two factors, the substance sensibilatrice of Bordet or the alexin of normal serum, must possess a specific affinity for these red cells. It must be possible to show this experimentally." Such in fact is the case, and the experiments devised by him are as follows:

Experiment 1. — Ehrlich and Morgenroth, as already said, experimented with a serum that was specifically hæmolytic for sheep blood cells. They

¹ See the various papers in "Collected Studies on Immunity," Ehrlich-Bolduan, Wiley & Sons, New York, 1910.

made this inactive by heating to 55°C ., so that then it contained only the substance sensibilatrice. Next they added a sufficient quantity of sheep red cells, and after a time centrifuged the mixture. They were now able to show that the red cells had combined with all the substance sensibilatrice, and that the supernatant clear liquid was free from the same. In order to prove that such was the case they proceeded thus: To some of the clear centrifuged fluid they added more sheep red cells; and, in order to reactivate the serum, a sufficient amount of alexin in the form of normal serum was also added. The red cells, however, did not dissolve — there was no substance sensibilatrice. The next point to prove was that this substance had actually combined with the red cells. The red cells which had been separated by the centrifuge were mixed with a little normal salt solution after freeing them as much as possible from fluid. Then a little alexin in the form of normal serum was added. After remaining thus for two hours at 37°C . these cells had all dissolved.

In this experiment, therefore, the red cells had combined with all the substance sensibilatrice, entirely freeing the serum of the same. That the action was a chemical one and not a mere absorption was shown by the fact that red blood cells of other animals, rabbits or goats for example, exerted no combining power at all when used instead of

the sheep cells in the above experiment. The union of these cells, moreover, is such a firm one that repeated washing of the cells with normal salt solution does not break it up.

So far as concerns the technique of the experiments, I should like to observe that the addition of red cells in this as well as in all the following experiments was always in the form of a 5% mixture or suspension in 0.85%, i.e., isotonic, salt solution.

The second important question solved by these authors was this: What relation does the alexin bear to the red cells? They studied this by means of a series of experiments similar to the preceding.

Experiment 2. — Sheep blood was mixed with normal, i.e. *not* hæmolytic, goat serum. After a time the mixture was centrifuged and the two portions tested with substance sensibilatrice to determine the presence of alexin. It was found that in this case the red cells acted quite differently. In direct contrast to their behavior toward the substance sensibilatrice in the first experiment, they now did not combine with even the smallest portion of alexin, and remained absolutely unchanged.

Experiment 3. — The third series of experiments was undertaken to show what relations existed between the blood cells on the one hand, and the substance sensibilatrice and the alexin on the other, when both were present at the same time,

and not, as in the other experiments, when they were present separately. This investigation was complicated by the fact that the specific immune serum very rapidly dissolves the red cells for which it is specific, and that any prolonged contact between the cells and the serum, in order to effect binding of the substance sensibilatrice, is out of the question. Ehrlich and Morgenroth found that at 0°C . no solution of the red cells by the hæmolytic serum takes place. They therefore mixed some of their specific hæmolytic serum with sheep blood cells, and kept this mixture at 0° – 3°C . for several hours. No solution took place. They now centrifuged and tested both the sedimented red cells and the clear supernatant serum. It was found that at the temperature 0° – 3°C . the red cells had combined with all of the substance sensibilatrice, but had left the alexin practically untouched.

It still remained to show the relation of these two substances to the red cells at higher temperatures. At 37° – 40°C ., as already mentioned, hæmolysis occurs rapidly, beginning usually within fifteen minutes. It was possible, therefore, to leave the cells and serum in contact for not over ten minutes. Then the mixture was centrifuged as before. The sedimented blood cells mixed with normal salt solution showed hæmolysis of a moderate degree. The solution became complete when

a little normal serum was added. The supernatant clear fluid separated by the centrifuge did not dissolve sheep red cells. On the addition, however, of substance sensibilatrice it dissolved them completely.

From this experiment Ehrlich concludes that the substance sensibilatrice possesses one combining group with an intense affinity (active even at 0°C.), for the red cell, and a second group possessing a weaker affinity (one requiring a higher temperature) for the alexin.

Nomenclature.—In place of the name substance sensibilatrice Ehrlich first introduced the term *immune body*; later on he called it the *amboceptor*, to express the idea that it served as a link between alexin and cell. Other names proposed for this substance have been *substance fixatrice* by Metchnikoff, *copula*, *desmon*, *preparator* by Müller. Instead of the name alexin, Ehrlich now uses the term *complement* in order to express the idea that this body completes the action of the immune body.

According to Ehrlich the red blood cells possess specific affinity for the immune body, but none whatever for the alexin. The alexin, therefore, possesses no combining group which can attach itself directly to the red blood cell. It acts on these cells only through an intermediary, the immune body, which therefore must possess two binding groups one of which attaches to the red blood cell and the other to

the alexin of normal serum. As already stated, the group which attaches to the red blood cell possesses a much stronger affinity than that which combines with the alexin. This follows from the last two experiments of Ehrlich before cited, in which he showed that at the lower temperature, and with both substances present with the blood cells, only the immune body combined with the cells, while the alexin remained uncombined. At the higher temperature the alexin also exerted its affinity, for then the red cells combined with all the immune body and with part of the alexin. We saw that after a time the red cells partially dissolved, but that complete solution occurred only after some fresh alexin had been added. This showed that although the red cells had combined with all the immune body necessary for their solution, they had been unable to bind all the alexin necessary. We may say, therefore, that that group of the immune body which combines with the red cell has a stronger affinity than that which combines with the alexin.

Rôle of the Immune Body. — According to Ehrlich, then, the rôle of the immune body consists in this, that it attaches itself to the red cell on the one hand, and to the complement on the other, and in this way brings the digestive powers of the latter to bear upon the cell, the complement possessing no affinity for the red cell. Immune body and complement

have no very great affinity for each other. At 0°C . they may exist in serum side by side, and they combine only at higher temperatures.

The amount of immune body which combines with the red cells may vary greatly, as the experiments of Bordet and of Ehrlich clearly show. Some red cells combine with only just enough immune body to effect their solution. Others are able to so saturate themselves with immune body that they may have a hundred times the amount necessary for their solution.

On what the Specificity Depends. — From the preceding it follows that the specific action of the hæmolytic sera, and, I may at once add, of the bactericidal sera also, is due exclusively to the immune body. This possesses a combining group which is specific for the cells with which the animal was treated; e.g., the combining group of an immune body produced by treatment with rabbit blood will fit only to a certain group in the blood cells of rabbits; an immune body produced by treatment with chicken blood will fit only to parts of the red cells of chickens; one produced by treating an animal with cholera bacilli will fit only to this species of bacteria and combine only with the members of it. Keeping to the well-known simile of Emil Fischer, the relation is like that between lock and key, each lock being fitted only by a particular key.

To repeat — for the point is of the greatest importance — the rôle of the immune body consists in tying the complements of normal serum, which have no affinity for the red cells or for the bacteria, indirectly to these cells so that their solution and digestion may be effected by the complements. In other words, the immune body serves to concentrate on the corpuscular element to be dissolved all the widely distributed complement found in normal serum.

Ehrlich's conception of the relation existing between complement, immune body (i.e., amboceptor) and erythrocyte is shown in the accompanying figure.

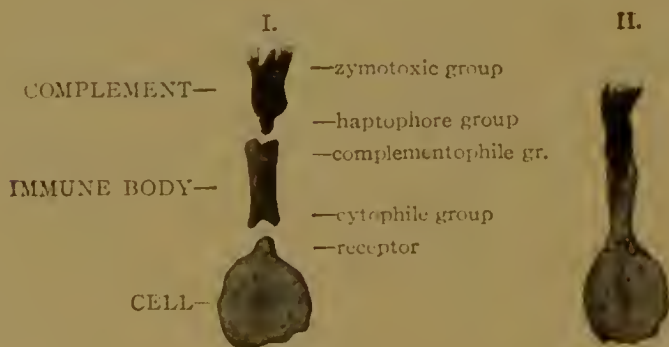


FIG. 7

Difference between a Specific Serum and a Normal One. — The difference, then, between a specific hæmolytic or a specific bactericidal serum and a normal one consists in this — that the specific serum contains an immune body which is specific for a

certain cellular element and by means of which the complement present in all normal serum can be concentrated on this element to cause its solution. We shall return to this subject later.

Diverging Views of Ehrlich and Bordet. — Now if we recall the first experiments of Bordet and his conclusions respecting the manner in which the factors concerned acted, we shall at once see how Ehrlich and Bordet differ. Bordet assumes that the substance sensibilatrice (the immune body) acts as a kind of mordant on the red cells or bacteria, sensitizing these to the action of the alexin (complement). That is to say, neither the cell nor the immune body has alone any manifest affinity for the alexin, but they form by their union a complex which can absorb alexin, in other words which has particular properties of adhesion. According to Bordet, then, *there is no such thing as an amboceptor, and no complementophile group.* He cites the experiments of Muir as showing that the hypothesis of a complementophile group is untenable. This author found that blood corpuscles which had fixed the sensitizer (immune body) and had been saturated with alexin could subsequently, by diffusion, lose a certain amount of their sensitizer, although they retain the complement, and what is more, in this instance they lose as much sensitizer as if they had not absorbed complement. Consequently, says Bordet, it is in no way through the

mediation of the sensitizer that the alexin attaches itself to the corpuscles; if this were the case the removal of the sensitizer would necessarily imply that of the alexin (complement), which, as we have just mentioned, does not leave the corpuscles.

According to Ehrlich, however, the process is not analogous to a staining process, but follows definite laws of chemical combination, there being, in fact, no affinity whatever between the complement and the blood cells or bacteria. Furthermore, according to this authority, the complement always acts through the mediation of the immune body, which possesses two combining groups; one, the cytophile group, combining with the cell, and another, the complementophile group, combining with the complement.

The Side-Chain Theory Applied to these Bodies. —

All of the specific relations which, in a previous chapter, we saw existed between toxin and anti-toxin, Ehrlich and Morgenroth in their experiments above noted found existed also between immune body and the specific blood cell. The immune body must therefore possess a haptophore group which fits exactly to certain receptors or side chains of the red cells, just as the anti-body according to the side-chain theory possesses a group that fits exactly into the specific combining group — i.e., haptophore group — of the toxin or toxoid used for exciting the immunity.

If, for example, we produce a hæmolytic serum specific for red cells of a rabbit by injecting an animal with these cells, the haptophore groups of this serum, i.e., the free side chains thrust off, must possess specific combining relations with the red cells of rabbits. That such is the case in the hæmolytic immune serum we saw from the experiments of Ehrlich and Morgenroth.

In consequence of all this, Ehrlich widened the application of his side-chain theory so as to include not only the production of antitoxin but also the production of bactericidal, hæmolytic, and other immune bodies. He expressed this somewhat as follows: *If any substance, be it toxin, ferment, constituent of a bacterial or animal cell, or of animal fluid, possess the power by means of a fitting haptophore group to combine with side chains (receptors) of the living organism, the possibility for the overproduction and throwing off of these receptors is given, i.e., the possibility to produce a corresponding anti-body.*

Specific anti-bodies in the serum as a result of immunizing processes can only be produced, therefore, by substances which possess a haptophore group¹ and which, in consequence, are able to form a firm union with a definite part of the living organisms, the receptor. This is not the case with

¹ Such substances Ehrlich terms "haptins." See page 16.

alkaloids, e.g., morphine, strychnine, etc., which according to Ehrlich enter into a loose union, a kind of solid solution with the cells. It is for this reason that we are unable to produce any anti-bodies in the blood serum against these poisons. Ehrlich says further that all of the substances taking part in the production of immunity, including of course complement and immune body, have certain definite affinities for each other, and in order to act they must fit stereochemically to each other.

As we have already seen, we are able by means of the injection of a variety of substances or cells to produce a similar variety of immune bodies in the serum. Thus we can immunize a rabbit so that its serum will possess specific hæmolytic bodies against the red cells of guinea pigs, goats, chickens, and oxen and specific bactericidal bodies against cholera and typhoid bacilli, etc., and as we shall see, still other groups of anti-bodies.

Multiplicity of Complements. — Under these circumstances an important question presents itself: Is there in normal serum one single complement which completes the action of all these various immune bodies, one, for example, which in the above illustration will fit all the hæmolytic immune bodies as well as all the bactericidal ones, or are there a great many different complements?

Ehrlich, as a result of his experimental work with Morgenroth, claims that the latter is the case; namely, that it takes a different complement to fit the immune body specifically hæmolytic for guinea pig blood than it does to fit that specific for chicken blood.

Bordet, on the other hand, assuming that the immune body plays the rôle of mordant, believes that there is but one single complement in the serum. According to him, this complement is able to dissolve blood cells as well as bacteria after these have been sensitized by their specific immune body. Each of these authors supports his claims by means of ingenious experiments, for the details of which, however, we must refer to the original articles, as they require the knowledge of a specialist for their comprehension. We shall, however, give one of Bordet's ¹ experiments on this point in some detail, since it has found extensive application in another direction.

The Bordet-Gengou Phenomenon. — Bordet sensitized blood corpuscles with appropriate amboceptors, and then exposed them to the action of a freshly drawn normal serum. If now he waited for the occurrence of hæmolysis and then added *sensitized* cells (bacteria or blood corpuscles of a different species), the latter remained entirely unchanged, although the serum that had been used as complement was capable in its original con-

¹ Bordet and Gengou, *Annal. Inst. Pasteur*. Vol. xv, 1901.

dition of destroying these also. When fresh serum was first brought into contact with sensitized *bacteria*, similar results were obtained. The blood corpuscles subsequently added did not then undergo hæmolysis. *If such an action on one of the sensitive substrata has once taken place, the active sera, as a rule, are deprived of all their complement functions, from which Bordet concludes that the destruction of the most varied elements by one and the same serum must be due to a single complement.*

It may be said in passing that Ehrlich admits the correctness of the above experimental results, but brings forward arguments to show that Bordet's interpretation as to the existence of only a single complement cannot be accepted.

This experiment of Bordet is usually spoken of as the "Bordet-Gengou phenomenon" and is now used largely in determining whether or not a given serum possesses certain amboceptors. The serum to be tested is first heated and then mixed with a small quantity of fresh normal serum (complement) and with an emulsion of the bacterium whose amboceptors it is desired to discover. After standing for six hours at room temperature, red blood cells previously treated with heated hæmolytic serum are added. If there is no hæmolysis it is held to mean that the complement in the fresh serum which was suitable for lysis of properly prepared blood corpuscles, has been absorbed by the bacteria by reason of the presence of specific amboceptors in the serum tested.

Wassermann¹ has attempted to apply this method in measuring the amboceptor content of specific mening-

¹ Wassermann, Neisser and Bruck, *Deutsche med. Wochenschr.*, 1906; Wassermann and Plaut, *Ibid.*

gococcus sera and has successfully adapted the Bordet-Gengou test to the diagnosis of syphilitic infections.¹

Neisser and Sachs² have recently described a procedure for the forensic diagnosis of blood stains. The principle of this is the same as in the preceding, although in so far as a specific *precipitin* serum is made use of, the procedure is really modelled after the "Gengou-Gay" phenomenon.

If human blood serum is mixed with a specific human precipitin serum derived from rabbits, it will be found that the mixture binds complement. Hæmolysin subsequently added is unable to dissolve its specific red blood cells, owing to this locking up of the complement. Only the serum of monkeys has a similar effect. The amount required is extremely minute, 100000 to 1000000 cc. human blood or monkey blood sufficing. Extracts of human blood stains will also produce the desired effect. The authors believe that the immunization with human blood serum gives rise not only to precipitins but also to amboceptors which then are able to unite with their corresponding unformed albuminous bodies and so bind complement. Others are of the opinion that the complement is bound by the precipitin-precipitum combination.

The test is extremely delicate and has been found trustworthy by a number of investigators. In view of the importance of such tests in medico-legal cases, Neisser and Sachs suggest that it should always be used in addition to the well known Wassermann-Uhlenhuth precipitin test.

Normal Serum, its Hæmolytic and Bacteriolytic Action. — Inquiring now into the essential differ-

¹ See also page 185.

² Neisser and Sachs, *Berliner klin. Wochenschrift*, 1905.

ence between a specific hæmolytic or bactericidal serum and a normal one, we must first of all study the behavior of normal serum toward alien red cells and bacteria. It has long been known to physiologists that fresh normal serum of many animals has the power to dissolve blood cells of another species. This was studied especially by Landois. One-half to one c.c. of normal goat serum, for example, is able to dissolve 5 c.c. of a 5% mixture (in normal salt solution) of rabbit or guinea pig red cells. In the same way, these red cells are dissolved by the sera of oxen, of dogs, etc. This *normal globulicidal* property of the serum corresponds to another which fresh normal serum was found to possess, namely, the property to dissolve appreciable quantities of many species of bacteria. This analogy was pointed out by Fodor, Nutall, Nissen, and especially by Buchner. We call this the *bactericidal* property of fresh normal serum.

This property is well illustrated by the following protocol from Park.

No. of bacteria in 1 cc. fluid.	Amount of serum added.	Approximate number alive after being kept at 37° C.		
		One hour.	Two hours.	Twenty seven hrs.
30,000	0.1 cc.	400	2	0
100,000	0.1 cc.	5,000	1,000	200,000
1,000,000	0.1 cc.	400,000	2,000,000	10,000,000

It is at once apparent that the number of bacteria introduced is an important factor, the normal serum being able to kill off only a certain number.

Buchner, as we have already seen, had studied this bactericidal action carefully and ascribed the action to a substance found in all normal serum, which he called *alexin*. According to his experiments, this is a very unstable substance, decomposing spontaneously on standing or on heating for a few minutes to 55° C., or readily on the action of chemicals. According to this author all the globulicidal and bactericidal functions of normal serum are performed by this one substance, the alexin.

Active and Inactive Normal Serum.—In taking up the study of the hæmolytic action of normal serum Ehrlich and Morgenroth sought particularly to discover whether in normal serum the hæmolytic property depended on the action of a single substance, the complement (Buchner's alexin), or whether here as in the specific hæmolytic serum it depended on the combined action of two substances. For this purpose they used guinea-pig blood, which is dissolved by normal dog serum. If this serum was heated to 55° C., it lost its hæmolytic power. It was necessary now to show that in this inactive dog serum there remained a second substance which could be reactivated after the manner of reactivating an old specific hæmolytic serum. This had its difficulties, for they could not add normal dog serum. This, as we saw, is already hæmolytic for guinea-pig

blood. "Possibly," said they, "there exists a complement of another animal which will fit the hypothetical second substance of this dog serum." This proved to be the case, the complement of guinea-pig blood fulfilling the requirements. If they added to the inactive normal dog serum about 2 c.c. normal guinea-pig serum the hæmolytic property was restored and the guinea-pig red cells dissolved completely. According to Ehrlich, this can only be explained by assuming that in guinea-pig blood there exists a complement which happens to fit the haptophore group of the second substance or inter-body, of the normal dog serum. This combination of guinea-pig blood, inactive normal dog serum, and a reactivating normal guinea-pig serum is well adapted to demonstrate the existence in normal dog serum of an inter-body; for the guinea-pig serum should be the best possible preservative for the guinea-pig red cells. The hæmolysis following the addition of this serum shows positively the existence of a substance in the dog serum which has acted with something in the guinea-pig serum.¹

¹ Of such combinations, i.e., combinations in which a complement derived from the same animal from which the red cells are derived fits to the inter-body of other species of animals, causing the solution of red cells of the latter, Ehrlich and Morgenroth found still other examples. For instance, guinea-pig blood, inactive calf serum, guinea-pig serum; goat blood, inactive rabbit blood, goat serum; sheep blood, inactive rabbit blood, sheep serum; guinea-pig blood, inactive sheep serum, guinea-pig serum.

Inter-body and Complement. — We see, then, that the hæmolytic action of normal sera depends, just as that of the specific hæmolytic sera, on the combined action of two bodies: one, the *inter-body*, which corresponds to the immune body of the specific sera, and a second or *complement*. In speaking of the constituents of *normal* serum, Ehrlich and Morgenroth prefer to use this term *inter-body* to distinguish it from the *immune bodies* of *specific* hæmolytic sera.

Action not Entirely Specific. — It has also been found that there frequently exist normal sera which are hæmolytic not only for one species of red cell, but for several. We saw, for instance, that normal goat serum dissolved the red cells of guinea pigs and rabbits. The question now arises, Is this property of normal goat serum due to two inter-bodies existing in the serum side by side, one fitting the red cells of the guinea pig, the other those of the rabbit? Ehrlich and Morgenroth answered this in the affirmative, for in the following experiment they succeeded in having each of the two inter-bodies combine with its respective cell. To some inactive normal goat serum they added rabbit blood and centrifuged the mixture. To the separated clear fluid they again added some rabbit red cells as well as normal horse serum to reactivate the mixture. Horse serum is not hæmolytic for rabbit red cells. The mixture remained

unchanged, no hæmolysis taking place. If, however, they added some of this normal horse serum to the centrifuged red cells, the latter immediately dissolved. Now, to the clear centrifuged fluid, which as we have seen would not dissolve rabbit red cells, they added guinea-pig red cells and again some normal horse serum to reactivate the mixture. The guinea-pig red cells all dissolved. This proved conclusively that in the normal goat serum there had existed two specific inter-bodies. One, for rabbit red cells, had been tied by these cells and carried down with them in centrifuging; the other, specific for guinea-pig red cells, had remained behind.

Multiplicity of the Active Substances. Further investigation led these authors to assume a still greater multiplicity in the substances in normal serum which are concerned in hæmolysis. In addition to the two interbodies just mentioned, they demonstrated the existence in goat serum of two specific complements, one for each interbody, and they were able by means of Pukall filters to separate these two. In this filtration the complement fitting the inter-body for rabbit blood remained behind for the greater part, while that fitting the inter-body for guinea-pig blood mostly passed through.

Whereas then, according to Buehner, only one substance, the alexin, is concerned in the hæmo-

lytic action of this normal goat serum, these experiments of Ehrlich and Morgenroth show us four substances, viz., two inter-bodies and two complements. This at once makes clear the opposing views of these authorities. According to Ehrlich, however, the number of active substances in normal serum is still greater, for it often happens that a specific inter-body shows itself to be made up of several inter-bodies, all, to be sure, fitting the same specific red cell, but differing from each other by their behavior toward different complements. Ehrlich, therefore, regards the substances concerned in hæmolysis which occur in normal serum to be of great number and variety.

Difference between a Normal and a Specific Immune Serum. — Practical Application. — Returning now to the question of the difference between a specific immune serum and a normal one, we find this to be as follows: Normal serum contains a great variety of inter-bodies, in very small amounts, and a considerable amount of complements. In immune serum, on the other hand, the amount of a specific inter-body, the one which fits the haptophore group of a certain cell, is enormously increased. This specifically increased inter-body, it will be remembered, is called the immune body. The complement, as shown by v. Dungern, Bordet, Ehrlich and Morgenroth and Wassermann, is in no way

increased by the immunizing process. The increase affects solely the immune body. It is therefore possible to have a serum which contains more immune body than complement to satisfy it, and if we withdraw such a serum from an animal we shall find that it contains some free immune body. This serum can only then exert its full power when the full amount of complement is present, i.e., when some normal serum is added. If we treat a rabbit with the red cells of an ox, as v. Dungern did, we shall obtain a serum which is hæmolytic for ox blood. Of this freshly drawn serum 0.05 c.c. suffice to dissolve 5.0 c.c. of a 5% mixture of ox blood. If now we add to this hæmolytic serum a little normal rabbit serum, we shall find that only one-tenth of the amount of serum is required; i.e., only 0.005 c.c. to dissolve the same quantity of ox blood. This means that through the addition of the rabbit serum, which, of course, is not hæmolytic for ox blood, a sufficient amount of complement was added to enable all the immune body of the specific serum to act. This specifically increased power of the immune serum to act on certain definite cells depends on the fact that the immune body resulting from the immunizing process concentrates the action of the complement scattered through the serum, on cells for which it has definite affinities. If 2 c.c. of normal guinea-pig serum are able to dissolve, we will say

5 c.c. of a 5% defibrinated rabbit-blood mixture, and if we find that after the immunizing process 0.05 c.c. of the guinea-pig serum suffice to dissolve the same amount of rabbit blood, we conclude that through this process the inter-body, i.e. the immune body, has been increased forty times. We know that the complement has not been increased, but this is now able to act by means of forty times increased combining facilities. This increase, however, is exclusively for rabbit-blood cells. In a bactericidal immune serum this specific increase is sometimes as much as 100,000 times that of normal serum.

The practical idea to be gained from this for the therapy of infectious diseases is this: that with the injection of an immune serum we supply only one of the necessary constituents to kill and dissolve the bacteria, and that is the immune body.

We do not, however, supply the second, i.e. the complement, for this we have seen is not increased by the immunizing process. As matters stand, then, the use of a specific immune serum for therapeutic purposes assumes that the complement which is essential for the action of the immune body will be found in the organism to be treated. Since in certain infectious diseases the required complement is present in too small amounts in the organism, Wassermann suggested that the

curative power of many bactericidal sera might be increased by the simultaneous injection of the sera of certain normal animals in order thus to gain an increased amount of complement; but we shall soon see that this procedure, while of great value in animal experiments, presents certain difficulties.

Nature of the Immune Body — Partial Immune Bodies of Ehrlich — Turning now to a closer study of the nature of the immune body, we again find a difference of opinion. Whereas Bordet, Metchnikoff, and Besredka assume each immune body to be a single definite substance, Ehrlich and Morgenroth as a result of their experiments hold to a plurality of bodies.

These authors say that each immune body is built up of a number of *partial-immune bodies*, a point to which we have already alluded. In support of this view they offer the following experiment. On immunizing a rabbit with ox blood, they obtained a serum hæmolytic not only for ox blood but also for goat blood; on immunizing a rabbit with goat blood they obtained a serum hæmolytic for goat blood and ox blood.¹

The conditions present can be readily understood by reference to Fig. 7, which represents schematically three portions of the combining groups

¹ We have already called attention to these exceptions to the rule of specific action.

of the blood cells. Of these α is present only in the ox-blood cells, ψ only in the goat-blood cells, and β in both. If a rabbit is injected with ox blood, the immune bodies corresponding to groups α and β will be formed. On subjecting such a serum to absorption with ox-blood cells we shall find that these, by means of their α and β groups will be able to absorb *all* the immune bodies, whereas goat-blood cells will in a similar test absorb only the immune

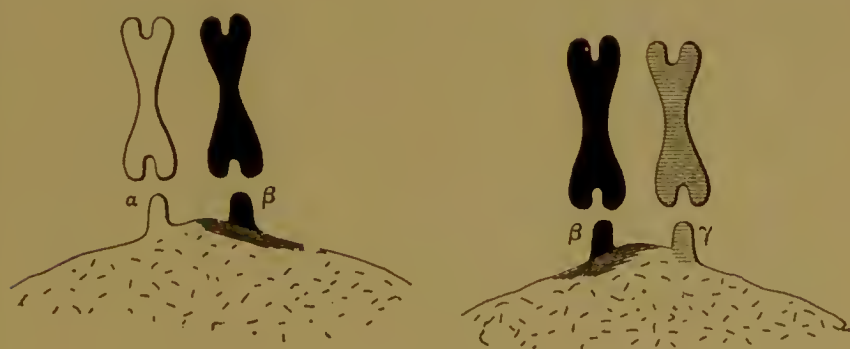


FIG. 7

body of portion β , leaving the immune body of portion α in solution.

According to Ehrlich's theory, then, the red cells of the ox possess certain receptors which are identical with receptors possessed by the goat red cells. From this it follows that in a single red cell there are several or many groups each of which is able, when it finds a fitting receptor, to take hold of a

single immune body. Ehrlich and Morgenroth, therefore, claim that the immune body of a hæmolytic serum is composed of the sum of the partial immune bodies which correspond to the individual receptors used to excite the immunity. It may be assumed, then, that not all of the combining groups of a cell, be this a blood cell or a bacterium, will find fitting receptors in every animal organism, and that therefore not all the possible partial immune bodies will be equally developed. In one animal there may be receptors which are not present in another, and in this way there might be a different variety of partial immune bodies in the two animals. This would lead to the possibility of the occurrence of immune bodies, for the same species of blood cell or bacterium, differing from each other in the partial immune bodies composing them, according to the variety of animals used in preparing the serum.

Metchnikoff's Views — Practical Importance of the Point. — This view is directly opposed to that of Metchnikoff and Besredka, who believe that a certain immune body, e.g. one specific for ox blood, is always the same no matter from what animal it is derived. The point is not merely theoretical, but under certain circumstances of great practical importance. If we believe, as Ehrlich does, that the immune body differs according to the species of animal from which it is derived, i.e., that it is made

up of different partial-immune bodies, then we must admit that we have better chances for finding fitting complements if we make use of immune bodies derived from a variety of animals. We would, for instance, be likely to achieve better results in treating a typhoid patient with a mixture of specific bactericidal typhoid sera derived from a variety of animals than if we used a serum derived only from a horse. For in such a mixture of immune bodies the variety of partial-immune bodies must be very great and the chances that the complements of the human body will find fitting immune bodies, and so lead to the destruction of the typhoid bacilli, are greatly increased. Ehrlich and his pupils have actually proposed such a procedure in the use of bactericidal sera for therapeutic purposes.¹

Support for Ehrlich's View. — Besides the above experiments we possess others which support the theory that the immune body is not a simple but a compound substance. v. Dungern had already shown that following the treatment of an animal with ciliated epithelium from the trachea of an ox, there were developed immune bodies which acted not only on the ciliated epithelium but also on the red cells of oxen. We must assume, therefore, that

¹ Reasoning along similar lines, namely, that the human complement must fit the immune body of the therapeutic serum, Ehrlich has also proposed that these bactericidal sera be derived from animals very closely related to man, e.g., apes, etc.

the ciliated epithelium and the red cells of the ox possess common receptors. Analogous to this is the action of the immune body resulting from the injection of spermatozoa, as was pointed out by Metchnikoff and Moxter.

We see, then, that the specific action of immune bodies is not so limited as to apply only to the cells used in the immunizing process, but extends to other cells which have receptors in common with these.*

So far as concerns the site in the organism where the substances used in immunizing find their receptors, this is not known for the hæmolytic immune body. For the bactericidal immune bodies of cholera and typhoid the researches of Pfeiffer, Marx, and others show that the chief site of production is in the bone-marrow, spleen, and lymph bodies. Wassermann's experiments on local immunity indicate that the site of infection determines largely the site of the development of the immune bodies.

Antihæmolysins: their Nature — Anti-complement or Anti-immune Body? —A further step in the study of hæmolysins is one discovered independ-

* The same holds good for the agglutinins and the precipitins still to be studied. In these the action extends also to closely related cells and bacteria, or in the case of the precipitins to closely related albumins, as these possess a number of receptors which are common to them and to the cells or substances used for immunizing.

ently by Ehrlich and Morgenroth on the one hand, and Bordet on the other. These authors succeeded in producing an *antihæmolysin*. The procedure is closely related to the results gained by immunization against bacterial poisons. A specific hæmolyisin, one, for example, specific for rabbit blood, derived by treating a guinea pig with rabbit red cells, is highly toxic to rabbits. Injected into the animals intravenously in doses of 5 c.c. it kills the animals acutely, causing *intra vitam* a solution of the red cells. Such a hæmolytic serum, then, acts the same as a bacterial poison, and it is possible to immunize against this just as well as against a bacterial poison. For example, to keep to our illustration, rabbits are injected first with very small doses of this specific hæmolytic serum. The dose is gradually increased until it is found that the animal tolerates amounts that would be absolutely fatal to animals not so treated. If some of the serum of this animal is now abstracted and added to the specific hæmolytic serum, it is found that the power of the latter will be inhibited. According to Ehrlich an *antihæmolysin* has been formed. As we know that the action of the hæmolysin depends on the combined action of two substances, the immune body and the complement, the question arises to which of these two the antihæmolysin is related. Is it an anti-immune body or an anti-complement? A study of this question shows that

both these substances are apparently present. In the serum of the rabbit treated with specific hæmolyisin, both an anti-immune body and an anti-complement have been found. Ehrlich and Morgenroth were further able to show that the action of the anti-complement depended on a haptophore group which it possessed, enabling it to combine with the haptophore group of the complement, thus satisfying this and hindering its combination with the complementophile group of the immune body.

“*Anti-complement.*”¹—Since the complements are constituents of normal serum, it should be possible to produce anti-complements by injecting animals merely with normal serum; and they can, in fact, be so produced. If rabbits are treated by injecting them several times with normal guinea pig serum, a serum may be obtained from these rabbits which contains anti-complements against the complements of normal guinea-pig serum. A serum obtained in this way of course contains only one of the anti-hæmolytic bodies, the anticomplement, and not the anti-immune body. This is because normal serum is too poor in immune body (inter-body) to excite the production of any anti-immune body.

¹ The existence of anti-complements is denied by Bordet, Gay and others. For a statement of their views see page 96.

If to a hæmolytic serum derived from guinea pigs we add an anticomplement serum derived, as just stated, from rabbits, and containing an anticom-

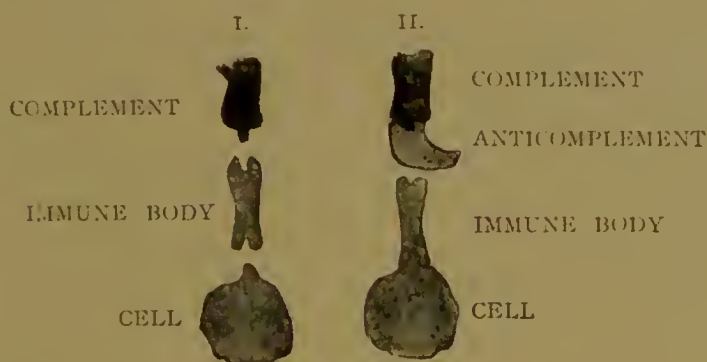


FIG. 8. (After Levaditi.)

plement specific for guinea-pig complement, the hæmolytic action of the former will be inhibited, for the reason that the complement necessary for the hæmolysis to take place has been bound by the anticomplement. (See Fig. 8.) One must, however, observe the precaution to heat the anticomplement serum of the rabbit to 55°C . before so mixing it, in order to destroy the complement which it contains and which would otherwise reactivate the guinea-pig immune body.

From the foregoing we see that either anti-immune body alone, or anticomplement alone, is able to inhibit the hæmolytic action. Hæmolysis cannot take place when either of the two

necessary factors is bound and prevented from acting.¹

The action of anticomplements is specific, i.e., an anticomplement combines only with its specific complement. Thus an anticomplement serum derived from rabbits by treatment with guinea-pig serum combines only with the complement of normal guinea-pig serum, not, however, with the complements of other animals. Exceptions to this are those cases in which the complement of the other species possess receptors identical with those of the first.

In order that a normal serum of species *A*, injected into species *B*, produce anticomplements there, the side-chain theory demands that the complements of *A* find fitting receptors in species *B*. According to Ehrlich, however, normal serum contains many different complements and not merely a single one. Under the circumstances, it is easily possible that only a few of the complements in the serum of *A* find fitting receptors in species *B*. We shall then obtain an anticomplement serum which inhibits the action of some, but not of all the complements of species *A*. Thus it might inhibit the

¹ By treating animals with normal sera of certain other species, it is possible to produce not only anti-complements, but also specific anti-bodies against certain other constituents of normal serum. These are, for example, anti-agglutinins, which inhibit the action of the haemagglutinins of normal serum, and anti-precipitins, which we shall discuss later.

action of a complement fitting to a certain bactericidal immune body and not of one contained in the same serum which fitted a certain hæmolytic immune body, etc.

Fluctuations in the Amount of Complement and other Active Substances in the Blood.—We have come to know certain conditions under which there may be a decrease of certain complements in normal serum. Ehrlich and Morgenroth showed that in rabbits poisoned with phosphorus and in whom, therefore, the liver was badly damaged, the serum on the second day (the height of the disease) had lost its power to dissolve guinea-pig blood, and that this was due to a disappearance of the complement. Metchnikoff also reported that in an animal suffering from a continuing suppurating process, the complement had fallen considerably in amount. Especially interesting are the experiments of v. Dungern, who showed that animal cells, hence emulsions of fresh organs, are able to attract and combine with complements.

Even more important than the question of a decrease in complements, or an inhibition of their action, is that of the possibility to artificially increase them. A number of authors, among them Nolf and Müller, have answered this question in the affirmative. They believe they have noticed such an increase following the injection of an animal with

all sorts of substances, such as normal serum of another animal, sterile bouillon, etc. v. Dungern, Wassermann and others, have not been able to convince themselves of the possibility of such a definite increase. Wassermann tried to excite the increased production of complement by injecting guinea pigs for some time with anticomplement. This being the opposite of the complement, he hoped to be able by immunizing to excite an increase of the complements. In this he was unsuccessful, though of course it may be possible with another species of animal.

Despite all this, we must believe that the amount of complement, as well as the amount of other active substances of the blood, inter-bodies, agglutinins, antitoxins, ferments, antiferments, etc., is subject to great fluctuations even in the same individual, a constant change going on within the organism. Ehrlich, in particular, has pointed out these individual and periodic variations and has insisted on their importance. Very likely, under circumstances of which we now know very little, these substances are at certain times produced in greater amounts, at other times in lesser; sometimes they may be absent entirely in an individual in whom they were previously present. For example, the serum of a dog will at times dissolve the red cells of cats, rabbits, and guinea pigs, at other times not. Furthermore, the serum of one and the same animal may

possess specific hæmolytic properties for certain cells, and later on may lose this property entirely. In human serum these same individual and periodic variations may be demonstrated, as Wassermann was able to prove experimentally. However, the circumstances on which these variations depend are as yet entirely unknown to us. Possibly we are dealing here with subtle pathological changes.

Source of the Complements — Leucocytes as a Source — Other Sources. — *Where do the complements or alexins originate?* This question has been studied particularly by Metchnikoff and by Buchner; also by Bail, Hahn, Schattenfroh, and others. These investigators believe that the leucocytes are the source of the complements or alexins. There is, however, this difference between the views of Metchnikoff and Buchner: whereas Buchner believes the alexins to be true secretory products, Metchnikoff believes that they originate on the breaking up of the leucocytes, i.e., that they are decomposition products. Metchnikoff bases his belief chiefly on the work of his pupil, Gengou, who showed that although the serum was rich in alexin (i.e., complement) the plasma contained none at all.

Other authors, as Pfeiffer and Moxter, as a result of their experiments, are not willing to assume the existence of any relationship between the alexins and the leucocytes. Gruber as well as Schattenfroh are ready to believe the leucocytes to be the

source of *an* alexin, but claim that this is different from that found in serum. Wassermann believes that the leucocytes are a source of complements (alexins), for he succeeded in producing anticomplement by means of injections of pure leucocytes which had been washed free from all traces of serum, and which had been obtained by injections of aleuronat. In view of the plurality of the complements, Wassermann expressed the view that the leucocytes are probably *one* source, but not the only one, for the complements of the serum. Landsteiner and Donath have confirmed this experimentally. They succeeded in producing anticomplement by the injection, not only of leucocytes, but of other animal cells. Furthermore, the experiments of Ehrlich and Morgenroth already mentioned, in which the complements disappeared after the destruction of the liver function, show that the liver cells are concerned in the formation of complements.

Structure of Complements—Haptophore and Zymotoxic Groups—Complementoids.—Ehrlich's conception of the structure of complements is based principally on the fact that when complement is heated to 55° C., its complementing power is lost. If animals are injected with such a heated serum, anticomplement will be produced, showing, according to Ehrlich, that heating has not destroyed the entire complement body, but only that part which effects the digesting, solvent action. The haptophore

phore group must have remained intact. Ehrlich therefore concludes that the complements consist of a combining *haptophore group* which withstands heating to 55° C., and another, more fragile group, which possesses the actual solvent properties and which Ehrlich calls the *zymotoxic group*. The complements, according to this conception, are entirely analogous to the toxins already studied.

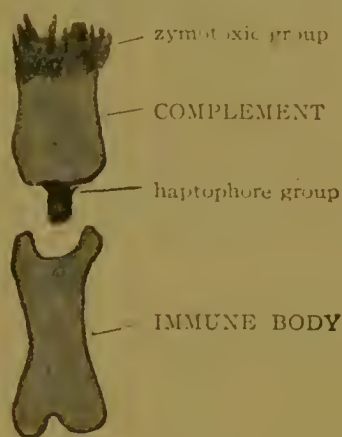


FIG. 9.

And just as those toxins which had lost their toxophore group were called toxoids, so Ehrlich terms complements which have lost their zymotoxic group, *complementoids*.

Bordet, it will be recalled, does not believe in any complementophile group in the amboceptor. So also in the case of the complement, or alexin as he prefers to call it, he refuses to accept Ehrlich's view that this contains a zymotoxic and a haptophore

phore group. The fact that heated complement produces an anticomplementary serum is readily explained by Bordet as due to the absorption of complement by precipitates. An illustration will bring out this point more clearly.

Goat serum heated to 55° C., and therefore containing no active complement, is injected into a rabbit. According to Ehrlich it excites the production of an "anticomplement" in response to the "complementoid" which it contains. As we shall see in the next chapter, the injection of this heated goat serum in addition to whatever else it may do does actually excite the production of a specific precipitin in the serum of the rabbit, so that when such a rabbit serum is mixed with goat serum, a precipitate will be produced. Ehrlich's demonstration of the "anticomplement" is somewhat as follows: Ox blood corpuscles, plus suitable amboceptor (serum of a rabbit injected with ox blood corpuscles), plus fresh, normal goat serum as complement, undergo hæmolysis. When, however, the fresh, normal goat serum is mixed with the serum of a rabbit previously injected with goat serum, and then the above combination carried out, no hæmolysis occurs. The rabbit serum contains an "anticomplement," says Ehrlich. On the other hand, Bordet and Gay believe that the anticomplementary action is due to the absorption of goat complement by the precipitate produced by the mix-

ture of goat serum and its precipitin, the rabbit serum. Moreschi comes to the same conclusion.

Isolysins — Autolysins — Anti-isolysins. — All of the preceding studies in hæmolysis have concerned themselves with the results obtained by injecting animals of one species with blood cells of another. Ehrlich and Morgenroth now sought to discover what the results would be if they injected an animal with blood cells of its own species. They injected goats with goat blood, and found that when the amount injected at one time was large the serum of the goat injected acquired hæmolytic properties for the blood of many other goats, but not for all. The substances thus formed the authors called *isolysins*. These, then, are substances which will dissolve the blood of other individuals of the same species. Substances which dissolve the blood cells of the same individual are called *autolysins*. But autolysins have so far been demonstrated experimentally only once (by Ehrlich and Morgenroth). If one tests the properties of an isolysin of a goat on the blood of a great many other goats, it will be found that this will be strongly solvent for the blood of some, slightly for the blood of others, and not at all for still others.

By using a blood that was readily dissolved by the isolysin, and proceeding in the same series of experiments which we have already studied under hæmolysis, Ehrlich and Morgenroth showed that

the isolysins, like the hæmolysins, consist of an immune body and a complement of the normal serum. The experiments undertaken by these authors were made on thirteen goats, and the surprising fact developed that the thirteen resulting isolysins were all different. For example, the iso-hæmolytic serum of one goat dissolved the red cells of goats *A* and *B*; that of a second goat those of *C* and *D*; of a third those of *A* and *D*, but not of *C*, and so on. If now they produced *antiisolysins* by injecting animals with these isolysins, they found that these antiisolysins were specific; i.e., the anti-isolysin of *A* would inhibit the action only of isolysin of *A*, but not of *C*, etc. These results are of the highest clinical interest, for they show a *difference in similar cells of the same species*, something that had never before been suspected. In the above, the blood cells of species *A* must have a different biological constitution than those of species *C*, etc.

Moss finds that isolysins occur in about 25% of adult human individuals, and that the relative frequency is the same in health and disease, so that the presence of isolysins has no diagnostic significance. The subject has recently acquired importance because of the revival of homologous transfusion, and it is customary now to always test the blood of both donor and recipient prior to carrying out such a transfusion. The technique of the test is described in the appendix.

The fact that after injections of large amounts of

cells of the same species isolysins develop, but that autolysins are almost never formed, caused Ehrlich and Morgenroth to assume that the body possesses distinct regulating functions which naturally prevent the formation of the highly destructive autolytic substance. It is obvious that if there were no such regulating facilities, the absorption of large bloody effusions and hemorrhages might lead to the formation by the organism of autolysins against its own blood cells. Gengou, a pupil of Metchnikoff, believes to have shown experimentally that the destructive action of these autolysins is hindered by the simultaneous production of an auto-antiimmune body which immediately inhibits their action.

In order that isolysins may be formed, it seems necessary to overwhelm the organism once or several times with large amounts of cells or cell products of the same species; to produce, as Ehrlich says, an *ictus immunisatorius*. Wassermann tried, by using various blood poisons, such as hæmolytic sera, toluylenediamine, etc., for a continued length of time, to cause the formation of these isolysins, but without success, although in these experiments each injection was followed by an appreciable destruction of red cells and absorption of their decomposition products. The gradual and even repeated absorption of not too large quantities of decomposed red cells does not therefore lead to the

formation of isolytins; but, as already said, a sudden overwhelming of the organism by large amounts of the cells or their products is necessary.

Deflection of Complement. — In the use of the antitoxic sera, experience has shown that the employment of a large dose is of paramount importance. So far as the antitoxic action is concerned¹ one cannot do harm by giving a large excess. Concerning the action of bactericidal sera, however, the literature contains a number of examples which indicate that here an excess of immune serum is occasionally injurious. Perhaps the earliest protocol of this kind is that published by Löffler and Abel² on their experiments with *bacillus coli* and a corresponding immune serum. Out of nineteen guinea pigs which had been inoculated with the same amount of culture and had received varying amounts of immune serum, only six animals were protected, those which had received doses of 0.25 c.c. to 0.02 c.c. Eight animals with larger doses, as well as five with smaller doses of serum died. Neisser and Wechsberg³ encountered the same phenomenon in bactericidal *test-tube* experiments, and concluded as a result of their experiments

¹ We shall discuss the rash production, or "serum sickness," page 141.

² F. Löffler and R. Abel, *Centralblatt Bacteriol.*, 1896, Vol. xix, p 51

³ M. Neisser and F. Wechsberg, *Münch. med. Wochenschrift*, 1901. No. 18.

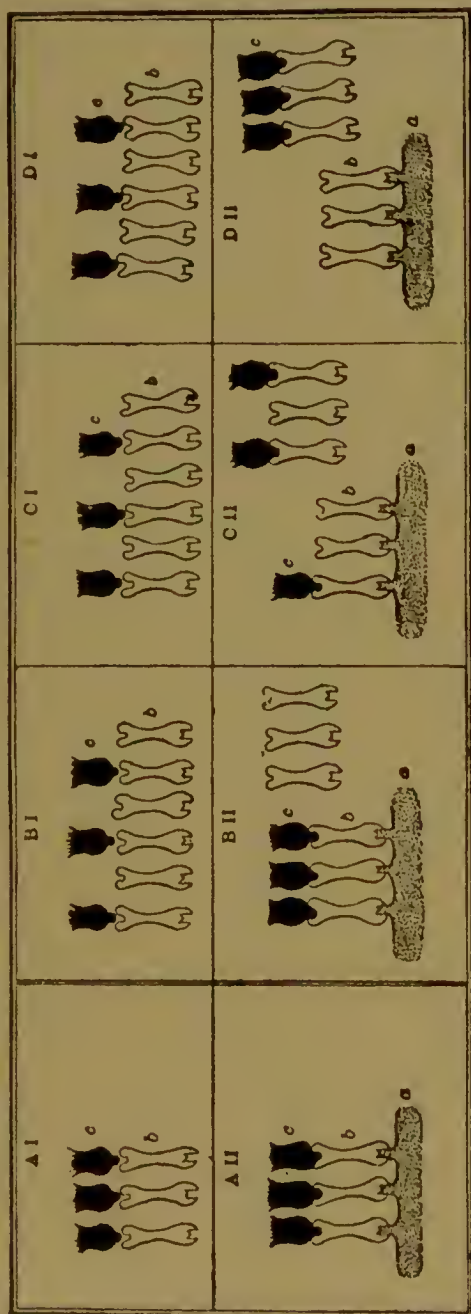


FIG. 10

that the only satisfactory explanation was one based on the views of Ehrlich and Morgenroth. In Fig. 10, *A II* represents schematically a bacterium *a* with a number of receptors; for there are many reasons for assuming that each bacterium possesses a number of receptors of the same kind. According to the side-chain theory, if we inject this bacterium into an animal an over-production of the corresponding group will occur, resulting in a serum which is rich in body *b*. This body *b*, however, is not able by itself to injure the bacteria, and a bacterium all of whose receptors are laden with *b* need not at all be injured in its vitality. Body *b* normally possesses a peculiar function, namely, to serve as a coupling member or link, and hence it possesses two groups (amboceptor). As has already been discussed, one of these groups fit the receptors of the bacterium on the one hand and the complement on the other. When, therefore, to a normal serum which contains suitable complement, we add equivalent amounts of immune serum, the condition pictured in *A I* will result. On adding the corresponding bacterium to this we get the condition shown in *A II*, in which all the bacterial receptors are occupied with immune bodies, or more accurately, with immune bodies which on their part are loaded with bacteriolytic complement *c*. In the case here presented let us say that it requires the occupation of all of the receptors

with complemented interbodies to cause the death of the bacterium.

If now to an equivalent mixture of complement and inter-body we add an excess of inter-body, it will be possible for only a *part* of the inter-body to be loaded with complement, leaving a portion of the inter-body uncomplemented. On adding the corresponding bacteria a number of conditions may result; the affinity of the inter-body for the bacterial receptor may, as a result of the loading with complement, (1) remain *unchanged*, (2) it may thereby be *increased*, or (3) be *diminished*.

In the figure, *B II* shows the condition of increased affinity. Of the six inter-bodies only those combine with the bacterium which have become laden with complement. In this case, therefore, the excess of inter-bodies will have no influence on the bactericidal effect. The condition is really the same as *A II*, except that free inter-body is also present.

C II shows the condition of unchanged affinity. In this case, if we add the bacterium to the mixture of complement and excess of inter-body, all the receptors of the bacterium will, to be sure, be occupied by inter-bodies, but this will be entirely without regard to the fact that these inter-bodies are or are not loaded with complement. It may therefore happen that only a few of the bacterial receptors will be occupied by complemented (i.e., active)

inter-bodies, while the rest of the bacterial receptors are occupied by uncomplemented (hence inactive) inter-bodies. As already stated, however, the vitality of such a bacterium is not necessarily destroyed.

D II represents the last conceivable case. It is assumed that the "completion" of the inter-body has resulted in a diminution of the latter's affinity for the bacterial receptor. In this case primarily only the uncomplemented inter-bodies will combine with the bacterial receptors, while the free fluid will contain complemented inter-bodies.

In cases *C II* and *D II*, therefore, the *excess of inter-body exerts a deflecting action on the complement*, thus diminishing the end results.

It is difficult to say to what extent "deflection of complement" really occurs in the experiments referred to above. Studies by Buxton, Gay, and others show that deflection of complement will not always explain the phenomenon, and that in these instances other factors must be responsible for the paradoxical results.

Practical Value of Injections of Anti-Bacterial Sera.—We use the term "antibacterial" advisedly, because, after all, when we immunize an animal against a certain bacterium we do not produce merely a bacterioid serum, but one which contains agglutinins, precipitins, opsonins, and perhaps still other antibodies as well. The use of

specific antibacterial sera has been tried in man both to cure existing infections and as a preventive of infection. The therapeutic use has in most instances been rather disappointing, though in dysentery, plague, gonococcus and meningococcus infections the results have been somewhat better. Recently also, fairly good reports are heard from the administration of large doses of antistreptococcus serum. In susceptible animals injections of some of the very virulent bacteria, as pneumococci, streptococci, typhoid bacilli and cholera spirilla, can be robbed of all danger if small doses of their respective sera are given before the bacteria have increased to any great extent in the body. If given later they are ineffective. Conditions in man are probably very similar.

The reasons for the failure of these antibacterial sera when used therapeutically demand a moment's consideration. It is apparent from all that has gone before that a deeper insight into the mechanism of immunity discloses many difficulties to be overcome before we can hope for much in a practical way. In the case of the bacteria sera, for example, we have as yet found no method of increasing the complements, and these are apparently highly important in destroying the invading bacteria. Nor have we any way to determine the proper dose so as to avoid the phenomenon termed "deflection of complement." Possibly, also, as Ehrlich has sug-

gested, the complements present in human serum may not be able to reactivate immune bodies derived from the horse, sheep, or other animal furnishing the therapeutic serum. Probably the most important cause of the failure of these sera is that they do not reach the bacteria in the body. In the case of cholera, for example, it is hardly to be expected that the serum injected would affect the spirilla, for most of these are in the intestinal contents, and therefore really outside of the body. In many of the bacterial infections the organisms accumulate in the lymph glands and other sites where they cannot readily be reached by antibodies circulating in the blood. Instructive in this connection are the good results achieved by intraspinal injections of antimeningococcus serum, when exactly the same serum had proven valueless when given subcutaneously.

PRECIPITINS

Definition. — All of the foregoing experiments have concerned themselves with the results obtained by injection of cellular material of one animal into another. In the further study of this subject, experiments were made to discover what happens when *dissolved* albuminous bodies of one species are injected into animals of another species. This line of investigation was first pursued by Tchistowitsch,¹ who injected rabbits with the serum of horses and of eels. On withdrawing serum from such rabbits and mixing it with horse or eel serum, the mixture became cloudy, owing to the precipitation of part of the albumin of the horse or eel serum by that of the rabbit. Normal rabbit serum does not possess this property. Bordet was able to demonstrate that the same thing takes place if rabbits are treated with chicken blood. On mixing such a serum with chicken serum, a precipitate formed. The substances which develop in the serum by treating an animal with albuminous bodies of another animal, and which precipitate these albumins when the sera of the two animals are mixed, are

¹ Tchistowitsch, *Annal. Pasteur*. Vol. xiii, 1899.

called *precipitins*.¹ This power of the organism to react to the injection of foreign dissolved albuminous substances has been found to be very extensive.

Bacterial Precipitins. — In 1897, R. Kraus showed that the serum of a rabbit immunized against typhoid often produces a precipitate in the bacterial-free filtrate of a bouillon culture of typhoid bacilli. This fact has been verified by subsequent investigators and the reaction found to be specific. In general, the best results are obtained with old bouillon cultures which contain a larger proportion of the autolytic products. It was natural that this reaction should at once be applied to the diagnosis of typhoid and other diseases. Numerous experiments however have shown that Kraus' phenomenon is not nearly so constantly observed as that of agglutination, and the reaction is therefore but little used. Whether the bacterial precipitins are identical in character with those obtained by injecting an animal with an unrelated serum (zoöprecipitins), is still undecided. Rostoski, as well as Nuttall, believes that they are probably different. So much for bacterial precipitins.

Lactoserum — Other Specific Precipitins. — Bordet, by injecting cows' milk into rabbits', was able to produce a serum which precipitates the casein of

¹ It will be recalled that, besides the production of precipitins, the above procedure causes the formation of other anti-bodies such as anti-complements, anti-agglutinins, etc.

cows' milk. He called this *lactoserum*. Ehrlich, Morgenroth, Wassermann, Schütze, Myers, and Uhlenhuth showed that by treating a rabbit with chicken albumin a precipitin is formed which precipitates chicken albumin. Myers, by treating animals with Witte's pepton and globulin, produced a serum that contained specific anti-peptons and anti-globulins. Pick and Spiro, by using albumose, produced antialbumoses. Leclainche and Vallée, Stern, Mertens, and Zülzer treated animals with human albuminous urine and produced a serum which contained a precipitin specific for this substance. Schütze, by treating rabbits with a vegetable albumin, as well as with human myoalbumin, produced a precipitin specific for these albumins. This does not exhaust the recital of the work done in this field, and there is a host of other albuminous bodies which, when injected into an animal, are able to excite the production of precipitins.

Specificity of the Precipitins. — It was soon recognized that the specificity is not absolute. Above all, this depends upon the strength of the serum, i.e., its degree of activity. This is measured by the dilution in which it will still react. Thus a highly active serum, one, for example, which will still give a distinct reaction when diluted 1 : 1000 or over, will produce a marked precipitate with the serum used to excite its production; whereas, in the serum of other animal species it will produce slighter pre-

cipitates, or only cloudings. A less highly active serum will likewise cause a marked precipitate in the homologous blood solution, and a slight precipitate, or only a clouding, at the most, in a closely related species. For example, the serum of a rabbit which has been treated with sheep blood produces a marked precipitate in a solution of sheep blood; a slight precipitate in a goat-blood solution; and a still fainter one in an ox-blood solution. In some instances the two latter will show only a clouding. If we employ a very weak serum, even the cloudings will be absent, and a precipitate is formed only in the sheep-blood solution. If human blood or blood serum has been injected, the clouding and precipitation will occur most readily (aside, of course, from human-blood solution) in that of apes. In the precipitin reaction, therefore, the relationship of the single animal species is an important factor. This peculiar behavior has first been thoroughly studied by Nuttall¹ who made observations on five hundred different animals. As a result of these we know that a weak human-blood antiserum, besides reacting on human blood, causes a clouding only in the blood of anthropoid apes (chimpanzee, gorilla, orang-outang); a stronger serum causes a clouding also in the blood of other monkeys; finally

¹ *British Medical Journal*, 1901, Vol. ii, and 1902, Vol. i. See also Nuttall, *Blood Immunity and Blood Relationship*, 1904. The Macmillan Co., N. Y.

a very highly active serum reacts with the blood of all the mammalia. In that case, of course, only a faint clouding is produced even after considerable time. Nuttall also obtained antisera, each of which was specific for one of the large animal classes (birds, reptiles, amphibia). Here, too, the same quantitative differences were noted.

Nature of the Precipitins. — The precipitins are fairly resistant bodies, whose power gradually declines at a temperature of 60°C ., but is not lost until 70°C . is reached. Once their action is lost, it cannot be restored by the addition of normal sera, showing according to Ehrlich, that the precipitins are receptors of the second order and are not amboceptors. The resulting precipitate is soluble in weak acids and alkalies. Peptic digestion destroys the substances which effect the precipitation. Leblanc found that the precipitins were precipitated from the serum in that fraction which Hofmeister calls the *pseudo globulins*. Eisenberg, on the other hand, in his experiments found them in the *eu-globulin* fraction. The latter result was also obtained by Obermayer and Pick in precipitins obtained from goats and rabbits. The discordant results are comprehensible in view of recent publications concerning the unreliability of ammonium sulphate fractionation of serum globulins. The nature of the resulting precipitate has also been studied by Leblanc. He finds that it is a combination of the

precipitated albumin with the antibody of the specific serum. In this combination the properties of the *pseudo* globulin predominate showing that it is the specific serum which furnishes the greater part of the precipitate. The presence of salts seems to be necessary for the precipitin reaction. A temperature of 37° C. hastens, while a low temperature markedly retards the reaction. In either case, the amount of precipitum is uninfluenced. The presence of even small quantities of acids or alkalies markedly reduces the amount of precipitum formed, but an increase of salt (NaCl) has little effect.

Practical Application. — These precipitins have very recently found a practical application. Fish, Ehrlich, Morgenroth, Wassermann, and Schütze investigated the specific action of lactoserum. They found that a serum derived by treating an animal with cows' milk contained a precipitin which reacted only on the casein of cows' milk, but not on that of human milk or goats' milk. The serum of an animal treated with human milk was specific for the casein of human milk, etc. Ehrlich, Morgenroth, and Wassermann also experimented with the serum resulting from treatment with chicken egg albumin, and found that this, while not strictly specific so far as closely related species are concerned, is yet so against other species. *The precipitins, therefore, react on closely related albumins, but are specific against those of unrelated species.*

The Wassermann-Uhlenhuth Blood Test. — As a result of these researches Wassermann proposed, at the Congress for Internal Medicine, 1900, to use these sera as a means of differentiating albumins, i.e., to distinguish the different albumins from one another, and particularly to distinguish those derived from man from those of other animals. This proposal thus to use the Tschistowitsch-Bordet precipitins had important practical and theoretical results. Uhlenhuth, Wassermann, Schütze, Stern, Dicudonné, and others showed that a serum could be produced from rabbits by injecting them with human serum, by means of which it is possible to tell positively whether a given old, dried blood stain is human blood or not.

Uhlenhuth¹ tested nineteen kinds of blood and only obtained a reaction with human blood upon adding antihuman serum to the series of dilutions. He, moreover, found that human blood which had been dried four weeks on a board could be readily distinguished by means of antihuman serum from the blood of the horse and ox. On the following day Wassermann² demonstrated experiments similar to Uhlenhuth's at the meeting of the Physiological Society, Berlin. Outside of human blood only that of a monkey gave the reaction with antihuman serum.

¹ Uhlenhuth, Deutsche med. Wochenschrift, 1901. xxvii.

² Wassermann A. and Schütze, Berliner klin. Wochenschr. 1901. No. xxviii.

The reliability of this reaction in medico-legal questions has been abundantly established. In the forensic blood diagnosis the subjects of the test are usually blood stains on clothing, and on wood and metal objects. After such a doubtful stain has been dissolved in physiological salt solution, one first proceeds to determine that it is really blood. For this purpose Teichmann's test (the production of hæmin crystals), the guaiac test, and the spectroscopic examination are undertaken. This is of considerable importance, for not merely blood but other albuminous solutions derived from the same animal react with an antiserum obtained by injecting an animal with blood or serum. Having found that the stain is that of blood, we next determine the special kind of blood.

Immunizing the Animals. — For the production of the antisera, we make use of rabbits. These can be injected either with sterile, freshly-defibrinated blood or with sterile serum, the latter being preferable for intravenous inoculation. It is well to begin with small doses and gradually increase; thus for intravenous inoculations the first injection should be about one c.c. and increased up to three or four c.c. With intraperitoneal injections about double these doses can be given. Ordinarily, the interval between injections is three or four days, and the entire duration of treatment from two weeks to a month. Long-continued treatment

leads to a disappearance of precipitins from the blood.

Collecting the Serum. — When the animals have received five to six injections, and some days have elapsed it is well to draw off samples of the blood and to test for precipitins. This is easily done by shaving the ear and cleansing the skin with alcohol and sterile water. An incision is then made into the marginal vein and a few drops of blood collected in a small test-tube. This is then set aside to allow the blood to coagulate. After the serum has separated it can be tested and if it prove insufficiently powerful, treatment may be continued, otherwise the animal may be killed, preferably a week or ten days after the last injection. The animals may be killed in a variety of ways. Uhlenhuth chloroforms them, opens the thoracic cavity under aseptic precautions, and, cutting through the beating heart, the blood is allowed to flow into the thoracic cavity, whence it is removed by means of sterile pipettes to suitable vessels. Nuttall's method is to shave the neck and disinfect the skin with lysol solution; bend the animal's head backward to put the skin of the neck on the stretch, and have an assistant make a clean sweep with a sterilized knife through the tense skin to and through the vessels. The blood spurts into a large sterile dish which is immediately covered when the main flow has ceased. The dishes are placed horizontally until a clot has

formed; they are then slightly tilted, and as soon as serum enough has been expressed, this is pipetted off into sterile test containers which are stored in a cool place. It is well not to add any preservative to the serum, as such an addition may occasionally lead to *pseudo* reactions.

The Test. — In carrying out the test the suspected clot is mixed with a small quantity of normal salt solution and then filtered. Whether or not the blood specimen has gone into solution can best be judged by the *foam test*. Air is blown gently through the pipette which is used for transferring the solution into the test-tubes. Solutions of blood or serum of 1:1000 and over, still foam well. The color of the fluid is not so reliable an index of solution. To some of this solution in a test-tube, about double the amount of the specific serum (derived as above) is added. As a control test, we place a little blood of another species, e.g., of an ox, in a second test-tube together with some of the specific serum and a little normal salt solution. In a third tube we place some of the suspected blood solution, and in a fourth some of the specific serum mixed with the normal salt solution. All four tubes are placed in the incubator at 37° C. for one hour, or are left at room temperature for several hours. If the suspected clot was one of human blood, the first tube will show distinct evidence of precipitation, while all the control tubes will have remained clear. It

is desirable to dilute the suspected blood as far as possible when testing, for when concentrated sera are brought together reactions may occur which will lead to erroneous conclusions. In medico-legal work it will be well to progressively dilute a suspected blood sample and to reach a conclusion upon the highest (within limits) which reacts to a given antiserum. In routine work one can commence with dilutions of the suspected blood of 1:100 or 1:200. We must not omit to say that it is necessary to test to litmus all solutions to be examined, and to neutralize any that are found decidedly acid or alkaline.

Appearance of the Reaction. — When antiserum is added to blood dilution it sinks to the bottom of the tube, forming a milky white zone at the point of contact. The milkiness gradually extends upward until the whole fluid is clouded. Where the fluids have been mixed by shaking this diffuse cloudiness undergoes a change; after ten to twenty minutes, or later, very fine granules of precipitum begin to appear, and the upper layers of the fluid begin to clear, due to sedimentation of the precipitum. The fine particles soon become aggregated into coarser ones, and these into flocculi which, gradually sinking to the bottom of the tube, give rise to more or less deposit of a whitish appearance. With blood dilutions of, say 1:40 to 1:200 and over, the deposit formed is usually sharply defined; where more con-

centrated dilutions are used, the deposit may form an irregular mass at the bottom of the tube.

The reaction may be followed microscopically by means of the hanging-drop method. By this method a reaction can be observed within ten to fifteen minutes, which macroscopically becomes visible only after two hours.

Delicacy of the Precipitin Test. — Whereas the ordinary chemical tests for blood cease to give reactions in dilutions of about 1:1000, powerful antisera greatly exceed this limit, as the reported results of independent observers have shown. Working with an antihuman serum, Strube reports a reaction with a blood diluted 20,000 times, and Stern one with a blood diluted 50,000 times. Ascoli obtained a reaction with a specific serum with egg albumin diluted 1,000,000 times.

Other Applications of the Precipitin Test. — It can be readily understood that this test finds ready application in the detection of horse, dog, or cat meat in sausage.

The principle and the method are the same in all these various applications. We treat animals with the albumins which we wish to differentiate, and so obtain sera specific, each for its particular kind of albumin. These sera, then, produce precipitates only in solutions of their respective albumins. For example, if we wish to determine whether a given

sample of meat is horse-flesh or not we must inject an animal with horse serum, or, if we prefer, with an extract of horse-flesh. The serum derived from this animal will then produce a precipitate in the aqueous extract of the meat if this be horse-flesh, but not if it be beef. Animals treated with dog serum yield a serum which precipitates an aqueous extract of dog-flesh, etc. The method of examination consists in scraping the meat and extracting it with water or normal salt solution. It takes a long time to extract the meat in some cases. An extract is suitable for testing when it foams on being shaken. If the extract is very cloudy it should be cleared by filtration through a Berkefeld filter. In testing, add ten to fifteen drops of antiserum to 3 cc. of the saline meat extract.

Antiprecipitins — **Iso-precipitins**. — Biologically, the precipitins are found to behave like the substances already studied. It is possible, for example, by injecting an animal with a precipitin, say lactoserum, to obtain an *antiprecipitin*, an anti-lactoserum, which counteracts or inhibits the action of the precipitin. This is entirely analogous to the antihæmolysins, the antispermotoxin, etc.

If rabbits are treated with rabbit serum, a serum is obtained which will, in certain cases, precipitate the serum of other rabbits. This was done by

Schütze, and he called this serum *iso-precipitin*. Whether or not iso-precipitins ever occur in normal serum has not yet been definitely established. Their occurrence in human serum might be of importance in homologous transfusion.

II. CYTOTOXINS

Cytotoxins — Definition — Leucotoxin — Nature of the Cytotoxin — Anticytotoxin. — After it had been found that the injection of an animal with red blood cells of another animal was followed by the production of definite, specific reaction substances, investigators experimented to see whether this was also the case if other animal cells were used. Injections were made with white blood cells, spermatozoa of other animals, etc., and there resulted a series of reaction substances, entirely analogous to the hæmolysins, which were specific for the cells used for injection. These sera Metchnikoff calls *cytotoxins*. After Delezenne had published a short article on a serum hæmolytic for white blood cells, Metchnikoff undertook a study of the substances produced in sera of animals treated with leucocytes of another species. He injected guinea pigs with the mesenteric glands and bone marrow of a rabbit. He also injected for several weeks half an Aselli's pancreas at a time, at intervals of four days. If he withdrew serum from such a guinea pig he found this to be intensely solvent for white blood cells of a rabbit. He called this serum *leucotoxin*. This leucotoxin is very poisonous for these animals, and

kills them within a few hours. Non-fatal doses at first excite a marked hypoleucoeytosis, which is followed after a few days by a compensatory hyperleucoeytosis. Leucotoxin destroys the mononuclear as well as the polynuclear leucocytes of the animal, as was shown by Funk. Leucotoxin which had been derived by injection of the leucoeytes of horses, oxen, sheep, goats, or dogs acted only on the leucocytes of that species, not on the leucoeytes of man. So far as the mechanism of the cytotoxic action is concerned, it has been found that this is the same as that of the hæmolysins. The action of the specific cytotoxic serum is always due to the combined action of two substances in the serum, a specific immune body, and an alexin or complement present also in normal serum. The cytotoxic sera, like the hæmolytic sera, are rendered inactive by heating to 55° C. In other respects also the cytotoxic sera maintain the analogy to the hæmolytic sera. Thus it is possible by immunizing with a cytotoxin to obtain an *anticytotoxin*. Metchnikoff, for example, was able to produce an antileucotoxin by injecting animals with leucotoxin. This antibody inhibited the action of the leucotoxin.

Neurotoxin. — Delezenne and Madame Metchnikoff have injected animals with central-nervous-system substance, and so produced a specific *neurotoxin*. They injected ducks intraperitoneally, giving

them five or six injections of ten to twenty grammes of dog brain and spinal cord mixed with normal salt solution. The serum of these ducks injected intracerebrally into dogs in doses of 0.5 c.c. caused the dogs to die almost at once in complete paralysis, whereas if normal duck serum was injected in the same way no effects of any kind were produced. If smaller doses of the specific neurotoxic serum were administered, say 0.1 to 0.2 c.c., various paralyses and epileptiform convulsions set in, from which the animals sometimes recovered. The action of this serum is specific, i.e., the serum of ducks treated with dog brain causes these symptoms only in dogs, while on rabbits it acts no differently than normal duck serum.

Spermatoxin. — Another specific cell-dissolving serum was produced by Landsteiner, Metchnikoff, and Moxter, by injecting animals with the spermatozoa of other animals. Such a serum rapidly destroys the spermatozoa of the animals whose product was injected. This cytotoxin was named *spermatoxin*. If animals are treated with spermatozoa there is produced a serum which is not only a spermatoxin, but which is also hæmolytic for the red cells of that animal. This was demonstrated by Metchnikoff and Moxter, and has already been referred to in discussing hæmolysins. If, for example, we inject the spermatozoa of sheep into rabbits, we shall obtain a serum that is sperma-

toxic for sheep, as well as hæmolytic for sheep red cells.

Common Receptors. — At first it was thought that the hæmolysin so produced was due to the presence of small quantities of blood injected with the spermatozoa. The same result however was obtained when all traces of blood could be excluded;¹ furthermore a number of investigators produced hæmolysins by the injection of fluids entirely free from red corpuscles, such as serum and urine. The production of this hæmolysin is not hard to explain if we hold fast to the side-chain theory. We have merely to assume that the spermatozoa or these other substances possess certain receptors in common with the red blood cells of the same animal. Ehrlich and Morgenroth² have repeatedly pointed out that *specificity is a matter not of cells, but of receptors*. Despite these very conclusive demonstrations later investigators, who attempted to produce antisera for the cells of various organs, continued to use emulsions of unwashed organs, in utter disregard of the presence of free receptors in the organ juices and also without consideration of the antibodies certain to be produced by the red cells normally present.

Cytotoxin for Epithelium. — As far back as 1899,

¹ Von Dungern. See "Collected Studies on Immunity," Ehrlich-Bolduan, p. 47. Wiley & Sons, New York, 1910.

² Ehrlich and Morgenroth. *Ibid.*, p. 100.

von Dungern showed that it was possible to produce an *antiepithelial* serum by treating animals with the ciliated tracheal epithelium of oxen. This serum was rapidly destructive for this particular kind of epithelium, but it contained also a specific hæmolytic body just as was the case in the spermotoxic serum, and for the same reasons. This antiepithelial serum aroused considerable interest since it indicated the possibility of producing sera which were cytotoxic for certain varieties of epithelial cells, especially those of pathological origin, as carcinoma. The numerous experiments made in this direction failed however to produce the desired results. Owing to the extensive distribution of common receptors the antisera were found to exhibit quite general properties and to lack that degree of cell specificity, essential for practical purposes.

Cytotoxins by the Use of Nucleo-Proteids. — In order to prevent the adventitious formation of those bodies resulting from impure methods of immunization, and also in the hope of obtaining greater specificity, a few investigators have utilized the nucleo-proteids of the cell for immunization. This method seems to have been tried first by Marrassini in 1903, but with indifferent results. In 1905 Beebe¹ published an extensive study along

¹ S. P. Beebe, Cytotoxic Serum Produced by the Injection of Nucleo-Proteids. Journ. Exper. Medicine, Vol vii, 1905.

these lines and described the formation of a nephro-toxic serum which caused albuminuria and acute degeneration of the kidney without changes in the other organs. Albuminuria appeared generally on the fourth or fifth day, increased rapidly in amount, and was accompanied by the excretion of hyaline and granular casts. Subsequently Pearce and Jackson,¹ after a careful experimental study on the production of cytotoxic sera by the injection of nucleo-proteids, conclude "that the results do not support the theory that specific cytotoxic sera may be developed in this way, but indicate, rather, that such sera have certain mildly toxic properties acting in a general way and affecting especially the principal excretory organ, the kidney."

¹ R. M. Pearce and Holmes Jackson, *Journal of Infectious Diseases*, Vol. iii, 1906.

OPSONINS OR BACTERIOTROPIC SUBSTANCES

Historical. — The early work of Nuttall and others on the bactericidal action of normal serum, and Pfeiffer's demonstration of the bacteriolysis of cholera and typhoid bacilli by immune sera in the absence of cells, formed the chief basis on which rested the *humoral theory*, which attributed the protection in such cases to the destructive action of the serum on the microbes. It was found, however, that cases of protection resulting from the use of immune serum occurred where no such bacteriolytic action could be demonstrated; infection with plague or streptococcus may be mentioned as examples. It is now pretty generally accepted that immunity in these cases is due largely to the *phagocytic* action of the leucocytes. As far back as 1858 Haeckel had observed that particles of indigo injected into the veins of certain molluscs could shortly afterwards be found in the blood cells of the animal. However, the significance of this and other observations was not appreciated until Metchnikoff¹ in 1883 called attention to their bearing on infection and immunity. The outcome

¹ Arbeiten des Zoölog. Institutes in Wien, 1883, Vol. v.

of his investigations was the establishment of the well-known doctrine of *phagocytosis*, the principle of which is that the wandering cells of the animal organism, the leucocytes, possess the property of taking up, rendering inert, and digesting micro-organisms which they may encounter in the tissues. Metchnikoff believes that susceptibility to or immunity from infection is essentially a matter between the invading bacteria on the one hand and the leucocytes of the tissues on the other. He realizes that the serum constituents play an important rôle, but this rôle consists in their *stimulating the leucocyte* to take up the bacteria.

Thus if a highly virulent organism is injected into a susceptible animal, the leucocytes appear to be repelled, and to be unable to deal with the microbe, which multiplies and causes the death of the animal. If, however, the suitable immune serum is injected into the animal before inoculation, the phagocytes attack and devour the invading micro-organisms. Admitting that the phagocyte plays an important part in certain infections the question must still be considered whether the immune serum has acted on the injected microbes or on the phagocytes. Metchnikoff, we have seen, takes the latter view.

In 1903 A. E. Wright¹ called attention to certain substances present in serum which acted on bacteria

¹ Wright and Douglas, Proc. Royal Society, Vol. 72, 1903.

and rendered them more easily taken up by the phagocytic cells. He called this substance *opsonin* and showed that it is present in normal as well as immune sera. By means of absorption tests modelled after those of Ehrlich and Morgenroth, he showed that the opsonin has a specific affinity for the bacteria and none for the leucocytes. The opsonins for staphylococcus prepare only staphylococci for the leucocytes, those for tubercle bacilli only these bacteria, etc. As a result of his observations Wright supposes that the phagocytes play only a passive rôle, which depends on the preliminary action of the opsonin.

Bacteriotropic Substances. — Independently of Wright, though somewhat later, Neufeld and Rimpau¹ of Berlin published experiments on the phagocytic effect of immune sera. They also found that in these sera there exists a substance which has no direct action on the phagocytes, but which can fix itself on the corresponding bacteria and so modify these that they are more readily devoured by the phagocytes. They call this constituent a "bacteriotropic substance." There is little doubt that this bacteriotropic substance and Wright's opsonin are identical. Certain differences in the effect of heat are probably to be explained by the differences in the quantities of these sensitizing substances in normal and immune sera.

¹ Neufeld and Rimpau, Deutsche med. Wochenschrift, 1904.

Opsonins Distinct Antibodies. — It was natural to question whether these “ opsonins ” were really distinct from other antibodies, or whether they were perhaps identical with the immune body (or substance sensibilatrice). In a series of papers on this subject Hektoen¹ shows that the former is the case, opsonins are distinct substances. This is not only indicated by the results of absorption tests, but by the fact that, by immunization, a serum can in certain cases be obtained which is opsonic but not lytic, or in other cases one which is lytic but not opsonic. Similar experiments have differentiated opsonins from agglutinins.

Structure of Opsonins.—In structure the opsonins are like the agglutinins. Following Ehrlich's conceptions they possess two groups, opsoniferous and haptophore. On heating an opsonic serum the former group is destroyed, but the haptophore group remains intact, as can be seen from suitable combining experiments. There is still considerable difference of opinion as to the degree of heat necessary to inactivate the opsonins. Once the opsoniferous group has been destroyed it is impossible to restore the opsonic action by the addition of a complementing substance. Hence the opsonins are to be regarded as receptors of the second order and similar in structure to the agglutinins and precipitins. In this connection it will be well to

¹ Hektoen, L., *Journal Infect. Diseases*, 1905 and 1906.

remember Bordet's objections to the assumption of two groups in the agglutinin molecule. These have already been considered on page 40.

The Opsonic Index.—In the study of these opsonins Wright developed the idea that they were highly important in combating a number of bacterial infections, such as staphylococcus and tubercle. His observations showed that inoculations of the corresponding bacteria produced marked changes in the opsonic contents of the infected individual and that it was possible to estimate accurately the immunizing effect of such inoculations.

Technique. — Wright's technique of measuring the opsonic power is a slight modification of the Leishman¹ method and is as follows: An emulsion of fresh human leucocytes is made by dropping twenty drops of blood from a finger prick into 20 c.c. normal salt solution containing one per cent sodium citrate. The mixture is centrifuged, the supernatant clear fluid removed and the upper layers of the sedimented blood cells transferred by means of a fine pipette to 10 c.c. normal salt solution. After centrifuging this second mixture the supernatant fluid is pipetted off and the remaining suspension used for the opsonic tests. Such a "leucocyte emulsion," of course, contains an enormous number of red blood cells; the proportion of leucocytes, however, is greater than in the original blood.

¹ Leishman, British Medical Journal, Jan., 1902.

One volume of this emulsion is mixed with one volume of the bacterial suspension to be tested and with one volume of the serum. This is best accomplished by means of a pipette whose end has been drawn out into a capillary tube several inches in length. With a mark made about three-quarters of an inch from the end it is easy to suck up one such volume of each of the fluids, allowing a small air bubble to intervene between each volume. All three are now expelled on a slide and thoroughly mixed by drawing back and forth into the pipette. Then the mixture is sucked into the pipette, the end sealed and the whole put into the incubator at 37°C . The identical test is made using a normal serum in place of the serum to be tested. Both tubes are allowed to incubate fifteen minutes and then examined by means of smear preparations on slides spread and stained in the usual way. The degree of phagocytosis is then determined in each by counting a consecutive series of fifty leucocytes and finding the average number of bacteria ingested per leucocyte. This number for the serum to be tested is divided by the number obtained with the normal serum and the result regarded as the *opsonic index* of the serum in question. The presence of a high opsonic index Wright regards as indicative of increased resistance. He further states that the fluctuation of the opsonic index in normal healthy individuals is not more than from .8 to 1.2, and that

an index below .8 is therefore almost diagnostic of the presence of an infection with the organism tested.

Application of the Opsonic Measurements. — At the present time Wright has correlated all his observations and built up a system of treating bacterial infections by means of active immunization controlled by opsonic measurements. The principles underlying his method may be briefly summarized as follows: In localized bacterial infections the infected body absorbs but small amounts of bacterial substances or antigens. In consequence of this the amount of active immunity developed is but slight. Localized infections therefore tend to run a chronic course. The logical method of effecting a cure in these cases is to actively immunize the body with the invading organism. In a number of infections, notably those of staphylococcus, streptococcus, and tubercle, the degree of immunity is measured accurately by the opsonic index. Following an inoculation with the infecting bacteria (dead cultures in salt solution) there is first a drop in the opsonic index, the "negative phase," then, depending on the size of the dose and the reacting power of the individual, there comes a rise of the index, the "positive phase," or a continuation of the negative phase. The former is obtained with proper dosage; the latter with doses too large or too small. In estimating the size of the dose given, Wright counts the number of bacteria per c.c. of emulsion injected.

Thus in the case of localized staphylococcus infections the doses for adult humans range from 100 million to 500 million bacteria. In the case of streptococcus the doses are smaller, averaging about 50 to 100 million. The bacterial suspensions are heated to 60° C. for twenty minutes, 0.5% carbolic acid is added, and tests are made to insure sterility. The time for inoculation is governed by the opsonic index. If the first inoculation has been properly gauged there is a brief negative phase, followed by a positive phase of some days' duration. As this positive phase gradually drops, one gives another inoculation and watches the effect on the opsonic index. If the index drops markedly and rises but little, the dose has been too large. Or if the negative phase is slight, and the positive phase slight and transitory, the dose has been too small. With proper dosage the negative phases are small, and the opsonic index is kept fairly well above normal. Hand in hand with this goes an improvement in the clinical symptoms.

Wright and his pupils have published accounts of a large number of cases successfully treated according to this method. The results are reported as especially good in cases of severe acne, multiple boils, lupus, tubercular glands, and bone tuberculosis.

In judgingⁿ of the value of Wright's method we must bear clearly in mind that the essential feature of it is the *control by opsonic measurements*; treat-

ment of bacterial infections by the inoculation of dead cultures has long been known.

The results obtained by most workers in this country fail to bear out Wright's claims for the method. Thus the author¹ finds that the variation in the opsonic indices of several normal persons is often considerable; that opsonic counts based on fifty leucocytes may occasionally vary by more than 50% and that it is therefore necessary to count from 150 to 200 leucocytes for each test; that duplicate, triplicate and more tests made of the same serum, at the same time, and under identical conditions so far as one can tell, frequently give widely divergent results; that the opsonic index and the clinical course of the disease do not always run parallel. Cases may do very well and have the index remain low; other cases may do poorly with an increased opsonic index. It is to be noted, furthermore, that some of these variations in results are unavoidable, at least with the present technique.

To one who has followed the progress of immunity studies, it is not at all surprising to find that the opsonic index is not necessarily a measure of the patient's immunity. When Gruber and Durham published their observations on agglutinins the phenomenon was at once hailed and interpreted by many as measuring the degree of immunity possessed by the patient. The same error was made when

¹ Bolduan, Long Island Med. Journal, Vol. i, 1907.

some time later the bacteriolytic substances were discovered. In both cases it was soon found that these were but accompaniments of greater or less significance to the complex phenomenon of immunity. When we consider how manifold are the defensive agencies which the animal organism possesses, and how very complex they become the more they are studied, we shall not marvel at the absence of parallelism between the clinical course of the disease and the opsonic index. There is little doubt that the opsonic indices do measure a certain fraction or phase of the immunity reaction; we do not believe that they replace clinical observations in measuring the effect of immunizing injections.

VII. SNAKE VENOMS AND THEIR ANTISERA

Despite the fact that venomous serpents have excited the fear and interest of mankind for centuries it is only very recently that we have come to know anything definite about their poisons. This is perhaps in part due to the fact that Europe possesses but few poisonous snakes, and so offered little material for study. Some idea of the importance of the subject for certain countries, however, can be seen when it is stated that in India more than 20,000 persons annually die from the bite of the hooded cobra. It was quite natural, therefore, that one of the earliest modern researches into the nature of snake venom, that of Calmette,¹ should have come from that country. This author also found that he could produce an antitoxic serum by injecting animals with the snake venom.

The Venoms. — Our present knowledge of snake venoms and their antisera is due largely to the researches of Flexner and Noguchi² and of Kyes and Sachs.³ The venoms of different snakes vary

¹ Calmette, *Annal. Inst. Pasteur*, Vol. vi, 1892; *Comptes rend. Soc. Biol.*, 1894.

² Flexner and Noguchi, *Journal Exp. Medicine*, 1902, et seq.

³ Kyes and Sachs. See in *Collected Studies on Immunity*, Ehrlich-Bolduan, New York, 1910.

a great deal in their toxic properties, and this is due to their relative contents of different constituents, as follows:—hæmagglutinins, hæmolysin, hæmorrhagin, and neurotoxin. The first two act exclusively on the blood cells, the hæmorrhagin on the endothelium of the blood vessels, and the neurotoxin on the cells of the central nervous system. The last named causes death by paralysis of the cardiac and respiratory centers. The venoms of the cobra, water-moccasin, daboia and some poisonous sea snakes are essentially neurotoxic, although they have strong dissolving powers for the erythrocytes of some animals. In studying the hæmolytic powers of the venoms of cobra, copperhead, and rattlesnake, Flexner and Noguchi found cobra venom to be the most hæmolytic and that of rattlesnake the least. They attribute the toxicity of rattlesnake poison chiefly to the action of hæmorrhagin. The venoms of the water moccasin and the copperhead also contain hæmorrhagin.

Unlike the bacterial toxins the action of the snake venoms is preceded by no appreciable incubation period. In addition to this the poisons are very rapidly absorbed. Thus Calmette found that a rat inoculated into the tip of the tail could not be saved by amputating the tail one minute later. Such animals died within about five minutes of the time required for control animals.

The hæmolysin and neurotoxin and perhaps also

the other cytotoxic substances of venom consist of amboceptors which find a complement in the body of the poisoned animal. Not only does ordinary serum-complement serve for activation, but, according to Noguchi,¹ the fatty acids contained in the red blood cells also act as complement. Lecithin is also able to reactivate the hæmolysins of cobra venom, forming, according to Kyes, a "cobra-lecithid." Recent experiments by Manwaring,² however, show that the product obtained by Kyes was really a venom-free lecithin derivative and not a "lecithid."

Antivenins. — Calmette was the first to produce an antiserum against snake venom, utilizing for this purpose rabbits. He began with injections of $\frac{1}{20}$ of a fatal dose, and injected gradually increasing doses until at the end of four or five weeks the animals tolerated double a fatal dose. By continuing the treatment he finally got the animals to stand 80 fatal doses (40 mg.) without any reaction whatever. Five drops of the serum of such an animal neutralized 1 mg. cobra poison. It has been found that anticobra serum protects against the neurotoxic components of other snake venoms, furthermore against scorpion poison and the poison of eel blood. The serum also contains

¹ Noguchi, *Journ. Exper. Medicine*, Vol. ix, 1907.

² Manwaring, *Johns Hopkins Hospital Bulletin*, September, 1910,

an antihæmolysin, but no antibody against hæmorrhagin (of the rattlesnake). It is therefore without effect on rattlesnake venom. Antivenin for the latter may be prepared by immunizing goats with corresponding venoms which have been attenuated by weak acids. Such a serum, of course, possesses no antineurotoxin and is therefore useless against cobra and viper venoms.

VIII. ANAPHYLAXIS

Historical.—In 1898 Richet and Hericourt showed that when dogs were injected with eel serum they not only failed to develop an immunity against this poison, but actually became more susceptible. Subsequently they made similar observations with a toxin, mytilo-congestin, isolated from mussels. Richet applied the term “anaphylaxis” to this phenomenon to distinguish it from immunization or prophylaxis. Arthus, in 1903, reported that similar effects could be obtained with substances ordinarily not poisonous. Thus he found that if rabbits were injected with horse serum they were rendered very susceptible to a second injection made after an interval of six to eight days. The second injection produced severe symptoms, and sometimes led to death in these animals. Little or no attention was paid to these observations. Following a statement made to him by Theobald Smith in 1904, Ehrlich caused his pupil, Otto, to study why guinea pigs which had been injected with toxin-antitoxin mixtures in the course of standardization of diphtheria antitoxin, should so often be killed by a subsequent injection of horse serum. Independently of this the subject was being investi-

gated by Rosenau and Anderson in the Hygienic Laboratory. Almost simultaneously with the appearance of these studies came a comprehensive monograph on the serum rashes by v. Pirquet and Sehick, and this fitted in so well with the laboratory studies of Otto and of Rosenau and Anderson that a great deal of interest was aroused in this subject.

The Phenomenon.—As a result of all the work that has been done we now know that when an animal is injected with an alien proteid, there develops, after a time, a specific hypersusceptibility for this proteid. After a definite interval if the animal is given a second injection of the same proteid, violent symptoms appear, often leading to the death of the animal. The reaction is specific, so that animals sensitized, for example, to horse serum, manifest little or no hypersusceptibility to other sera. It is possible, however, to sensitize an animal to several proteids simultaneously. The sensitizing dose may be very small—even as little as one millionth cubic centimeter of horse serum has sufficed to render guinea pigs sensitive. A varying length of time must elapse after the sensitizing injection before the animal becomes fully sensitized. In guinea pigs injected with small doses of horse serum, from twelve to fourteen days suffice. With larger doses, however, the time required is much longer, and may extend over

weeks or even months. The hypersusceptibility is transmitted from mother to offspring, and may also be passively transferred to other animals by transferring some of the serum of the sensitized animal to normal animals. Animals recovering from the symptoms induced by the second injection are thereafter no longer hypersensitive to the proteid employed, but are immune. This immunity is spoken of as "antianaphylaxis." This condition can also be brought about artificially by injecting the animals after they have received their sensitizing injection and just before the end of the anaphylactic incubation time, with comparatively large quantities of the same proteid. Rosenau and Anderson have shown that animals may be sensitized by feeding them with the proteid. Whether this has any practical application to the clinical use of specific immune sera derived from horses in persons habitually eating horse flesh is not known.

Serum Rashes.—Turning our attention for a moment to the serum rashes, we find that in 1874 Dallera reported that urticarial eruptions might follow the transfusion of blood. Neudorfer as well as Landois also refer to this complication. In the year 1894 the use of diphtheria antitoxin introduced the widespread practice of injecting human beings with horse serum. In the same year several cases were reported in which these injections were fol-

lowed by various skin manifestations, mostly of an urticarial character. Following these came a great mass of evidence which made it clear that following the injection of antidiphtheric serum these sequelæ were usually comparatively harmless. Heubner in 1894 and von Bokay somewhat later expressed the opinion that these manifestations were due to other properties than the antitoxin in the serum, and this has proved to be the case. Johannessen produced the same effects by injecting normal horse serum. It has been shown that the skin eruptions and other symptoms follow in direct proportion to the amount of serum injected, a fact which has led to the concentration of the sera by getting rid of the non-antitoxic proteid constituents. In their exhaustive study, already mentioned, v. Pirquet and Schiek described the various clinical manifestations following the injection of horse serum into man, and gave the name "serum disease" to the symptom complex. The principal symptoms of this disease are a period of incubation varying in length from eight to thirteen days, fever, skin eruptions, swelling of the lymph glands, leucopenia joint symptoms, cedema, and albuminuria.

Theories of Anaphylaxis.—It was difficult to account for the long period of incubation in the production of these serum rashes. With poisons capable of self-multiplication (bacteria, etc.), this period was usually referred to the time necessary

for them to accumulate in sufficient number and virulence to produce symptoms. But serum is not a poison capable of multiplication. Pfeiffer's work on the endotoxins led to the view that the antibodies played an important part in bringing on the symptoms by setting free the endotoxins. The results of these observations are very closely related to von Pirquet and Schick's explanation of the production of serum disease. The endotoxin theory, in the sense of bacteriolysis, naturally cannot be applied to albuminous substances in solution. We can only accept it in the sense that by means of the reaction between the antibodies and the antigen the poisonous substance is formed. The period of incubation, both in serum rashes and in bacterial infections, is thus readily understood, for it is at once apparent that the formation of antibodies requires time. The general idea underlying von Pirquet and Schick's theory of serum disease is that the injection of the horse serum into man causes the development of specific reaction products which are able to act upon the antigens introduced. These antibodies encounter the antigens, i.e., some of the serum still present in the body, and so give rise to a poisonous substance. This accounts also for the cases of "immediate reaction" described by von Pirquet and Schick in which the second injection of a serum produces an attack of serum disease without any period of

incubation. Here the second injection comes at a time when the accumulation of antibodies is at its height. It has been claimed that this explains the cases of sudden death in humans following injections of serum, but investigation shows that most of these deaths occurred after but a single injection of serum. Moreover in most of them such conditions as status lymphaticus sufficed to explain the fatal ending.

This theory has found some experimental confirmation from the work of Vaughan and Wheeler, who have been able to prepare a number of split products from the proteid molecule, some of which in animals give rise to a symptom complex not unlike that of typical anaphylaxis.

Gay and Southard hold a somewhat different view. According to them the "horse serum contains a substance, anaphylactin, which is not neutralized, and is eliminated from the body with great slowness. When a normal guinea pig is injected with a small amount of horse serum, the greater part of its elements are rapidly eliminated; the anaphylactin, however, remains and acts as a constant irritant to the body cells, so that their avidity for the other assimilable elements of horse serum which have accompanied the anaphylactin becomes enormously increased. At the end of two weeks of constant stimulation on the part of the anaphylactin, and of increasing avidity on the part

of the somatic cells, a condition has arrived when the cells, if suddenly presented with a large amount of horse serum, are overwhelmed in the exercise of their assimilating functions, and functional equilibrium is so disturbed that local or general death may follow." According to this view the intoxication caused by the second injection depends upon constituents of the serum eliminable by the animal body.

Allergy.—It is apparent that what has been said concerning the production of anaphylaxis, in response to serum injections will apply also to bacterial infections, for in these the body is injected, as it were, with bacterial proteids. The phenomena of anaphylaxis are therefore of general application in immunity. This is well expressed by von Pirquet,¹ who calls attention to the fact that the main difference between a normal and an immune individual is one relating to the alteration in the latter's reactivity. He speaks of this alteration as "allergy": from *ergeia*, reactivity, and *allos*, altered, meaning thereby a changed reactivity as a clinical conception unprejudiced by bacteriological, pathological or biological findings. This alteration may relate to the quality and quantity of the symptoms and to their rate of development. Allergy seems to be associated more with some infections than with others. Experimentally it can best be studied by

¹ C. E. von Pirquet, Archives of Internal Medicine, Feb., 1911.

observing the effect of cow-pox inoculation in primary and subsequent vaccinations. The re-vaccinated overcomes the whole process with a very slight local reaction a few millimeters in size, while the person vaccinated the first time shows extensive local inflammation, fever, and other general symptoms. If the reaction is studied on the day following the vaccination, we shall find that the re-vaccinated is really hypersensitive, because at this time the first vaccinated does not show any reaction, while the revaccinated responds with a local inflammatory process. In tuberculosis, glanders, and other infections the injection of extracts of the infecting bacterium (tuberculin, mallein, etc.) produces characteristic local and general symptoms, because of the specific hypersensitive condition present in such infections. These reactions can therefore be employed in the diagnosis of such infections. The symptoms of hay fever, and of urticaria appear to be merely examples of proteid hypersensitiveness.

Supposed Relation to Precipitin Action. — Attempts have also been made to associate the phenomena of anaphylaxis with the action of precipitins. Hamburger and Moro were the first (1903) who found that man forms precipitins after the injection of horse serum. Precipitin was present after the appearance of serum rashes; therefore they suggested a connection between serum exanthem

and precipitin formation, without looking on the precipitation itself as the cause of the rash. More recently Doerr and Russ, as the result of experiments, hold that the phenomena of anaphylaxis are due to a reaction between precipitins attached to the tissue cells, and the precipitable antigen. The anaphylactic shock is looked upon as an intracellular precipitin reaction. In quantitative investigations these authors showed that the amount of anaphylactic antibody in the serum of rabbits was always parallel to its precipitin content. It has also been found that animals which do not form precipitins, like white mice, are incapable also of forming the anaphylactic antibody. Against the view that precipitins have anything to do with anaphylaxis in man is the fact that the symptoms of serum disease appear within eight to thirteen days following the first injection of horse serum, whereas it requires about three weeks for precipitins to appear in the blood in children after the injection of horse serum. Furthermore, the formation of precipitins does not take place as readily in man following the injection of horse serum as it does in rabbits. In fact von Pirquet found that sometimes even after the injection of 200 cc. there was no production of precipitins. Finally it may be remembered that there is no evidence that the precipitin action is other than a test-tube phenomenon, or that it ever occurs *in vivo*. Friedemann

has shown that the precipitates produced in vitro will, when injected intravenously into animals, pass through the capillaries without harmful effects.

Pathology of Anaphylactic Shock. — Acute anaphylactic death in guinea-pigs was originally attributed to asphyxia of central origin. Auer and Lewis,¹ however, showed that the asphyxia is due to a tetanic contraction of the bronchial muscle, the contraction being so pronounced that the lumina of the smaller bronchi are occluded, thus preventing both the entrance and the escape of air. In a recent study of the subject, Schultz and Jordan² show that in guinea-pigs the point of occlusion is usually just beyond the place where the secondary bronchi leave the primary, and in all cases at points commanding large areas of lung tissue. At this point there is the greatest relative (to diameter of lumen) amount of smooth muscle, and there is also normally a thicker mucosa and greater degree of folding of the same relative to the lumen. The fatal asphyxia observed in guinea-pigs is therefore due to the peculiar anatomical condition of the bronchial tree in these animals. In white mice the anaphylactic reaction shows itself by increased peristalsis, contractions of the bladder, increased irritability of the skin, etc. The respiratory symp-

¹ Auer and Lewis, *Journal Exp. Medicine*, Vol. xli, 1910.

² Schultz and Jordan, *Journal Pharmacol. and Exp. Therapeutics*, Vol. ii, March, 1911.

toms are absent. This is clearly because the mucosa of the bronchial tree is nowhere sufficiently thick or folded, relative to the amount of muscle and to the diameter of lumen, to produce occlusion under the amount of constriction produced by the contracting musculature. The recent work of Schultz shows that serum anaphylaxis is essentially a hypersensitization of smooth muscle generally.

It is possible that the occasional occurrence of severe symptoms and even of death in man following the injection of serum is sometimes due to an abnormal development or condition of the mucous membrane and smooth muscle of the bronchi. Some support is given to this view by the more frequent occurrence of these disturbances in asthmatic individuals.

Relation of Anaphylaxis to Serum Therapy.—

Returning now to the relation of the experimental work in anaphylaxis to serum therapy, attention should be called to the work of Steinhardt and Banzhaf, who show that the anaphylactic reaction in rabbits differs considerably in character from that observed in guinea-pigs. These authors, therefore, warn against utilizing the results of experiments on guinea-pigs without reservation for the interpretation of phenomena observed in human beings. It is probable that man cannot be sensitized in the same way as guinea pigs, the most susceptible of the laboratory animals. Children have in numerous

instances been injected with antidiphtheric horse serum at short and long intervals, without, so far as we are aware, causing death. Certain serums, for example, the antitubercle serum of Maragliano and the antirheumatic serum of Menzer, are habitually used by giving injections at intervals of days or weeks. It may, of course, be objected that possibly these injections are so spaced as to produce antianaphylaxis. If a person had once before had an injection of horse serum, would it be safe, say some months, or a year, or several years later, to give him another injection of horse serum? Or if a child had been immunized against diphtheria would it be safe to repeat the injection a year later if the child were again exposed? The experience of clinicians is practically unanimous in showing that such second injections need not be feared. Even if the results obtained in guinea pigs were applicable to man, a subcutaneous injection in man comparable to the amount required to produce sickness in a guinea pig would be over 200 cc. To date about twenty cases of sudden death following the injection of horse serum have been recorded in the literature, and while this undoubtedly does not represent all the cases that have occurred, the total number is insignificant when compared to the enormous number of such injections already made. In New York City, in over 50,000 persons injected, but two deaths attributed to the serum injection

have occurred. A number of fatal cases have been reported in asthmatic individuals, and this may be borne in mind when about to make serum injections. It is also of interest to know that Banzhaf and Famulener have shown that chloral in large doses will prevent the anaphylactic reaction in sensitized guinea pigs. Such animals after the second injection are immune to further injections.

IX. INFECTION AND IMMUNITY

Infection. In the preceding chapters we have studied the formation and mode of action of the various antibodies. Let us now summarize briefly our knowledge concerning the factors involved in infection and immunity. An infectious disease is one caused by a living organism which has gained access to the tissues of the body. A study of infection and immunity, therefore, embraces a study of the pathogenesis of these organisms on the one hand and of the defensive agencies of the body on the other. So far as the invading organism is concerned, we know that this may remain localized or be widespread through the body. The absorption of chemical products from a local infection may produce general symptoms. This is known as an intoxication, and is observed in cholera, diphtheria, tetanus, local abscess, etc. In general we apply the term 'pathogenic' to organisms capable of producing disease, but it must be borne in mind that this is a relative term, for an organism pathogenic for one species of animal need not necessarily be pathogenic for another species.

The Infecting Agent.—In studying the pathogenicity of various bacteria, it is apparent that

we can distinguish several classes of organisms. One class is characterized by the secretion of highly toxic soluble substances, both in the living body and in the culture fluid. The type of this class is the diphtheria bacillus. Another class produces highly toxic substances, which instead of being given off, remain within the body of the bacterium. These poisons may be demonstrated in old cultures in which a certain amount of dissolution ("autolysis") has taken place, or they may be obtained by mechanically breaking up the bacteria by pressure and grinding. These substances are spoken of as endotoxins, and are liberated in the body when the bacteria are disintegrated by the bacteriolytic agencies. The type of this class is the spirillum of cholera, an organism which produces a powerful endotoxin and which very readily undergoes bacteriolysis. In addition to these two classes we know of a large number of bacteria which neither secrete a highly toxic soluble substance as do diphtheria bacilli nor disintegrate as readily as the cholera spirilla, and which nevertheless are extremely pathogenic. Hiss has suggested that many organisms, if not all, secrete substances which are not soluble in their condition at secretion, but which are susceptible to digestion in the animal body. These substances thus become soluble and assimilable, and when toxic act harmfully on the body cells. Under ordinary circum-

stances these substances are broken up within the leucocytes and the poisons thus set free at once neutralized by neutralizing bodies present within the cells. According to this conception the leucocytes exercise a double function, one bactericidal and bacteriolytic, the other a poison-neutralizing one. The bactericidal and bacteriolytic bodies appear to escape from the leucocytes quite readily, and can be demonstrated in the blood plasma; the neutralizing bodies, on the other hand, do not appear to be given off from the cell. It is obvious, therefore, that the bacterial substances may be broken up in the blood plasma, and from them may thus be liberated a poisonous body. When this poisonous body is assimilated in sufficient quantity by the higher cells of the animal organism, death ensues, and ensues the more quickly the more rapid the process of liberation.

In discussing allergy it was pointed out that the phenomena of anaphylaxis should also be applied to bacterial infections, because in these the body was treated with small doses of bacterial proteid. As a result of his studies, Friedberger concludes that it is unnecessary to assume the existence of specific endotoxins in bacteria to account for the various symptoms seen in bacterial infections. By repeatedly injecting sensitized animals with minute doses of sheep or horse serum, he found it possible to produce all manner of fever curves at will, merely by varying

the size of the dose and the interval between injections. From this he concludes that the diversity of clinical symptoms of various infectious diseases can readily be explained on the assumption of but a single poison. He speaks of it as *anaphylatoxin*, and regards it as a cleavage product of proteid of whatever origin introduced parenterally. Just as in enteral digestion uniform cleavage products are formed from most diverse proteids, so he believes that in the parenteral proteid decomposition leading to the formation of anaphylatoxin, a uniform poison is produced. Whether or not in addition to the anaphylatoxin there are other specific poisons for the various infectious diseases is entirely immaterial; their existence has not been proved¹ and the assumption of their existence is unnecessary. In considering the diversity of the clinical symptoms of various infectious diseases, it must be remembered that the various species of bacteria differ in their virulence and in their rate of multiplication, and the invaded organisms also differ considerably in their antibody production. All these factors serve to modify the clinical picture. According to Friedberger the assumption of a common "anaphylatoxin" is only apparently in contradiction to the well-known law of specificity of the infectious diseases. In the infectious diseases it is not the poison

¹ This applies only to the infectious bacteria, not to those producing extra-cellular toxins.

which is specific, but only the mode of its production. The production of anaphylatoxin requires the action of antibodies; the mere solution or disintegration of bacteria by other means does not suffice. In other words, a definite cleavage of the proteid molecule is necessary. The anaphylatoxin, therefore, is not identical with Pfeiffer's "endotoxins," though perhaps the latter may be the mother substance from which the anaphylatoxin is derived.

Another important factor in pathogenesis, according to Bail, is the ability of many bacteria to produce certain neutralizing substances, not directly injurious, but able to inhibit or neutralize the antibacterial activities of the body. These substances Bail calls aggressins. There is still some doubt whether they are a distinct class of bacterial products. Wassermann and Citron, Doerr, and others regard them as consisting of dissolved bacterial substances, extracted endotoxins and toxins.

Resistance Against Infection.—The ability of an animal to resist the effects of a pathogenic organism is spoken of as *immunity*, and may be either *natural* or *acquired*. For example, it is well known that the lower animals are immune against syphilis and gonorrhœa, that dogs and goats are rarely affected with tuberculosis, and that man is naturally immune against chicken cholera and rinderpest. These are instances of natural im-

munity. Furthermore, it is well established that with certain diseases one attack usually protects the individual for life. This is well seen in small-pox, scarlet fever, and measles. Inasmuch as the individual was previously susceptible, this form of immunity is spoken of as acquired immunity.

Natural Immunity.—It is seldom that natural resistance is absolute. Young animals are often susceptible to an infection against which adults are resistant. Thus young pigeons are readily infected with anthrax while older pigeons are usually refractory. Moreover, the resistance of animals toward infections against which they are relatively immune can often be lowered by artificial means. Frogs can be infected with anthrax if they are kept in water at a temperature of 35° C. Conversely, chickens, which also are relatively immune to anthrax, can be infected if they are chilled. White rats, which are ordinarily resistant to anthrax infection, become susceptible after fatigue or when fed on an exclusively vegetable diet.

Ehrlich believes that natural immunity is sometimes due to the absence, in the body of the invaded animal, of suitable receptors for the virus. After what has been said in connection with the side chain theory, it is obvious that the virus cannot exert its pathogenic action if there are no receptors whereby it is anchored to the body cells. More-

over, as Ehrlich points out, even if such receptors are present, it is possible for the animal to be immune provided the receptors are situated only in indifferent, vitally unimportant tissues. In some of the lower animals there is reason to believe that the toxin of tetanus does combine with such tissue (Metchnikoff).

Acquired Immunity.—This may be either active or passive, and either form may be acquired naturally or artificially. As examples of naturally acquired active immunity we may mention the immunity developed by one attack of small-pox, scarlet fever, etc. The immunity against small-pox conferred by vaccination is an example of artificially acquired active immunity; so is the preventive inoculation with bacterial vaccine against typhoid fever. The best illustration of artificially acquired passive immunity is the injection of diphtheria antitoxin into humans, while the transmission of antitoxic immunity from mother to offspring is an example of naturally acquired passive immunity.

So far as maternal transmission of immunity is concerned, a number of writers, among whom may be mentioned Ehrlich,¹ Anderson,² and Theobald

¹ See Morgenroth's article in Kolle and Wasserman's Handbuch, Vol. iv, p. 784.

² Anderson, Bull. Hyg. Lab. U. S. Pub. Health and Mar. Hosp. Serv., No. 30.

Smith,¹ noted that an actively immunized female parent may transmit antibodies to the immediate young, who, receiving the immunity passively, soon lose it again. The male parent is unable to transmit any immunity. In his classical studies with ricin and abrin, Ehrlich showed that lactation played an important part in the transmission of immunity from female mice to their immediate offspring. By immunizing a nursing mother mouse (after the birth of the litter) he was able to demonstrate the transmission of immunity to swine plague to the nursing young. Smith, on the other hand, in his experiments with guinea pigs, immunized against diphtheria toxin, found that lactation played no appreciable part in the passive immunity of the young. Salge nursed infants with the milk of goats which had been immunized against diphtheria and against typhoid, and was unable to demonstrate the passage of antibodies to the infants.

In contrasting active with passive immunization we may say that the former is usually more effective, more lasting, and productive of a general immunity and not merely of one particular kind. It is, however, sometimes difficult to carry out, may involve some risk to the patient, and takes time. Passive immunization, on the other hand, is usually productive of only a limited kind of immunity, i.e., antitoxic, bactericidal, opsonic, etc.,

¹ Smith, *Jour. Exper. Med.*, Vol. xi, 1909.

and therefore is often ineffective. Consisting, as it usually does, in the injection of an alien serum, passive immunization produces an immunity of but short duration, the body apparently getting rid of the alien proteid as rapidly as possible. The great advantage of this form of immunization, however, is its convenience, freedom from risk to the patient, and above all, the fact that the immunity is produced instantaneously.

Mechanism of Immunity.—Infection, whether natural or artificial, is usually followed by a remarkable series of alterations in the tissues of the infected host. Representing, as it does, all the tissues of the body, it is natural that these changes are most strikingly exhibited in the blood. The alterations vary, however, both with the kind of bacterium, and with the animal species involved. Against the true toxins, including probably the leucocidins and hæmolysins, the body produces anti-toxins; against the bacterial bodies it directs the action of the leucocytes and the lytic combinations formed by the union of amboceptor and complement; against the so-called aggressins it directs the opsonins and perhaps also the bacteriolysins.

Before leaving the consideration of the reaction of the body to infection, attention should be called to the comprehensive investigations of Opie. This observer showed that the cells which accumulate in response to an irritant contain enzymes, the enzyme of the polynuclear leucocytes resembling trypsin

and the enzyme of the macrophages resembling pepsin in its action. The blood serum, on the other hand, contains an antienzyme. The varying relation existing between these enzymes and the antienzymes serves to explain how the same irritant in the same quantity may cause two different types of inflammation. This is well illustrated by the following experiment made by Opie:¹ If a small quantity of turpentine is injected into the subcutaneous tissue of dog, a large fluctuating abscess filled with creamy pus is formed within four days; there is a widespread undermining of the skin. The same quantity of turpentine injected into the pleural cavity causes a serofibrinous inflammation which undergoes resolution so that the pleural cavity is restored to its normal condition after about ten days; there is no destruction of tissue and a scar is not formed. In the subcutaneous tissue only a small amount of oedematous exudate can accumulate; the undiluted irritant causes active migration of leucocytes so that the antibody of the exuded serum is soon overbalanced by the enzyme set free by disintegrated pus cells. In the pleural cavity, on the contrary, a large quantity of serum quickly accumulates and the exudate is serofibrinous instead of purulent; the antienzyme it contains is capable of holding in check the enzyme

¹ E. L. Opie, Lecture before the Harvey Society, New York, Feb. 1910. The Harvey Lectures, J. B. Lippincott Co. 1910.

of the accumulated leucocytes. If a bit of the fibrinous exudate is suspended in the exuded serum, it is preserved intact. Nevertheless, by repeated injection of turpentine at short intervals into the pleural cavity, accumulation of leucocytes can be prolonged so that finally a condition is produced in which antienzyme can no longer restrain the enzyme. The softened fibrin of such an exudate quickly disintegrates in the serum of the exudate. These observations, as Opie points out, help to explain how the typhoid bacillus produces abscesses in certain situations such as the kidney and bone; how the pneumococcus, which rarely causes abscess of the lung, in which conditions are somewhat similar to those within the pleural cavity, may cause suppuration in other localities, such as the middle ear, or in the subdural space, etc.

In addition to the antibodies already mentioned, the animal body produces agglutinins and precipitins directed against the invading bacteria, but the relation of these antibodies to immunity is not at all clear. So far as the action of the agglutinins is concerned, we have already pointed out (on page 36) that this appears to have no destructive effect on the agglutinated organisms. Moreover, while agglutination is often observed to precede lysis, there is no reason to believe it a necessary factor in the lytic process, nor even an aid thereto. Whether this list exhausts the number of serum antibodies is doubtful.

The antibody content of the serum, however, is not always the same as that of the blood plasma. Thus Gengou, by collecting the plasma in vaselined tubes, found it often to be almost devoid of bactericidal power, while the corresponding serum was capable of destroying large numbers of microorganisms. In these cases, it is evident, we cannot regard plasma destruction of bacteria as the important factor in immunity. Moreover, in the case of bacteria containing considerable quantities of endotoxin, it is conceivable that plasma destruction of bacteria may even do considerable harm by causing an enormous liberation of endotoxin. This point is perhaps of practical importance as contra-indicating the use of bacteriolytic sera in the curative treatment of certain infections.

From what has been said it is evident that the exact mechanism of immunity, at least so far as most infections are concerned, is still very obscure. Like most biological phenomena, the deeper we analyze the problem, the more complex and more marvelous it becomes. Enough has, however, been presented to show some of the difficulties to be overcome and the method of attacking the subject.

Relation of Anaphylaxis to Immunity.—We have already discussed the relation of anaphylaxis to infection and may now take up briefly its relation to immunity. We know that the subcutaneous,

intraperitoneal, or intravenous introduction of alien proteid is followed by the formation of antibodies; at the same time it can readily be shown that no antibodies develop after the oral introduction of milk, eggs, or even of raw meat. In other words, there is a marked contrast in the behavior of the body between the *enteral* and the *parenteral* introduction of proteid. In the former the proteid is acted on by the gastric and intestinal juices (pepsin, trypsin, and enterokinase). These so break down the proteid molecule that it loses its species identity. After this, absorption takes place, and with it there is a synthesis, or rearrangement, of the molecule whereby it is built up into the specific proteid of the body. Under normal conditions it is impossible to produce specific antibodies by feeding alien proteid. Precipitins have, however, been produced by overfeeding animals with large quantities of alien blood. When proteid is introduced *parenterally* it gives rise to the formation of specific antibodies, and thus to the state of anaphylaxis. The term anaphylaxis is unfortunate, for the condition is not always opposed to immunity. Von Pirquet, it will be remembered, called attention to the altered reactivity during the anaphylactic state. We must not lose sight of the fact that the symptoms of anaphylaxis are brought on when sensitized animals are subsequently injected with relatively *large* quantities of the same proteid. Following

such an injection there is a sudden liberation of large amounts of toxic material. The parenteral introduction of large quantities of alien proteid must, however, be very exceptional under natural conditions. The number of bacteria primarily involved in an infection certainly represents but a very small amount of alien proteid. If the body is in the condition of allergy (anaphylaxis) at the time of infection it will be able to respond more quickly than otherwise and perhaps destroy the invaders. Under these circumstances it is conceivable that the condition is really an immunity reaction. Looking at the entire question broadly we may regard the mechanism which lies at the bottom of the phenomenon of anaphylaxis as a useful contrivance which enables the organism to rid itself of alien proteid, both organized and unorganized, which has been introduced parenterally.

Immunity Reaction on the Part of Bacteria.—It may be well at this point to call attention to a view advanced by Welch some years ago. According to this it is reasonable to suppose that just as the animal body produces antibodies against an invading organism, so does the latter, owing to the action of the body fluids, produce antibodies directed against the tissues of the invaded body. In this way the infecting organism would be adapting itself to unfavorable surroundings, and this we know it often does. It is certain that the animal body often

successfully overcomes an infectious disease without entirely overcoming the infecting bacteria. This is well shown by what we call chronic germ carriers. Deutsch regards the increase in virulence brought about by successive passage of a bacterium through a susceptible animal as representing an immunity developed by the bacterium against the anti-bacterial agencies of the body.

Atrepsy.—Ehrlich has investigated this phenomenon in the case of trypanosomes. He found that a monkey which had been infected with a particular strain of trypanosome and then cured by means of chemo-therapeutic agents, when tested with the original strain was not immune, the disease reappearing after a long incubation. If mice were inoculated with blood from the diseased animal, i.e., with blood containing trypanosomes, they became infected and died. Curiously, however, if the trypanosomes were first removed from this monkey blood, it was found that the serum was able to kill the *original* strain of trypanosomes. This showed that the trypanosomes had undergone some change in the body of the monkey; they differed from the original strain in their behavior toward the serum; they had become "serum-fast." Similar observations were made at the same time by Kleinc, and recently also by Mesnil.

In explanation of this adaptation, Ehrlich suggests that certain particular receptors of the para-

site are concerned entirely with the parasite's nutrition. Owing to the destruction brought about by the chemical agent, some of these receptors pass into the monkey's body, and, acting as antigens, excite the production of antibodies directed against these particular receptors. When living parasites are brought into contact with this antibody, either in vitro or in vivo, the antibody is anchored by the parasites. As a result of this occupation of its receptors, the parasite undergoes a biological alteration which consists in the disappearance of the original receptor group and its replacement by a new group. Ehrlich's researches lead him to believe that the antibody has merely an antinutritive action, blocking the nutritive receptor of the parasite and so bringing about starvation. The parasite thus develops immunity by getting rid of certain of its nutritive receptors, and replacing them with different ones. This form of immunity Ehrlich speaks of as "atresy," while the antibodies developed against the nutritive receptors he terms "atrep-sins." A somewhat different example of atresy is the following: Bird-pox, virulent for both fowl and pigeon, if passed through the pigeon becomes completely avirulent for the fowl. Ehrlich believes that the parasite in passing through the pigeon has to assimilate substances different from those assimilated in its passage through the fowl. Therefore that part of the receptors which deals with the

nutritive substances in the fowl's organism is not in use during the passage through the pigeon and may become atrophied, so that on the parasite being transferred back to the fowl, supposing one of the specific constituents of fowls to be necessary for its proliferation, it would no more be able to grow. We have, therefore, a loss of certain receptors which are absolutely necessary for nutrition.

Ehrlich suggests that probably the majority of so-called non-pathogenic micro-organisms, if introduced into an animal's body, perish by this mechanism. It is not necessary to assume the presence of special poisons in the body, it suffices to suppose that the bacteria in question do not find the needful means of existence in the body and therefore cannot multiply. They thus fall a prey to the phagocytes which destroy the invaders in a non-specific manner.

X. BACTERIAL VACCINES

Historical.—Early in the eighteenth century attention was called to the fact that in Oriental countries individuals were immunized against small-pox by inoculating them with a little small-pox virus under the skin. In 1796 Jenner showed that similar immunity could be produced by inoculating the virus of cow-pox, and this procedure was free from the dangers that attended small-pox inoculations. Following the discovery of the specific microbe of anthrax, attention was directed to the problem of combating this disease. Pasteur, who had been greatly impressed with Jenner's work with cow-pox, felt that attempts should be made to produce a mild attack of the disease, and that this would then protect against a virulent infection. After considerable experimental labor he devised the plan of inoculating animals with cultures of anthrax which had been attenuated by being grown at high temperatures, 43° C. These animals had a mild attack of the disease from which they soon recovered, and then were resistant to infection with virulent virus. Soon after this, inspired by Pas-

teur's work, successful vaccines¹ were prepared against chicken cholera and swine plague.

The discovery of diphtheria antitoxin in 1893 by v. Behring marked the beginning of the search for specific sera, and it was not long before a number of such were produced and employed clinically. The use of sera for therapeutic purposes was very attractive, because it was possible to have some animal, like the horse, manufacture the antibodies, and one needed then merely to transfer the animal's immunity to the patient by injecting some of the animal's serum. Clinical trials, however, soon showed that most of these sera had little therapeutic value, and subsequently laboratory experiments disclosed a large number of difficulties in their practical application. After what has been said under hæmolysins and bacteriolysins it will be unnecessary to dwell on these difficulties. Among them is the problem of providing sufficient complement, the determination of the optimum dose so as to avoid the paradoxical results known as the Neisser-Wechsberg phenomenon, the ability of producing really effective antibodies, and finally the question whether immunity in a given case is really directly due to the presence of these antibodies in the serum.

¹ The French have long used the term "vaccin" to denote any virus which is used for immunization, and that is the sense in which the term is used here. There is, of course, nothing of the cow, *vacca*, about them.

In the past few years it has become more and more apparent that the limitations of serum therapy, at least in the great majority of infectious diseases, are at present almost insuperable. Attention was therefore again turned to treatment by active immunization. It was perhaps only natural, in view of his discoveries in fermentation, that Pasteur should have believed that the production of immunity required the action of the living virus. He therefore vigorously combated the idea that immunity could be brought about by means of dead virus, or of lifeless products of growth of the virus. Touissant, as far back as 1880, had held out for the latter possibility, but the imperfections of his technique were such that his views were not accepted. To Salmon and Smith of this country belongs the honor of first clearly demonstrating the possibility of immunization with dead cultures.

Methods of Active Immunization.—Active immunization can be carried out in several ways:

(1) By means of living cultures of the virus. Usually the cultures are attenuated, but there are some exceptions.

A number of different procedures may be employed to attenuate the virus. Thus, by drying, as is done with rabies virus in the Pasteur treatment; or by growing the virus at a temperature unsuited for the development of virulence, as is done in the case of

anthrax; or by passing the virus through a less susceptible animal, as is done in vaccination against smallpox; or by means of chemicals such as the addition of iodine solution to diphtheria toxin, as was formerly done by Behring; or by means of heat, as was also formerly done with diphtheria toxin.

(2) By means of dead cultures of the virus. The cultures can be killed either by heat or by the use of chemicals.

(3) By the so-called "combined method," i.e., by first administering a dose of the specific immune serum and subsequently the virus. This method has been used in typhoid fever, cholera, and plague.

(4) By means of the products of autolysis of the cultures. This has also been used in typhoid fever, and seems to possess certain advantages over the use of native cultures.

(5) By means of various combinations of the preceding methods.

The choice of these various methods of immunization depends on the nature of the infecting virus. With some infections dead cultures apparently are able to cause the production of full protective powers, while in other infections the body seems to require a greater stimulus. In these, the use of attenuated living cultures may bring about the desired immunity. Finally there are infections in which nothing short of fully virulent cultures seems to bring about the development of sufficient

immunity. In these cases it is necessary to first prepare the way by the use of dead or of attenuated cultures.

Treatment with Vaccines.—The treatment of infections by means of active immunization has been greatly stimulated by the work of Wright, who has published favorable results in a large number of infections. Already several hundred thousand persons have been actively immunized against cholera, and large bodies of troops have been immunized against typhoid fever. Until recently the method found application particularly in the prophylactic immunization of persons liable to be exposed to infection. At the present time, however, owing largely to the efforts of Wright, the method has come to be used for curative purposes, i.e., for infections already in progress. This author has clearly formulated the conditions in which he thinks this form of treatment is indicated, and he has also devised methods for the more exact determination of doses than were formerly in use.

In the employment of bacterial vaccines, one must constantly keep in mind the nature of the bacterium with which one is working, and the kind of immunity one wishes to bring about. Everything depends on the way in which the vaccine is prepared. With bacteria making considerable quantities of a toxin, it will be necessary, if we

wish to immunize against this toxin, to grow the culture for the requisite length of time and under the proper conditions for producing the toxin. In the case of bacteria possessing certain endotoxins, it may be necessary to let the cultures autolyze, so as to set these substances free, or the bacteria may be crushed and ground for the same purpose. On the other hand, we may wish to use these bacteria for producing a specific agglutinating serum. In that case we often try to avoid injecting these toxic substances. Our entire procedure might then have to be quite the reverse of what has just been indicated.

The Vaccines.—Wright's method of preparing a staphylococcus, typhoid, streptococcus, or gonococcus vaccine, is as follows:

Several streak slant agar cultures are planted and incubated for twenty to twenty-four hours. The cultures are then washed off with normal salt solution, using from one to several cc. for each culture. These suspensions are next heated to 55° C. in order to kill the bacteria, and are then standardized. By this is meant determining the number of organisms per cc., for Wright always used definite numbers of bacteria in his inoculations. This standardization is readily accomplished by means of the method devised by Wright, which is as follows: From a finger prick a drop of blood is sucked up in a capillary tube to a mark made at any con-

venient point with a wax pencil. Next an equal amount of the bacterial suspension is drawn into the tube, allowing a tiny air-bubble to intervene. The two fluids are then expelled on a glass slide, and thoroughly mixed by sucking back and forth a number of times. After this has been done the mixture is spread in the ordinary way of making blood smears. If these blood smears, after staining, are examined with a microscope having a ruled eye-piece, it is a simple matter to determine the ratio of bacteria to blood cells. Taking the red blood cells as 5,000 million per cc., one calculates the number of bacteria per cc. In practice it is advisable to so dilute the bacterial suspension that the dose to be injected is contained in about one cc. of fluid. Finally $\frac{1}{2}$ per cent. of carbolic acid is added as a preservative. Such a suspension is a "bacterial vaccine." It goes without saying that the vaccines should be tested by means of cultures to insure sterility, and that contaminations should be excluded by means of microscopical examination.

Doses.—So far as doses are concerned, these vary with different bacteria, and also according to the indications, opsonic or clinical. The ordinary dose for the staphylococcus vaccine is from 200,000,000 to 1,000,000,000 organisms; for the streptococcus it is from 50 to 75 or 100,000,000, and for typhoid from 750,000,000 to 1,000,000,000 bacteria. All

the injections are given subcutaneously, and it is well to repeat the injections every three or four days.

Results.—The clinical results obtained by means of bacterial vaccines have varied. There seems considerable agreement on the part of most observers that certain localized infections, such as acne, multiple boils, etc., usually respond remarkably well with this method of treatment. In the treatment of bone tuberculosis the results are not so harmonious, and in the treatment of general infections many failures have been reported. There is no doubt, however, that treatment by means of bacterial vaccines is a valuable addition to our therapeutic armamentarium.

XI. LEUCOCYTE EXTRACTS IN THE TREATMENT OF INFECTIONS

Theory.—Attention has already been called to Hiss's view concerning the rôle of leucocytes in combating infections. Believing that the phagocytic power of leucocytes of persons suffering from infections to be less than that of leucocytes of the normal individual, Hiss was led to extract these cells with a view to utilizing their neutralizing and other protective substances in readily diffusible form. By this means it was thought possible to furnish to the infected organism such assistance as would enable its phagocytic cells to properly protect the various tissues from poisons elaborated from the invading bacteria.

Preparation of the Extracts.—In the preparation of the extract, double pleural inoculations of aleuronat ¹ suspensions are made into rabbits. After 24 hours the rabbits are killed and the turbid fluid collected from both pleural cavities. The quantity obtained varies from about 30 to 60 cc. The fluid is quickly centrifuged and the serum decanted. The cells are

¹ Aleuronat, a vegetable product similar to gluten, is prepared by Hundhausen, in Hamm, Westphalia, Germany, and is supplied in packages containing 100 grams. The suspensions are prepared with thin starch paste and boiled.

then thoroughly emulsified in distilled water, using about as much water as the volume of serum originally poured off, and the mixtures allowed to stand for a few hours at 37° C. This more or less autolyzed fluid is used for the injections, and the dose employed varies from 5 to 15 cc., repeated several times.

Application and Results.—In their work, clinical and experimental, Hiss and Zinsser¹ thought they saw little indication of immediate bactericidal power possessed by the leucocyte extract, but that the results pointed rather to a marked power on the part of the extract to reduce the purely toxæmic manifestations in infected subjects. Favorable clinical results have been reported by these authors in cerebrospinal meningitis, lobar pneumonia,² and other infections, and while the data are still too scanty to justify definite conclusions as to the value of this treatment, enough has been done to warrant further careful clinical investigations along this line.

¹ Hiss and Zinsser, *Journal Med. Research*, Vol. xix, Nov., 1908.

² See also Floyd and Lucas, *Journ. Med. Research*, Vol. xxi, Sept., 1909.

XII. THE PRINCIPLES UNDERLYING THE TREATMENT OF SYPHILIS WITH SALVARSAN ("606")

As a result of some of his earliest researches and in entire accordance with his views as already set forth under "Antitoxins," Ehrlich has always held that the action of a chemical substance on a given cell denotes the existence of definite chemical affinities between the substance and the cell. Applying this conception to the germicidal action of chemicals, he maintains that the latter must have a certain chemical affinity for the parasites in order to kill them. Substances having such affinities he terms "parasitotropic." It is clear, however, that substances which can destroy parasites will also be poisonous for the animal body, i.e., they will have chemical affinity for the vital organs of the host. They are, therefore, also "organotropic." In the employment of chemical substances in combating infectious diseases it follows that success can only be attained if their chemical affinity for the infecting parasite bears certain relations to their affinity for the infected body.

In his researches on trypanosomes Ehrlich found that if the dose of germicide administered to an infected animal was too small to kill all the parasites, there developed after a time a strain of organisms which were resistant to the further action of the germicide. It was usually futile to repeat the dose, for the resistant parasites would survive. At the same time the interesting observation was made that this resistance manifested itself only in the animal body; in vitro the parasites could still be killed by the germicide in question. It was also found that the various chemical substances which exerted therapeutic effects in animals infected with trypanosomes could be grouped into three classes, namely—1, various arsenicals (arsenious acid, atoxyl, arsacetin, arsenophenylglycin, and finally "606,"); 2, certain azo dyes (among them trypan red, trypan blue, and trypan violet); and 3, certain basic triphenylmethane dyes (among them parafuchsin, methyl violet, pyronine, etc.). Against each of these classes it was possible to produce specifically resistant strains of trypanosomes, so that a strain which had been made resistant to fuchsin was also resistant to related basic dyes, but vulnerable to the azo dye and to the arsenicals. Moreover, by appropriate treatment it was found possible to produce strains resistant against all three classes of trypanocidal agents.

According to Ehrlich the union of the chemical

with the parasites is brought about by certain "chemoreceptors" of the parasites. The arsenic compounds, for example, are anchored by arseno-receptors. When a strain therefore becomes resistant to the arsenicals one might suppose that this was due to the parasite ridding itself of its arseno-receptors; similarly also with the other classes of trypanocidal substances. But this appears not to be the case, for, as we have already said, the resistance is manifested only in the animal body, and not in vitro. Ehrlich explains this by assuming that in the resistant strain the affinity of that particular chemo-receptor has been reduced, so that when the germicidal agent is introduced into an animal infected with the resistant strain, the proportion of distribution of the germicide is altered in favor of the chemo-receptors of the organism. In other words the organotropic affinity is greater than the parasitotropic affinity. Had the parasite's chemo-receptors quite disappeared there should have been no difference in the resistance as manifested in the animal and in vitro.

In searching for germicidal substances whose parasitotropic affinity should be great in comparison to their organotropic affinity, Ehrlich made careful pharmacological studies with each of the three classes of trypanocidal substances already mentioned. After testing homologues and substitution products of almost every variety he finally concluded that

dioxydiamidoarsenobenzol fulfilled the requirements. As each of the substances was tested it received a laboratory number for identification; dioxydiamidoarsenobenzol bore the serial number 606, whence the designation by which this substance is still commonly known. Its trade name is "Salvarsan."

It is not our purpose, in these pages, to enter into the chemistry of "606," or to discuss the treatment of syphilis by this drug. Suffice it to say that Ehrlich lays considerable stress on the fact that in 606 the arsenic is in the trivalent unsaturated form. Pentavalent arsenic compounds, he believes, are less efficient in their trypanocidal action. As supplied in the market, Salvarsan is a bright yellow powder containing theoretically 34.15 per cent arsenic. It is the hydrochloride of dioxydiamidoarsenobenzol, and is administered by suspending or dissolving it in water with the addition of NaOH to neutralize, thus forming dioxydiamidoarsenobenzol plus NaCl and H₂O. Subcutaneous, intramuscular, and intravenous injections have been employed, as also solutions and suspensions of the drug. The results in the treatment of syphilis have been encouraging, but the time has not yet come to express a definite opinion concerning the ultimate value of this drug.

The principles here outlined, however, deserve to be carefully studied and tested experimentally,

for if correct they point the way for devising an effective therapy for many infections at present quite beyond our control.

LITERATURE

- EHRlich-BOLDUAN. Collected Studies in Immunity, (Chapter XXXIV) 1910. Wiley & Sons, New York.
- EHRlich. Experimental Researches on Specific Therapeutics. 1909. Hoeber, New York.
- EHRlich and HATA. Die Experimentelle Chemotherapie der Spirillosen. 1910. Springer, Berlin.
- MARTINDALE and WESTCOTT. "Salvarsan" ("606"). 1911. Hoeber, New York.
- WECHSELMANN. The Treatment of Syphilis with Dioxydiamidoarsenobenzol. (English translation by A. L. Wolbarst). 1911. Rebman Company, New York.

APPENDIX A

THE WASSERMANN TEST FOR SYPHILIS.

As has already been pointed out on page 72, Wassermann applied the principle of the Bordet-Gengou phenomenon to the detection of syphilis antibodies in the serum and cerebrospinal fluid of persons infected with syphilis. In the few years which have elapsed since Wassermann's first publication, the reliability of this method of diagnosing syphilis has been confirmed by a large number of investigators, and it has already proven of considerable value in several departments of medicine. In response to numerous requests, the writer has undertaken to give a clear description of the test, together with a brief review of the results thus far achieved by its use.

When an animal is repeatedly injected with red blood cells of another species, it reacts to such injections by producing substances in its serum which have the power to dissolve these foreign blood cells. Examined by means of a test tube experiment, it is found that the serum exerts this power only while it is fresh. Serum several days old is unable to dissolve the red cells. The fresh serum also

loses its solvent power by exposure to heat, say to 55° C. Investigations showed that the solvent action could be restored to these sera by the addition of small quantities of a fresh normal serum, i.e. of a serum which by itself had no solvent power whatever. The inactive serum had thus been reactivated. The original specific dissolving serum therefore contained two substances, one of which is very labile and the other stable. The stable substance is specific for the blood cells against which it is directed, i.e. against the cells used for immunizing the animal. It is called the "immune body," or the "amboceptor." The labile substance, as we have seen, is present in all *fresh* sera, and is spoken of as the "complement." The action of the immune body seems to consist in bringing the solvent action of the complement to bear on the given cells. We must conceive that the complement possesses the solvent power, but has no way of laying hold of the cell to be dissolved. The immune body merely effects this combination. Ehrlich's diagram on page 66 will serve to make this conception clear.

All that has been said regarding immune bodies and complement for the solution of blood cells, holds for the substances which effect destruction of bacteria when bacteria are used for immunization. In fact, the process is the same, no matter what cells are injected into the animal. The immune body is always directed specifically against the cells injected, and against no others.

As can be seen from Ehrlich's diagrams, the bacteria or blood cells combine directly only with the immune body. The complement, as already said, has no way of laying hold of the cells. As soon as the bacteria or cells have anchored the immune body, however, conditions change. The combination at once attracts and unites with the complement. If the amount of complement is not too large, the combination may unite with all of it, i.e. may abstract the complement from the serum.

Just let us examine this by means of an illustration: Let us suppose we have immunized an animal with typhoid bacilli, and have obtained a specific serum directed against these bacilli. This serum has been inactivated by heating it to 55°C. , so that now it will act on typhoid bacilli only when some fresh normal serum is added to complement the immune body. For this purpose we have provided ourselves with some freshly drawn serum from a guinea pig. The guinea pig serum, therefore, is the "complement." On mixing typhoid bacilli with the specific immune serum and then with the complement, these three factors enter into combination, and this results in the destruction of the typhoid bacilli. The quantities can easily be so arranged that this combination uses up all of the complement, so that the fluid contains not a trace of free complement after the substances have combined.

Suppose, now, that we also had a specific serum obtained by injecting an animal with red blood

cells, for example, by injecting a rabbit with sheep blood cells. This rabbit serum would then be specifically directed against sheep blood cells. Let us inactivate this serum, by heating it to 55°C. , so that now it requires the addition of a fresh normal serum to exert its solvent effect. For this purpose we can again use fresh, normal guinea-pig serum. When, then, we mix sheep blood cells with our specific immune serum (against sheep blood cells) and with the complement, i.e., with fresh normal guinea-pig serum, all three factors unite, and bring about destruction of the red blood cells. This is manifested by the blood cells dissolving and giving off their hæmoglobin to the rest of the fluid.

Let us now suppose we have carried out the first part of this experiment, that with the typhoid bacilli, and have left typhoid bacilli, specific typhoid serum and complement in contact for several hours in a warm place in order to cause the three factors to combine. At the end of this time let us add sheep blood cells and the specific serum directed against sheep cells, but let us add no further complement, because the fresh guinea-pig serum was able, as we saw, to serve as as complement also for the blood combination. The mixture is again placed in a warm place for several hours, and then for twenty-four hours in the refrigerator, after which it is examined. We shall find that no hæmolysis has occurred, from which we conclude that the previous combination (typhoid bacilli, immune serum and

complement), had used up all the complement, and left none for the blood combination.

If we were to repeat the whole experiment, but leave out, in the first part of the test, say the specific typhoid serum, we should find that the blood cells would be dissolved. This is readily understood when it is remembered that then we would have only typhoid bacilli and complement, two factors which cannot combine directly. The complement would therefore be left free to act in the blood combination.

If hæmolysis occurs we may therefore conclude that one of the factors in the first combination was absent, and conversely, if hæmolysis does not occur, we know that the first combination must have been perfect, i.e. all three factors must have been present.

It is at once apparent that in adapting this test to the detection of syphilis antibodies, pure cultures of the causative organism, i.e. of the "antigen," were not available. Wassermann therefore made use of extracts of syphilitic organs rich in spirochætes in place of the typhoid bacilli, and used either the serum or the cerebrospinal fluid of the suspected case in place of the typhoid antiserum. The rest of the test was similar to that described above. When hæmolysis of the sheep cells occurred, Wassermann said it showed that the first combination was incomplete; when hæmolysis was completely inhibited, it showed, he said, that the first combination was perfect, i.e. that the serum or

spinal fluid contained syphilis antibody. As in most such tests, only a positive result determines; a negative result does not necessarily exclude the presence of syphilitic infection.

While the above exposition will serve to fix the general plan of the test in the mind of the reader, we must at once say that the mode of action is not as simple as Wassermann first believed. Before going into this phase of the subject, it will be advisable to present a description of the technique of the test.

For carrying out the test the following materials are required:

(1) "Antigen," i.e. fluid containing syphilis material. This is comparable to the pure culture of typhoid in the test described above.

(2) Serum or cerebrospinal fluid from the patient to be examined.

(3) Sheep blood cells.

(4) Hæmolytic antibody, i.e. inactivated serum of a rabbit immunized against sheep blood cells.

(5) Complement, i.e. fresh normal serum from a guinea-pig.

For the syphilis antigen it is best to use the organs of a syphilitic foetus, i.e. one dead of hereditary syphilis, as these tissues are particularly rich in spirochætes. The organs are chopped up and macerated in a clean vessel in a mixture composed of water, 1000; NaCl, 8.5; carbolic acid, 5.0; one part of the tissue to four of the fluid. The mixture is shaken in a shaking apparatus for twenty hours;

the supernatant fluid poured off and centrifuged so as to be perfectly clear.*

The serum for the test is collected from the patient in the usual way by drawing from 5 to 10 cc. of blood from a vein at the elbow, placing the blood in a sterile test tube and allowing it to clot. Cerebrospinal fluid, obtained by lumbar puncture, is preserved with 0.5% carbolic acid, and then strongly centrifuged so as to make it perfectly clear.

The sheep blood cells are obtained by defibrinating sheep blood, centrifuging and washing the blood cells repeatedly with normal salt solution to remove traces of adherent serum. A 5% suspension in salt solution is used.

The hæmolytic antibody consists of the serum of a highly immunized (against sheep blood cells) rabbit, the serum being inactivated by heating to 56° C. In the tests cited by Wassermann, one cc. of a 1/1500 dilution of serum dissolved one cc. of 5% suspension of sheep blood cells at 37° C. in two hours.

The complement consists of freshly drawn guinea-pig serum. The test is carried out as follows:

To constant quantities of spinal fluid (e.g. 1 cc. of the 1/10 dilution) decreasing amounts of the extract of organs are added, thus 0.2, 0.1, 0.05 cc. Then 1 cc. of a 1/10 dilution of fresh normal guinea-

* In a very recent article, Wassermann states that more uniformly active extracts can be obtained by using 96% alcohol in place of the water in the above formula.

pig serum is added, and the mixtures allowed to remain in contact at 37° C. for one hour in order to bind the complement.

In this mixture we have *antigen*; we may or may not have *antibody*; we have *complement*.

If the antibody is present, the complement will be anchored by the combination, and so be unavailable for the hæmolytic test next in order. If no antibody is present, the complement will still be free to act in the hæmolytic test.

At the end of the hour, we add to the above mixtures: one cc. of a 5% suspension of sheep blood cells, and one cc. of the amboceptor dilution containing double the solvent dose for that amount of sheep blood cells. Thus, if the titer of the hæmolytic serum is 1/1800, we take one cc. of a dilution 1/900.

All the tubes are made up to the same volume with normal salt solution, namely, to 5 cc., and are then placed in the incubator at 37° C. and kept there for two hours. Then they are placed on ice until the next day, when the results are noted. The whole procedure is clearly shown by the protocol from Wassermann and Plaut reproduced on page 193.

Few experiments in immunity require such careful technique, or are open to so many sources of error as this serum test for syphilis. In view, too, of the enormous responsibility assumed in making a positive diagnosis of syphilis, it is apparent that the test should only be undertaken by

Syphilit Fœtus Extract. (0.2 gm.) One c.c. of $\frac{1}{2}$ Dilu- tion.	Spinal Fluid of Patient M. 0.2, i.e., One c.c. of the $\frac{1}{2}$ Dilu- tion.	Normal Guinea- pig Serum 0.1 cc., i.e. 1 cc. of a $\frac{1}{10}$ Dilution.	Hæmoly- tic Ambo- ceptor 1 cc. equals Double the Solvent dose for 1 cc. of a 5% Sus- pension.	Sheep Blood Cells 1 cc. of a 5% Susten- sion.	Results.
0.2	0.2	1.0	1.0	1.0	Complete inhibition of hæmolysis
0.1	0.2	1.0	1.0	1.0	Compl. inhibition
0.2	0.1	1.0	1.0	1.0	Marked inhibition
0.1	0.1	1.0	1.0	1.0	Marked inhibition
0.2	—	1.0	1.0	1.0	Complete solution
0.1	—	1.0	1.0	1.0	Complete solution
—	0.2	1.0	1.0	1.0	Complete solution
—	0.1	1.0	1.0	1.0	Complete solution
	Spinal fluid of Non-syphil. Person.				
0.2	0.2	1.0	1.0	1.0	Complete solution
0.1	0.2	1.0	1.0	1.0	Complete solution
0.2	0.1	1.0	1.0	1.0	Complete solution
0.1	0.1	1.0	1.0	1.0	Complete solution
—	0.2	1.0	1.0	1.0	Complete solution
—	0.1	1.0	1.0	1.0	Complete solution

highly trained laboratory workers. On the other hand, most who have busied themselves with the test agree that suitable controls always lead to a detection of possible sources of error, and that

therefore the reaction, when properly performed, can be relied upon.

When the test was first published Wassermann regarded the reaction which occurred as one between mutually specific bodies, i.e. between antigen and antibody, the resulting combination having the power to anchor the complement. Through the work of Marie and Levaditi, of Landsteiner, Müller and Pötzl, of Weil and Braun, and still other investigators, it soon became apparent that the test could also be carried out by using extracts of *non-syphilitic* tissue, i.e. of other pathological tissue or normal tissue. That, of course, meant that the view of a reciprocal specific relation between antibody and organ extract, in the sense that typhoid antibody and typhoid bacilli are reciprocally related, had to be abandoned. This does not, however, effect the reliability of the reaction for diagnostic purposes, for it has been found that positive results are still only obtained when the serum or spinal fluid is of syphilitic origin.*

Working under Wassermann's direction, Porges and Meier studied the nature of the substances concerned in the reaction, and began by precipitating the organ extracts with alcohol and testing

*It may be well to state that according to Landsteiner, Müller and Pötzl the serum of animals infected with dourine (trypanosomiasis) also gives rise to inhibition of hæmolysis when tested according to the above method. This has been confirmed by Hartoeh and Yakimoff. Whether this will affect the value of the Wassermann test in humans can only be decided by further clinical tests, especially in cases of human trypanosomiasis.

the resulting precipitate and clear fluid separately. It was found that the substance concerned in the reaction was soluble in alcohol, and the authors thereupon made alcoholic extracts of the syphilitic organs. These worked satisfactorily in making the test. It was natural to think that the substance which effected the reaction might be related to the lipoids, and so the authors next studied the behavior of alcoholic extracts of *normal* human and animal organs. While these extracts also sufficed to produce the reaction, the authors felt that they were not as active as extracts from syphilitic organs. After it had been found that alcoholic extracts could be used for the test, a number of authors almost simultaneously published favorable results with chemically defined substances. Porges and Meier used lecithin, Levaditi glycocholate of soda, Sachs and Altmann oleate of soda, and Fleischmann even used vaseline. The last-named also used cholesterin with favorable results, but Porges and Meier obtained only negative results with this substance.* On the whole, however, it seems that the extracts, especially of syphilitic organs, give the most uniform results.

At the present time, therefore, Wassermann believes that the really active principle in the antigen may be a combination of lipoids with certain protein-like substances, and that the latter component,

* See especially Noguchi, The Relation of Protein, Lipoids and Salts to the Wassermann Reaction, Journ Exp Medicine, vol. xi, 1909.

when it is derived from syphilitic material, has something of a specific character. In this connection Wassermann refers to the researches of Noguchi, Landsteiner, and others which show that minute quantities of proteid mixed with lipoids may cause extensive alterations in the physico-chemical behavior of the latter. He thinks that under certain circumstances this proteid component may play an important role in determining the reliability of the reaction, a view which is borne out by the investigations of Neisser and Bruck.

While Porges and Meier were engaged in the studies just mentioned, Fornet and also Michaelis showed that when the serum of individuals infected with syphilis was mixed with certain antigens a zone of precipitation might at times be observed at the point of contact of the two fluids. The antigen employed by Fornet was serum from individuals in the florid stage of syphilis; Michaelis used extracts of organs from a syphilitic foetus. This of course agrees with what was already known from the work of Bordet, Gengou, and Gay. In fact, according to Gay, the deflection or absorption of complement, on which the Bordet-Gengou test depends, may be due to the precipitate formed in the combination.

Porges and Meier thereupon tested the alcoholic extracts, and solutions of lecithin and of glycocholate of soda to see whether this zone of precipitation was at all constant, and whether it might not be possible to substitute such a simple precipitation test for the complicated Wassermann reaction.

While it was found that the test was roughly specific, it was soon realized that a precipitate might at times be produced with the serum of surely non-syphilitic individuals, and similar unfavorable results have since been published by other authors. At the present time, therefore, the only reliable serum diagnosis of syphilis is that based on the absorption of complement.

The results obtained with the Wassermann test are well reflected in the findings of Fleischmann, as follows:

The total number of persons tested was 230, of which 38 were controls. None of the latter gave a positive reaction. The other cases can be arranged into four groups thus:

(1) Cases surely syphilitic, with clinically manifest signs of syphilis at the time of the test. Of 89 such cases tested, 83 gave a positive reaction (93%).

(2) Cases surely syphilitic but without clinical symptoms at the time of the test. Of 64 such cases tested, 33 gave a positive reaction (52%), and 31 gave a negative reaction (48%).

(3) Cases with symptoms suggestive of syphilis, and with an indefinite history of infection. Of 32 such cases, 16 gave a positive reaction (50%), and the rest a negative reaction.

(4) Surely syphilitic individuals showing cutaneous lesions which the dermatologists diagnosed as very probably not syphilitic. Of 7 such cases, 1 gave a positive reaction and the rest a negative reaction.

Bruck and Stern tested 378 cases suspected to be

syphilitic, and obtained a positive reaction in 204. They also tested 157 surely non-syphilitic individuals as controls, and found all but two negative. These two gave a doubtful reaction.

In a recent paper Wassermann has collected data on about 3000 tests, as follows: There were 1010 tests on cases surely non-syphilitic (controls), and not one of these gave a positive reaction. Of the 1982 surely syphilitic cases tested, those examined at the time when they had manifest symptoms reacted in about 90% of the cases. When the cases tested were without manifest symptoms, so-called "latent syphilitics," about 50% reacted.

As a matter of interest it may be mentioned that Blumenthal and Wile tested the urine of syphilitic individuals, and found that this too would give the reaction.

Marie and Levaditi examined the *cerebrospinal fluid* of 30 cases of general paresis. All but two of the cases gave a positive reaction. When the *serum* was tested in place of the cerebrospinal fluid, the percentage of positive findings dropped to 59%.

Michaelis examined 20 cases of general paresis and obtained a positive reaction in 18 of them.

Citron examined 43 tabetics and paretics, and obtained a positive reaction in 34 cases (79%). He also tested the serum of 108 persons surely infected with syphilis, or suspected to be infected, and obtained a positive reaction in 80 (74%). None of the sera from 156 surely non-syphilitic individuals gave a positive reaction.

Favorable reports have also been published concerning the reliability of the test in ophthalmology, dermatology, and other departments of medicine.

From the researches of Noguchi and others it appears that with the progress of the cure of syphilitic infection not only do the symptoms of the disease abate, but the blood reaction also grows weaker and weaker, until ultimately, when cure has been established, the reaction can no longer be obtained. This is of clinical interest, for it enables one, by means of the serum test, to control the duration and efficacy of anti-syphilitic treatment. At the same time it must be remembered that the disappearance of the reaction need not necessarily be permanent. Its reappearance, of course, signifies that the virus is still present in the body.

APPENDIX B

NOGUCHI'S MODIFIED WASSERMANN REACTION.*

A MODIFICATION of the Wassermann test recently devised by Noguchi has been found both simple and reliable. In principle, the test is the same as Wassermann's, the only difference being in the hæmolytic system employed. By making use of an antihuman amboceptor, Noguchi avoids having constantly to obtain fresh sheep blood cells. In addition there is some gain in accuracy owing to the fact that human blood serum, in the quantities ordinarily employed in the Wassermann test, often contains normal amboceptors for sheep blood which interfere with the reaction.

The reagents employed by Noguchi are as follows:

1. Antihuman hæmolytic amboceptor. This consists of a serum obtained from rabbits immunized with washed human erythrocytes.
2. Complement. This consists of fresh guinea-pig serum.
3. "Antigen." Alcoholic extracts of organs, or crude preparations of lecithin. The preparation of the organ extracts has been described on page 152.

* Noguchi, A New and Simple Method for the Serum Diagnosis of Syphilis, Journ. Exp. Medicine, Vol. XI, 1909.

In using lecithin, 0.3 gms. lecithin are dissolved in 50 cc. absolute alcohol, shaken with 50 cc. physiological salt solution and filtered. The filtrate must be clear.

4. Suspension of human blood corpuscles. One drop of blood from a normal individual is mixed with 4 cc. physiological salt solution.

5. Serum to be tested. About 10 or 15 drops of blood are collected in a small tube and allowed to clot. The clear serum is used for the tests.

Method.—Take six clean test tubes, size 10 cm. by 1 cm. Into the first two of these place one drop from a capillary pipette of the patient's serum to be tested. Into each of the second two tubes (which serve as controls) put one drop of serum of a syphilitic case known to give a positive reaction. Into each of the third pair of tubes put one drop of serum of a normal person. Now to each of the six tubes add 1 cc. of the suspension of human blood corpuscles and 0.04 cc. fresh guinea-pig serum as complement. Lastly, into one tube of each pair put one drop of the "antigen" solution from a capillary pipette. The second tube of each pair receives no antigen.

After being well mixed by shaking, the six tubes are incubated at 37° C. for one hour, after which each tube receives two units* of antihuman amboceptor. The tubes are returned to the incubator for two

* The term "amboceptor unit" is used to designate the amount of amboceptor which, on the addition of the optimum amount of complement, just suffices to produce complete hemolysis.

hours longer, and then kept at room temperature, the reaction being read from time to time. The interpretation of the result is identical with that of the Wassermann reaction.

The following scheme accompanying Noguchi's article will serve to elucidate the technique.

	Test for Diagnosis.	First Control Set. Test with a known syphilitic serum. (Positive reaction.)	Second Control Set. Test with a normal serum. (Negative reaction.)
Control tube without "antigen" in each test	<i>a.</i> Patient's serum	<i>a'.</i> Posi- tive luetic serum	<i>a''.</i> Normal serum
	○ <i>b.</i> Human blood suspension	○ <i>b.</i> Human blood suspension	○ <i>b.</i> Human blood suspension
	<i>c.</i> Comple- ment	<i>c.</i> Comple- ment	<i>c.</i> Comple- ment
Determina- tive tube, contains "antigen"	<i>a.</i> } ○ <i>b.</i> } Ditto <i>c.</i> }	<i>a'.</i> } ○ <i>b.</i> } Ditto <i>c.</i> }	<i>a''.</i> } ○ <i>b.</i> } Ditto <i>c.</i> }
	<i>d.</i> Antigen	<i>d.</i> Antigen	<i>d.</i> Antigen
Incubation at 37° C. for one hour.			
Antihuman hæmolytic amboceptor to all tubes.			
Incubation at 37° C. for two hours longer, then at room temp.			

Noguchi has very cleverly adapted his method for clinical purposes by drying the various reagents on paper and standardizing them. He prepares three kinds of slips, as follows:

Antihuman amboceptor slips, each containing two amboceptor units.

Complement slips, each containing sufficient complement for one tube.

Antigen slips, each containing sufficient for one tube.

These papers can be kept indefinitely at room temperature in a dry place. Prior to using the slips for a blood test it is necessary to make preliminary tests to prove their activity and strength.

In employing the filter paper slips they are dropped by means of a forceps into the test tubes containing the human blood suspension and patient's serum in the order and at the intervals already stated for the respective reagents. If an incubator is not at hand, the tubes can be kept warm by carrying them in the vest pocket.

APPENDIX C

BLOOD EXAMINATION PREPARATORY TO TRANSFUSION

Reasons for Making the Examination.—We have already called attention to the occurrence of isoagglutinins, isohæmolysins, and isoprecipitins and their bearing on homologous transfusion. The mere occurrence of these substances in blood serum, to be sure, does not at all prove that isoagglutination or isohæmolysis or isoprecipitation occur when such transfusions are done. In fact we do not even know whether these substances exist at all in the blood plasma. Nevertheless, until we learn otherwise, it will be well to bear in mind the possible danger from this source, and to undertake no transfusions in which examination shows the existence of homologous antibodies.

Technique of the Tests.—It is evident that our tests must be reciprocal, i.e., we must test the serum of both donor and recipient against the blood corpuscles of the other. To do this we collect part of the blood from each individual, part in citrated salt solution and part in a plain test-

tube. The latter is allowed to clot and furnishes the serum; the former is prevented from clotting by the sodium citrate and serves to supply the blood corpuscles. Instead of using sodium citrate, Crile,¹ defibrinates the blood by shaking it in a test-tube with a glass bead, and suspending the blood corpuscles in physiological salt solution. Either method may be used, though with the sodium citrate it is necessary to centrifuge, wash the blood corpuscles, and then resuspend them in salt solution. The suspensions are usually 5% strength.

In carrying out the test equal parts of serum and blood suspension are mixed in a small test-tube, or, as Epstein² has suggested, in small pipettes such as Wright uses for his opsonic tests. After mixing, the tubes are placed in the thermostat for two hours. At the end of this time most of the cells have usually settled to the bottom and pronounced hæmolysis can be seen. For finer grades of hæmolysis it is usually necessary to allow the tubes to stand over night in the refrigerator.

Agglutination, when it occurs, is rather prompt, and can be readily observed in the gross by the clumping and sedimentation of the blood corpuscles.

¹ Crile, Hemorrhage and Transfusion, 1909, Appleton and Co., New York.

² Epstein and Ottenburg, Archives of Internal Medicine, Vol. iii, page 286, 1909.

It is important in the hæmolytic tests that all the glassware be absolutely clean and dry, though it need not be sterile.

In testing for the presence of isoprecipitins, equal parts of the two sera are mixed in a small test-tube, the mixtures kept in the incubator for two hours and then examined.

APPENDIX D

OTHER REACTIONS

The Conglutination Reaction

IN 1906 Bordet and Gay described the presence in bovine serum of a substance having the property of producing a characteristic clumping of red blood-cells and of accelerating their lysis provided the cells had first been treated with both sensitizer (amboceptor) and alexin (complement); its action was possible under no other circumstances. Subsequently it was shown that the same phenomenon could be produced with *bacteria* treated with a specific sensitizer and alexin. This substance is spoken of as "conglutinin." From the work of Bordet, Streng, Gay, and others it would appear that the conglutination reaction can be employed to discover the presence of specific sensitizers (amboceptors), and thus be applied to the diagnosis of bacterial infections. The experimental work thus far done on the subject is still too meagre to warrant any definite statements as to the diagnostic value of the reaction. In some infections studied by Gay the conglutination reaction appeared earlier than the agglutination reaction.

The Meiostagmin Reaction

Weichardt called attention to the fact that the union of antigen with its antibody in certain dilutions caused an increase in the rate of diffusion, i.e., gave rise to changes in the osmotic pressure and of the surface tension. Ascoli showed that the decrease in the surface tension arising when bacterial substances combined with their specific antigen could be measured by counting the number of drops per given time interval delivered from a Traube stalagmometer. Thus where a mixture of normal serum with extract of typhoid bacilli showed 56 drops, a similar mixture of serum from a typhoid fever patient with the extract showed 58 drops. Attempts have been made to utilize the meiostagmin reaction in the diagnosis of various infectious diseases, and while the results on the whole have shown the correctness of the underlying principles, they have also demonstrated that other reactions are far more convenient and decisive.

The Much-Holzmann Cobra Venom Reaction

It has long been known that cobra venom hæmolyzes red blood corpuscles, and that certain corpuscles, such as those of man, dog, pig, horse, rabbit, and guinea-pig hæmolyze directly on mixing them with cobra venom, while others require the

intervention of an activating substance. To the latter class belong the blood corpuscles of ox, sheep, and goat. As already pointed out in discussing snake venoms, the activating substance is present in blood serum; it is also present in commercial lecithin. Hæmolysis of either group of blood corpuscles can be inhibited by means of cholesterin, though just how this substance acts is not clear. Much and Holzmann showed that the blood serum of patients suffering from various mental disorders, especially dementia præcox, and manic-depressive insanity frequently inhibits hæmolysis of human red blood corpuscles, and they suggested that the reaction could be used for diagnostic purposes. While it appears to be true that the psychoses yield the largest proportion of positive reactions, the value of the reaction for diagnostic purposes is practically *nil*. At the same time it is interesting to note that diseases of the nervous system accompanied by demonstrable lesions of the nerve tissue give the same reaction as psychoses in which such lesions have not yet been demonstrated. Much¹ therefore concludes that in both cases the same substance circulates in the blood, and that, moreover, in both this substance is derived from a degeneration of the nerve tissue.

¹ Much, Die Immunitätswissenschaft. C. Kabitzsch, Wurzburg, 1911.

Weil's Cobra Venom Test in Syphilis

In studying the varying resistance of red blood corpuscles to hæmolytic agents, Weil¹ noted that the corpuscles of syphilitics were regularly more resistant to the action of cobra venom than those of normal individuals. Just what is the cause of this increased resistance is not entirely clear. It is known that syphilis attacks the lipoids of the body, and that the amount of lecithin which can be extracted from the tissues is less in syphilitic conditions than in normal individuals. The increased resistance has therefore been thought to be due to a decrease in the lecithin content of the red blood corpuscles.

In testing the corpuscles the blood is collected in citrated salt solution, and then washed by repeated centrifugalization in order to thoroughly remove the serum. Even slight traces of serum interfere decidedly with the reaction. The washed cells are made up into a four per cent suspension in 0.9 per cent common salt solution. Centrifugalization is done in very accurately graduated centrifuge tubes, as the accurate dilution of the cells is a matter of very great importance. The cobra venom, which is kept in the dried con-

¹ Weil, Richard, *Journal of Infectious Diseases*, Vol. vi, Nov., 1909. *Proceedings Society Exp. Biology and Medicine*, Vol. vi and Vol. vii.

dition, is made up very accurately in small quantities into a 0.05 per cent solution in 0.9 per cent solution of common salt. From this stock solution, higher dilutions are prepared as required, and are kept at a constant low temperature. In carrying out the test four solutions are prepared, namely, 1 in 10,000, 20,000, 30,000, and 40,000. One cc. of the corpuscle suspension is added to one cc. of the venom solution. The mixtures are incubated for one hour at 40° C., after which a preliminary inspection will almost invariably reveal the final results of the tests. At this stage, cells showing no hæmolysis with 1 to 10,000 venom are strongly positive, cells showing no hæmolysis with 1 to 20,000 are positive, cells showing very slight hæmolysis at 1 to 30,000 are weakly positive; cells showing moderate destruction at 1 to 30,000 are negative, and cells showing the least appreciable trace of hæmolysis at 1 to 40,000 are strongly negative. After the preliminary reading the mixtures are again shaken and then kept in the ice-box overnight, the final readings being made on the following morning. The figures here given are merely those of a particular specimen of cobra venom; it is possible that other samples might differ in strength. At all events a preliminary titration against corpuscles derived from cases of known syphilis and controls suffices to determine the values for any given sample. Inasmuch as the

strength of the dried venom does not vary, this preliminary titration gives a constant standard, which is practically permanent, since one gram of venom suffices for about 5000 complete tests. The standardization is different for infants as compared with adults, for the red corpuscles of infants and young children hæmolyze much more readily than those of adults. The cells, however, acquire the same degree of resistance as do those of adults in disease.

Weil states that the cobra venom reaction appears to have certain advantages over the Wassermann reaction. It is much simpler and the labor much less. Less blood is required, so that a few drops from the lobe of the ear suffice for a reaction; this is of some importance in infants. Cases of scarlet fever and of leprosy do not offer a source of confusion. The reaction is more marked in old, apparently dormant cases. The reaction persists much longer after mercurialization, thus offering a further diagnostic and therapeutic test. The reaction is possible in cases of jaundice, whereas the Wassermann is not.

Antitrypsin Determinations.

We have already pointed out that the animal body responds to the injection of ferments by the production of antiferments. Considerable in-

terest has been aroused by the discovery that in certain diseases, especially cancer, the antitrypsin content of the patient's serum is markedly increased. In cancer this increase is noted in about 90 per cent of the cases. The antiferment action is not entirely specific, but extends to other proteolytic ferments, and particularly to the ferment of leucocytes. At the present time, therefore, a marked increase in the antitryptic power of a patient's serum is taken to indicate an increased parenteral¹ destruction of proteid in the body.² The original method of demonstrating the presence of this antitrypsin was by placing drops of proteolytic ferment (trypsin) on the surface of a plate of Loeffler's serum, and causing the development of small concavities owing to the digestion of the medium. The addition of inhibiting serum to the drops was able to prevent the formation of the concavities. A more convenient and accurate method is the one developed by v. Bergmann and Meyer. This depends on the digestion of a perfectly clear solution of casein. If all the casein has been digested, the addition of acid is obviously unable to precipitate any casein from solution. On the other hand, if the acid causes clouding or precipitation, it follows that all the casein was not

¹ Other than intestinal.

² See the excellent digest of the work on this subject in *Jahresbericht der Immunitätsforschung*, Bd. V, 1909, Abteilung I, page 58.

digested. It is evident that the quantity of trypsin required to digest a given amount of casein can be exactly determined, and that by employing graduated amounts of the inhibiting serum accurate determinations of the antitryptic content can be made.

INDEX OF AUTHORS

- | | |
|--|---|
| <p>Abel, 17, 100
 Anderson, 142, 160
 Altmann, 195
 Aronson, 5
 Arrhenius, 28
 Arthus, 141
 Ascoli, 208
 Atkinson, 18
 Auer, 150</p> <p>Bail, 93, 158
 Banzhaf, 19, 151, 153
 Beebe, 125
 von Behring, 1, 5, 170
 Belfanti, 18
 von Bergmann, 013
 Besredka, 82, 84
 Blumenthal, 198
 von Bokay, 144
 Bolduan, 135
 Bordet, 28, 40, 52, 56, 67, 87,
 96, 108, 185, 207
 Praun, 194
 Bruck, 72, 196
 Buchner, 1, 6, 50, 75, 78, 93
 Buxton, 104</p> <p>Calmette, 18, 137
 Carbone, 18
 Castellani, 46</p> | <p>Citron, 158, 198
 Crile, 205</p> <p>Dallera, 143
 Delezenne, 121
 Deutsch, 166
 Dieudonné, 18, 113
 Doerr, 149, 158
 Donath, 94
 Douglas, 128
 von Dungern, 16, 54, 79, 85,
 91, 125
 Durham, 37</p> <p>Ehrlich, 6, 11, 13, 16, 23, 39,
 47, 58, 67, 84, 87, 92, 97,
 112, 159, 166, 179
 Eisenberg, 39, 111
 Epstein, 205</p> <p>Famulener, 153
 Field, 43
 Fischer, 9
 Fleischmann, 195, 197
 Flexner, 137
 Floyd, 178
 Fodor, 50
 Ford, 17
 Fornet, 196
 Friedberger, 42, 156</p> |
|--|---|

- Gay, 73, 88, 95, 104, 146, 196,
 207
 Gengon, 40, 71, 99, 163, 185,
 196
 Gibson, 18
 Grassberger, 29
 Gruber, 36, 93
 Grünbaum, 37
 Gscheidlen, 50

 Haeckel, 127
 Hahn, 93
 Hamburger, 148
 Hektoen, 130
 Hericourt, 141
 Heubner, 144
 Hiss, 177
 Holzmann, 208

 Jackson, 126
 Jansky, 35
 Jenner, 169
 Jordan, 150

 Kitasato, 1
 Kleine, 166
 Knorr, 2, 5
 Kraus, 108
 Kyes, 137

 Landois, 74, 143
 Landsteiner, 35, 54, 94, 123,
 194, 196
 Leblanc, 111
 Leclainche, 109
 Ledingham, 20
 Leishman, 131
 Levaditi, 194, 198
 Lewis, 150
 Löffler, 100
 Lucas, 178

 Manwaring, 139
 Marie, 194, 198
 Marrassini, 125
 Martin, 5
 Marx, 86
 Meier, 194
 Mertens, 109
 Mesnil, 166
 Metchnikoff, 38, 52, 58, 63,
 82, 84, 91, 93, 99, 121, 127,
 160
 Meyer, 213
 Michaelis, 196
 Moreschi, 97
 Morgenroth, 18, 29, 71, 109
 Moro, 148
 Moss, 35, 98
 Moxter, 86, 93, 123
 Much, 208
 Muir, 67
 Müller, 63, 91, 194
 Myers, 109

 Neisser, 72, 100, 196
 Neudorfer, 143
 Neufeld, 129
 Nissen, 74
 Noguchi, 137, 195, 199, 200
 Nolf, 91
 Nuttall, 50, 74, 110, 115

 Obermayer, 111
 Ottenburg, 205
 Otto, 141
 Opie, 162

 Park, 2, 74
 Pasteur, 169, 171
 Pearce, 126
 Pfaundler, 38

Pfeiffer, 37, 51, 86, 93
Pick, 19, 109, 111
von Pirquet, 142, 147, 164
Plaut, 192
Porges, 194
Potzl, 194

Richet, 141
Rimpau, 129
Rosenau, 142
Roux, 5
Russ, 149

Sachs, 73, 137, 195
Salge, 161
Salmon, 171
Schattenfroh, 29, 93
Schick, 142
Schultz, 150
Schütze, 109, 113, 120
Smith, 141, 161, 171
Southard, 146
Spiro, 109
Steinhardt, 151
Stern, 109, 113, 197
Streng, 207

Tehistowitch, 107
Touissant, 171
Traube, 50

Uhlenhuth, 109, 113

Vallée, 109
Volk, 39

von Wassermann, 13, 18, 72, 79,
86, 92, 93, 99, 109, 113, 158,
185, 192
Wechsberg, 100
Wechselmann, 183
Weichardt, 208
Weigert, 10
Weil, 194
Weil, R., 210
Weleh, 165
Wernicke, 5
Widal, 37
Wile, 198
Wright, 128, 131, 173

Zinsser, 178
Zülzer, 109

SUBJECT INDEX

	PAGE
Abrin.....	21
Absorption, in study of agglutinins.....	46
of complement, <i>see</i> Bordet-Gengou.....	71
and Neisser-Wechsberg.....	100
Active immunization, methods.....	171
Adsorption, as applied to toxin-antitoxin.....	30
elective character.....	31
Agglutination, in typhoid fever.....	37
influence of dilution on.....	34
phenomenon of.....	32
purpose of.....	36
specificity of.....	44
Agglutinin.....	32
against trypanosomes.....	34
effect of heat on.....	39
nature of.....	39
Agglutinoid.....	39
Aluronat, in production of leucocytic exudates.....	177
Alexin, <i>see</i> also complement.....	56
Allergy.....	147
Amboceptor, definition.....	63
unit.....	201
Antibacterial sera, practical value of.....	104
Antibodies, cells concerned in production of.....	11
Anticomplement.....	86, 88
views concerning.....	96
Anticomplementary serum, specificity of.....	90
Anticytotoxin.....	122
Antigen, definition of.....	16
for syphilis test.....	100
Antihæmolysins, their nature.....	86

	PAGE
Antiprecipitin.....	119
Antiserum, for snake venoms.....	137
Antitoxic globulins.....	18
Antitoxin, action on toxins.....	18
diphtheria, production of.....	2
diphtheria, testing of.....	5
historical.....	1
nature of.....	17
quantitative relation to toxin.....	31
relation to toxin.....	22
unit.....	22
Antitrypsin, determination of.....	212
Antivenins.....	139
Anaphylactic shock.....	142
pathology of.....	150
prevention by chloral hydrate.....	153
Anaphylatoxin.....	157
Aggressins.....	158
Anaphylaxis, historical.....	141
relation to precipitins.....	148
relation to serum therapy.....	151
theories of.....	144
Anthrax, symptomatic, toxin of.....	29
Antianaphylaxis.....	143
Asthma, relation to anaphylaxis.....	151
Atrepsy.....	166
Autolysins.....	97
Bacterial precipitins.....	108
Bactericidal serum, clinically.....	105
Bacteriolysins, historical.....	50
Bacteriolysis, of cholera spirilla.....	51
Bacteriotropic substances.....	127
Blood, biological reaction for.....	73, 113
examination for transfusion.....	204
its germicidal power.....	50, 74
cells, agglutination of.....	34
relationship, by means of precipitin test.....	110
test, Neisser-Sachs.....	73

	PAGE
Bordet-Gengou phenomenon.....	71
Brain tissue, effect on tetanus toxin.....	13
Cancer, immunization against.....	125
Casein, precipitin against.....	108
Chemoreceptors.....	181
Chloral hydrate, to prevent anaphylactic shock.....	153
Cholera spirilla, Pfeiffer's test.....	51
Clumping, <i>see</i> Agglutination.....	32
Cobra-lecithid.....	139
Cobra venom.....	137
as test in syphilis.....	210
in haemolysis.....	208
Colloids, coagulability of.....	42
in the study of agglutination.....	42
Common receptors.....	83
Complement.....	64
absorption by precipitates.....	96
absorption of, <i>see</i> Bordet-Gengou.....	71
artificial increase.....	92
definition of.....	63
deflection of.....	100
effect of heat on.....	94
effect of phosphorus poisoning on.....	91
multiplicity of.....	70
source of.....	93
structure of.....	94
Complementoid.....	94
Complementophile group, Bordet's views concerning.....	67
Conglutination.....	207
Copula, definition of.....	63
Croton.....	21
Cytotoxins, definition.....	121
produced by nucleo-proteid injections.....	125
specificity of.....	124
Desmon, definition of.....	63
Deviation of complement.....	100
Dissociation, in toxin-antitoxin action.....	29

	PAGE
Dosage, of lytic sera.....	100
Dyeing, analogous to antitoxin reaction.....	30
Electric charge, of agglutinins.....	43
Electrolytes, in relation to agglutination.....	41
Endotoxins, relation to infection.....	158
Endotoxin theory, applied to anaphylaxis.....	145
Epithelium, cytotoxin for.....	124
serum against.....	85
Fractional neutralization, in study of antitoxins.....	30
Ferments, in leucocytes.....	162
Globulin, antitoxic.....	18
Group agglutination.....	44
Gruber-Widal reaction.....	37
Hæmagglutinins.....	34
Hæmolysin, discovery of.....	52
Hæmorrhagin, of snake venom.....	138
Haptins, definition of.....	16
Haptophore group, nature of.....	7
Heat, effect on serum.....	40, 96
Hypersensitization, to serum.....	142
Immediate reaction.....	145
Immune bodies, site of production.....	86
Immune body.....	64
nature of.....	82
Immunity, acquired.....	160
hereditary transmission.....	160
mechanism of.....	162
natural, variations in... ..	159
relation of anaphylaxis to.....	163
varieties of.....	158
Immunization, active contrasted with passive.....	161
active, methods of.....	171
against diphtheria toxin.....	3
with partially neutralized toxins.....	12

	PAGE
Inactivated serum.....	54
Index, opsonic.....	132
Infection, nature of.....	154
relation of anaphylaxis to.....	156
relation of virus to.....	154
Inflammation, Opie's researches on.....	162
Interbody of normal serum.....	77
Isoagglutinin, occurrence of in normal human sera.....	35
relation to homologous transfusion.....	36
Isolysins, occurrence of in man.....	98
nature of.....	97
Isoprecipitin.....	120
 Lactoserum.....	 108
Lecithid, of cobra venom.....	139
Leucocytes, as source of complement.....	93
ferments in.....	162
phagocytic action of.....	127
Leucocyte extract, in treatment of infections.....	177
Leucotoxin.....	121
Liver, rôle in production of complement.....	91
L_0 and L_{\uparrow}	23
Lysins, relation to agglutinins.....	36
 Mastic, agglutination of.....	 41
Meiostagmin reaction.....	208
Milk, human and bovine tested biologically.....	112
Much-Holzmann reaction.....	208
Mussels, toxin of.....	141
 Negative phase.....	 133
Neisser-Sachs blood test.....	73
Neisser-Wechsberg phenomenon.....	100
Nephrotoxin.....	126
Neurotoxin.....	122
of snake venom.....	138
Noguchi's test for syphilis.....	200
Normal serum, agglutinating power of.....	34
contrasted with specific immune serum.....	79

	PAGE
Normal serum, its hæmolytic and bacteriolytic action	73
Nucleo-proteid, for immunization	125
Opsonic index, clinical value of	135
nature of	131
Opsonin, affinity of, for bacteria	129
historical	127
structure of	130
Organotropism	179
Osmotic pressure, of serum, in infections	208
Parasitotropism	179
Parenteral digestion, of proteid	164
Partial immune bodies	82
Partial saturation method	24
Pfaundler's thread reaction	38
Pfeiffer's phenomenon	51
Phagocytosis	127
Phosphorus poisoning, effect on complement production . . .	91
Physical chemistry, applied to toxin-antitoxin reaction . . .	28
Precipitates, relation to anti-complement action	96
Precipitins, definition of	107
in relation to animal species	110
method of immunization	114
nature of	111
specificity of	109
Precipitin test, for blood	116
various applications of	118
Preparator, definition of	63
Pro zone, in agglutination	39
relation of colloids to	43
Proteid, cleavage of	157
parenteral digestion of	164
Protoxoid	26
Psychoses, serum diagnosis in	209
Ptomaines	20
Rash, from serum injections	143
Receptors, common	83

	PAGE
Receptors, nature of.....	9
various orders of.....	47
Reversible reactions in toxin-antitoxin combination.....	29
Ricin.....	21
Salts, their relation to agglutination.....	41
Salvarsan, chemistry of.....	182
principles governing treatment by.....	179
Serum, active and inactive.....	54
against epithelium.....	125
antibody content of compared to plasma.....	163
effect of heat on.....	40
mode of collection.....	4, 115
Serum diagnosis, in typhoid fever.....	37
Serum-fast trypanosomes.....	166
Serum rashes.....	143
Sessile receptors.....	14
Side-chain theory, applied to antitoxins.....	6
applied to bacteriolysins and hæmolysins.....	68
Snake venom.....	137
Specific therapy, principles of.....	179
Specificity, cause of.....	65
nature of.....	65, 124
of agglutination.....	45
Spectrum, Ehrlich's so-called.....	25
Spermatoxin.....	123
Split products, in anaphylaxis.....	146
Stimulins.....	128
Substance fixatrice.....	63
sensibilatrice.....	55
Syntoxoid.....	27
Syphilis, cobra venom test in.....	210
serum diagnosis of.....	185
specific treatment of.....	179
tests for.....	196
Tetanus toxin, affinity for brain tissue.....	13
Thread reaction.....	38
Toxin-antitoxin reaction.....	22-31

	PAGE
Toxin, absorption of antitoxin by.....	31
diphtheria, preparation of.....	2
relation to antitoxin.....	22
Toxins, characteristics of.....	20
Toxoids, nature of.....	6
various kinds of.....	24
Toxons.....	24
Toxophore group, its influence in immunization.....	16
nature of.....	7
Transfusion, blood tests in.....	204
relation of isoagglutinins to.....	36
Trypanosomes, agglutinins against.....	34
serum-fast.....	166
Typhoid fever, Gruber-Widal reaction in.....	37
Uhlenhuth, blood test.....	116
Vaccines, bacterial.....	169
bacterial, doses.....	175
Vaccine therapy.....	173
Venoms, of snakes.....	137
Viscosity of blood, relation to serum reactions.....	40
Wassermann test, for syphilis.....	185
Weigert's overproduction theory.....	9
Welch's hypothesis.....	165
Widal reaction.....	37
Zootoxins.....	21
Zymotoxic group.....	94



SHORT-TITLE CATALOGUE

OF THE

PUBLICATIONS

OF

JOHN WILEY & SONS

NEW YORK

LONDON: CHAPMAN & HALL, LIMITED

ARRANGED UNDER SUBJECTS

Descriptive circulars sent on application. Books marked with an asterisk (*) are sold at net prices only. All books are bound in cloth unless otherwise stated.

AGRICULTURE—HORTICULTURE—FORESTRY.

Osby's Principles of Animal Nutrition.....	8vo,	\$4 00
Man's Forest Physiography. (In Press.)		
and Hansen's American Horticultural Manual:		
Part I. Propagation, Culture, and Improvement.....	12mo,	1 50
Part II. Systematic Pomology.....	12mo,	1 50
ott's Engineering for Land Drainage.....	12mo,	1 50
Practical Farm Drainage. (Second Edition, Rewritten.).....	12mo,	1 50
es's Forest Mensuration	8vo,	4 00
* Principles of Handling Woodlands.....	Large 12mo,	1 50
n's Principles of American Forestry.....	12mo,	1 50
enfelt's Principles of Modern Dairy Practice. (Woll.).....	12mo,	2 00
rick's Denatured or Industrial Alcohol	8vo,	4 00
mpand Waugh's Landscape Gardening. (New Edition, Rewritten)	12mo,	1 50
Kay and Larsen's Principles and Practice of Butter-making	8vo,	1 50
oard's Landscape Gardening as Applied to Home Decoration.....	12mo,	1 50
erson's Insects Injurious to Staple Crops.....	12mo,	1 50
erson and Headlee's Insects Injurious to Garden Crops. (In Preparation.)		
warz's Longleaf Pine in Virgin Forest.	12mo,	1 25
otaroff's Shade Trees in Towns and Cities.....	8vo,	3 00
bridge's Rocks and Soils.....	8vo,	2 50
on's Microscopy of Vegetable Foods.....	8vo,	7 50
s Handbook for Farmers and Dairymen.....	16mo,	1 50

ARCHITECTURE.

in's Steam Heating for Buildings.....	12mo,	2 50
Buildings and Structures of American Railroads.....	4to,	5 00
ire's Architectural Iron and Steel.....	8vo,	3 50
Compound Riveted Girders as Applied in Buildings.....	8vo,	2 00
Planning and Construction of American Theatres.....	8vo,	3 00

Birkmire's Planning and Construction of High Office Buildings.....	8vo,	\$3 50
Skeleton Construction in Buildings.....	8vo,	3 00
Briggs's Modern American School Buildings.....	8vo,	4 00
Byrne's Inspection of Materials and Workmanship Employed in Construction.....	16mo,	3 00
Carpenter's Heating and Ventilating of Buildings.....	8vo,	4 00
* Corthell's Allowable Pressure on Deep Foundations.....	12mo,	1 25
Freitag's Architectural Engineering.....	8vo,	3 50
Fireproofing of Steel Buildings.....	8vo,	2 50
Gerhard's Guide to Sanitary Inspections. (Fourth Edition, Entirely Revised and Enlarged.).....	12mo,	1 50
* Modern Baths and Bath Houses.....	8vo,	3 00
Sanitation of Public Buildings.....	12mo,	1 50
Theatre Fires and Panics.....	12mo,	1 50
* The Water Supply, Sewerage and Plumbing of Modern City Buildings.....	8vo,	4 00
Johnson's Statics by Algebraic and Graphic Methods.....	8vo,	2 00
Kellaway's How to Lay Out Suburban Home Grounds.....	8vo,	2 00
Kidder's Architects' and Builders' Pocket-book.....	16mo, mor.,	5 00
Merrill's Stones for Building and Decoration.....	8vo,	5 00
Monckton's Stair-building.....	4to,	4 00
Patton's Practical Treatise on Foundations.....	8vo,	5 00
Peabody's Naval Architecture.....	8vo,	7 50
Rice's Concrete-block Manufacture.....	8vo,	2 00
Richey's Handbook for Superintendents of Construction.....	16mo, mor.	4 00
Building Foreman's Pocket Book and Ready Reference.....	16mo, mor.	5 00
* Building Mechanics' Ready Reference Series:		
* Carpenters' and Woodworkers' Edition.....	16mo, mor.	1 50
* Cement Workers' and Plasterers' Edition.....	16mo, mor.	1 50
* Plumbers', Steam-Fitters', and Tanners' Edition.....	16mo, mor.	1 50
* Stone- and Brick-masons' Edition.....	16mo, mor.	1 50
Sabin's House Painting.....	12mo,	1 00
Siebert and Biggin's Modern Stone-cutting and Masonry.....	8vo,	1 50
Snow's Principal Species of Wood.....	8vo,	3 50
Wait's Engineering and Architectural Jurisprudence.....	8vo,	6 00
Law of Contracts.....	Sheep,	6 50
Law of Operations Preliminary to Construction in Engineering and Architecture.....	8vo,	3 00
.....	Sheep,	5 50
Wilson's Air Conditioning.....	12mo,	1 50
Worcester and Atkinson's Small Hospitals, Establishment and Maintenance, Suggestions for Hospital Architecture, with Plans for a Small Hospital.....	12mo,	1 25

ARMY AND NAVY.

Bernadou's Smokeless Powder, Nitro-cellulose, and the Theory of the Cellulose Molecule.....	12mo,	2 50
Chase's Art of Pattern Making.....	12mo,	2 50
Screw Propellers and Marine Propulsion.....	8vo,	3 00
* Cloke's Enlisted Specialists' Examiner.....	8vo,	2 00
* Gunner's Examiner.....	8vo,	1 50
Craig's Azimuth.....	4to,	3 50
Crehore and Squier's Polarizing Photo-chronograph.....	8vo,	3 00
* Davis's Elements of Law.....	8vo,	2 50
* Treatise on the Military Law of United States.....	8vo,	7 00
* Dudley's Military Law and the Procedure of Courts-martial.....	Large 12mo,	2 50
Durand's Resistance and Propulsion of Ships.....	8vo,	5 00
* Dyer's Handbook of Light Artillery.....	12mo,	3 00
Eissler's Modern High Explosives.....	8vo,	4 00
* Fieberger's Text-book on Field Fortification.....	Large 12mo,	2 00
Hamilton and Bond's The Gunner's Catechism.....	18mo,	1 00

* Hoff's Elementary Naval Tactics.....	8vo.	\$1 50
Ingalls's Handbook of Problems in Direct Fire.....	8vo.	4 00
* Lissak's Ordnance and Gunnery.....	8vo.	6 00
* Ludlow's Logarithmic and Trigonometric Tables.....	8vo.	1 00
* Lyons's Treatise on Electromagnetic Phenomena. Vols. I and II.	8vo, each.	6 00
* Mahan's Permanent Fortifications. (Mercur.).....	8vo, half mor.	7 50
Manual for Courts-martial.....	16mo, mor.	1 50
* Mercur's Attack of Fortified Places.....	12mo.	2 00
* Elements of the Art of War.....	8vo.	4 00
Nixon's Adjutants' Manual.....	24mo.	1 00
Peabody's Naval Architecture.....	8vo.	7 50
* Phelps's Practical Marine Surveying.....	8vo.	2 50
Putnam's Nautical Charts.....	8vo.	2 00
Rust's Ex-meridian Altitude, Azimuth and Star Finding Tables.....	8vo.	5 00
* Selkirk's Catechism of Manual of Guard Duty.....	24mo.	50
Sharpe's Art of Subsisting Armies in War.....	18mo, mor.	1 50
* Taylor's Speed and Power of Ships. 2 vols. Text 8vo, plates oblong 4to.		7 50
* Tapes and Poole's Manual of Bayonet Exercises and Musketry Fencing.	24mo, leather.	50
* Weaver's Military Explosives.....	8vo.	3 00
* Woodhull's Military Hygiene for Officers of the Line.....	Large 12mo.	1 50

ASSAYING.

Betts's Lead Refining by Electrolysis.....	8vo.	4 00
*Butler's Handbook of Blowpipe Analysis.....	16mo.	75
Fletcher's Practical Instructions in Quantitative Assaying with the Blowpipe.	16mo, mor.	1 50
Furman and Pardoe's Manual of Practical Assaying.....	8vo.	3 00
Lodge's Notes on Assaying and Metallurgical Laboratory Experiments.....	8vo.	3 00
Low's Technical Methods of Ore Analysis.....	8vo.	3 00
Miller's Cyanide Process.....	12mo.	1 00
Manual of Assaying.....	12mo.	1 00
Minet's Production of Aluminum and its Industrial Use. (Waldo.)...	12mo.	2 50
Ricketts and Miller's Notes on Assaying.....	8vo.	3 00
Robine and Lenglen's Cyanide Industry. (Le Clerc.).....	8vo.	4 00
* Seamon's Manual for Assayers and Chemists.....	Large 12mo.	2 50
Ulke's Modern Electrolytic Copper Refining.....	8vo.	3 00
Wilson's Chlorination Process.....	12mo.	1 50
Cyanide Processes.....	12mo.	1 50

ASTRONOMY.

Comstock's Field Astronomy for Engineers.....	8vo.	2 50
Craig's Azimuth.....	4to.	3 50
Crandall's Text-book on Geodesy and Least Squares.....	8vo.	3 00
Doolittle's Treatise on Practical Astronomy.....	8vo.	4 00
Hayford's Text-book of Geodetic Astronomy.....	8vo.	3 00
Hosmer's Azimuth.....	16mo, mor.	1 00
* Text-book on Practical Astronomy.....	8vo.	2 00
Merriman's Elements of Precise Surveying and Geodesy.....	8vo.	2 50
* Michie and Harlow's Practical Astronomy.....	8vo.	3 00
Rust's Ex-meridian Altitude, Azimuth and Star-Finding Tables.....	8vo.	5 00
* White's Elements of Theoretical and Descriptive Astronomy.....	12mo.	2 00

CHEMISTRY.

* Abderhalden's Physiological Chemistry in Thirty Lectures. (Hall and Defren.).....	8vo.	5 00
* Abegg's Theory of Electrolytic Dissociation. (von Ende.).....	12mo.	1 25
Alexeyeff's General Principles of Organic Syntheses. (Matthews.).....	8vo.	3 00
Allen's Tables for Iron Analysis.....	8vo.	3 00
Armsby's Principles of Animal Nutrition.....	8vo.	4 00
Arnold's Compendium of Chemistry. (Mandel.).....	Large 12mo.	3 50

Association of State and National Food and Dairy Departments, Hartford Meeting, 1906.....	8vo,	\$3 00
Jamestown Meeting, 1907.....	8vo,	3 00
Austen's Notes for Chemical Students.....	12mo,	1 50
Baskerville's Chemical Elements. (In Preparation.)		
Bernadou's Smokeless Powder.—Nitro-cellulose, and Theory of the Cellulose Molecule.....	12mo,	2 50
* Biltz's Introduction to Inorganic Chemistry. (Hall and Phelan.)	12mo,	1 25
Laboratory Methods of Inorganic Chemistry. (Hall and Blanchard.)	8vo,	3 00
* Bingham and White's Laboratory Manual of Inorganic Chemistry.....	12mo,	1 00
* Blanchard's Synthetic Inorganic Chemistry.....	12mo,	1 00
* Browning's Introduction to the Rarer Elements.....	8vo,	1 50
* Butler's Handbook of Blowpipe Analysis.....	16mo,	75
* Claassen's Beet-sugar Manufacture. (Hall and Rolfe.)	8vo,	3 00
Classen's Quantitative Chemical Analysis by Electrolysis. (Boltwood.)	8vo,	3 00
Cohn's Indicators and Test-papers.....	12mo,	2 00
Tests and Reagents.....	8vo,	3 00
* Danneel's Electrochemistry. (Merriam.)	12mo	1 25
Dannerth's Methods of Textile Chemistry.....	12mo,	2 00
Duhem's Thermodynamics and Chemistry. (Burgess.)	8vo,	4 00
Effront's Enzymes and their Applications. (Prescott.)	8vo,	3 00
Eissler's Modern High Explosives.....	8vo,	4 00
* Fischer's Oedema.....	8vo,	2 00
* Physiology of Alimentation.....	Large 12mo,	2 00
Fletcher's Practical Instructions in Quantitative Assaying with the Blowpipe.	16mo, mor.	1 50
Fowler's Sewage Works Analyses.....	12mo,	2 00
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	8vo,	5 00
Manual of Qualitative Chemical Analysis. Part I. Descriptive. (Wells.)	8vo,	3 00
Quantitative Chemical Analysis. (Cohn.) 2 vols.	8vo,	12 50
When Sold Separately, Vol. I, \$6. Vol. II, \$8.		
Fuertes's Water and Public Health.....	12mo	1 50
Furman and Pardoe's Manual of Practical Assaying.....	8vo,	3 00
* Getman's Exercises in Physical Chemistry.....	12mo,	2 00
Gill's Gas and Fuel Analysis for Engineers.....	12mo,	1 25
* Gooch and Browning's Outlines of Qualitative Chemical Analysis.	Large 12mo,	1 25
Grotenfelt's Principles of Modern Dairy Practice. (Woll.)	12mo,	2 00
Groth's Introduction to Chemical Crystallography (Marshall).	12mo,	1 25
* Hammarsten's Text-book of Physiological Chemistry. (Mandel.)	8vo,	4 00
Hanausek's Microscopy of Technical Products. (Winton.)	8vo,	5 00
* Haskins and Macleod's Organic Chemistry.....	12mo,	2 00
* Herrick's Denatured or Industrial Alcohol.....	8vo,	4 00
Hinds's Inorganic Chemistry.....	8vo,	3 00
* Laboratory Manual for Students.....	12mo,	1 00
* Holleman's Laboratory Manual of Organic Chemistry for Beginners. (Walker.)	12mo,	1 00
Text-book of Inorganic Chemistry. (Cooper.)	8vo,	2 50
Text-book of Organic Chemistry. (Walker and Mott.)	8vo,	2 50
* Holley's Lead and Zinc Pigments.....	Large 12mo,	3 00
Holley and Ladd's Analysis of Mixed Paints, Color Pigments, and Varnishes.	Large 12mo,	2 50
Hopkins's Oil-chemists' Handbook.....	8vo,	3 00
Jackson's Directions for Laboratory Work in Physiological Chemistry.	8vo,	1 25
Johnson's Rapid Methods for the Chemical Analysis of Special Steels, Steel-making Alloys and Graphite.....	Large 12mo,	3 00
Landauer's Spectrum Analysis. (Tingle.)	8vo,	3 00
Lassar-Cohn's Application of Some General Reactions to Investigations in Organic Chemistry. (Tingle.)	12mo,	1 00
Leach's Inspection and Analysis of Food with Special Reference to State Control.....	8vo,	7 50
Löb's Electrochemistry of Organic Compounds. (Lorenz.)	8vo,	3 00
Lodge's Notes on Assaying and Metallurgical Laboratory Experiments.	8vo,	3 00
Low's Technical Method of Ore Analysis.....	8vo,	3 00
Lowe's Paint for Steel Structures.....	12mo,	1 00
Lunge's Techno-chemical Analysis. (Cohn.)	12mo,	1 00

* McKay and Larsen's Principles and Practice of Butter making	8vo,	\$1 50
Maire's Modern Pigments and their Vehicles	12mo,	2 00
Mandel's Handbook for Bio chemical Laboratory.....	12mo,	1 50
* Martin's Laboratory Guide to Qualitative Analysis with the Blowpipe	12mo,	0 60
Mason's Examination of Water. (Chemical and Bacteriological).....	12mo,	1 25
Water-supply. (Considered Principally from a Sanitary Standpoint.)	8vo,	4 00
* Mathewson's First Principles of Chemical Theory.....	8vo,	1 00
Matthews's Laboratory Manual of Dyeing and Textile Chemistry.....	8vo,	3 50
Textile Fibres. 2d Edition, Rewritten.....	8vo,	4 00
* Meyer's Determination of Radicles in Carbon Compounds. (Tingle.)	12mo,	1 25
Third Edition.....	12mo,	1 00
Miller's Cyanide Process.....	12mo,	1 00
Manual of Assaying.....	12mo,	1 00
Minet's Production of Aluminum and its Industrial Use. (Waldo.).....	12mo,	2 50
* Mittelstaedt's Technical Calculations for Sugar Works. (Bourbakis.)	12mo,	1 50
Mixer's Elementary Text-book of Chemistry.....	12mo,	1 50
Morgan's Elements of Physical Chemistry.....	12mo,	3 00
Outline of the Theory of Solutions and its Results.....	12mo,	1 00
* Physical Chemistry for Electrical Engineers.....	12mo,	1 50
* Moore's Outlines of Organic Chemistry.....	12mo,	1 50
Morse's Calculations used in Cane-sugar Factories.....	16mo, mor.	1 50
* Muir's History of Chemical Theories and Laws.....	8vo,	4 00
Mulliken's General Method for the Identification of Pure Organic Compounds.		
Vol. I. Compounds of Carbon with Hydrogen and Oxygen. Large 8vo,		5 00
Vol. II. Nitrogenous Compounds. (In Preparation.)		
Vol. III. The Commercial Dyestuffs.....	Large 8vo,	5 00
* Nelson's Analysis of Drugs and Medicines	12mo,	3 00
Ostwald's Conversations on Chemistry. Part One. (Ramsey.).....	12mo,	1 50
" " " " Part Two. (Turnbull.).....	12mo,	2 00
* Introduction to Chemistry. (Hall and Williams.).....	Large 12mo,	1 50
Owen and Standage's Dyeing and Cleaning of Textile Fabrics.....	12mo,	2 00
* Palmer's Practical Test Book of Chemistry.....	12mo,	1 00
* Pauli's Physical Chemistry in the Service of Medicine. (Fischer.).....	12mo,	1 25
Penfield's Tables of Minerals, Including the Use of Minerals and Statistics		
of Domestic Production.....	8vo,	1 00
Pictet's Alkaloids and their Chemical Constitution. (Biddle.).....	8vo,	5 00
Poole's Calorific Power of Fuels.....	8vo,	3 00
Prescott and Winslow's Elements of Water Bacteriology, with Special Refer-		
ence to Sanitary Water Analysis.....	12mo,	1 50
* Reisig's Guide to Piece-Dyeing.....	8vo,	25 00
Richards and Woodman's Air, Water, and Food from a Sanitary Stand-		
point.....	8vo,	2 00
Ricketts and Miller's Notes on Assaying.....	8vo,	3 00
Rideal's Disinfection and the Preservation of Food.....	8vo,	4 00
Sewage and the Bacterial Purification of Sewage.....	8vo,	4 00
Riggs's Elementary Manual for the Chemical Laboratory.....	8vo,	1 25
Robine and Lenglen's Cyanide Industry. (Le Clerc.).....	8vo,	4 00
Ruddiman's Incompatibilities in Prescriptions.....	8vo,	2 00
Whys in Pharmacy.....	12mo,	1 00
* Ruer's Elements of Metallography. (Mathewson.).....	8vo,	3 00
Sabin's Industrial and Artistic Technology of Paint and Varnish.	8vo,	3 00
Salkowski's Physiological and Pathological Chemistry. (Orndorff.).....	8vo,	2 50
Schimpf's Essentials of Volumetric Analysis.....	12mo,	1 25
Manual of Volumetric Analysis. (Fifth Edition, Rewritten).....	8vo,	5 00
* Qualitative Chemical Analysis.....	8vo,	1 25
* Seamon's Manual for Assayers and Chemists.....	Large 12mo,	2 50
Smith's Lecture Notes on Chemistry for Dental Students	8vo,	2 50
Spencer's Handbook for Cane Sugar Manufacturers.....	16mo, mor.	3 00
Handbook for Chemists of Beet-sugar Houses.....	16mo, mor.	3 00
Stockbridge's Rocks and Soils.....	8vo,	2 50
Stone's Practical Testing of Gas and Gas Meters.....	8vo,	3 50
* Tillman's Descriptive General Chemistry.....	8vo,	3 00
* Elementary Lessons in Heat	8vo,	1 50
Treadwell's Qualitative Analysis. (Hall.).....	8vo,	3 00
Quantitative Analysis. (Hall.).....	8vo,	4 00

Smith's Manual of Topographical Drawing. (McMillan.).....	8vo,	\$2 50
Soper's Air and Ventilation of Subways.....	12mo,	2 50
* Tracy's Exercises in Surveying.....	12mo, mor.	1 00
Tracy's Plane Surveying.....	16mo, mor.	3 00
* Trautwine's Civil Engineer's Pocket-book.....	16mo, mor.	5 00
Venable's Garbage Crematories in America.....	8vo,	2 00
Methods and Devices for Bacterial Treatment of Sewage.....	8vo,	3 00
Wait's Engineering and Architectural Jurisprudence.....	8vo,	6 00
	Sheep,	6 50
Law of Contracts.....	8vo,	3 00
Law of Operations Preliminary to Construction in Engineering and Architecture.....	8vo,	5 00
	Sheep,	5 50
Warren's Stereotomy—Problems in Stone-cutting.....	8vo,	2 50
* Waterbury's Vest-Pocket Hand-book of Mathematics for Engineers. 2 $\frac{1}{2}$ × 5 $\frac{1}{2}$ inches, mor.		1 00
* Enlarged Edition, Including Tables.....	mor.	1 50
Webb's Problems in the Use and Adjustment of Engineering Instruments. 16mo, mor.		1 25
Wilson's Topographic Surveying.....	8vo,	3 50

BRIDGES AND ROOFS.

Boller's Practical Treatise on the Construction of Iron Highway Bridges..	8vo,	2 00
* Thames River Bridge.....	Oblong paper,	5 00
Burr and Falk's Design and Construction of Metallic Bridges.....	8vo,	5 00
Influence Lines for Bridge and Roof Computations.....	8vo,	3 00
Du Bois's Mechanics of Engineering. Vol. II.....	Small 4to,	10 00
Foster's Treatise on Wooden Trestle Bridges.....	4to,	5 00
Fowler's Ordinary Foundations.....	8vo,	3 50
Greene's Arches in Wood, Iron, and Stone.....	8vo,	2 50
Bridge Trusses.....	8vo,	2 50
Roof Trusses.....	8vo,	1 25
Grimm's Secondary Stresses in Bridge Trusses.....	8vo,	2 50
Heller's Stresses in Structures and the Accompanying Deformations....	8vo,	3 00
Howe's Design of Simple Roof-trusses in Wood and Steel.....	8vo,	2 00
Symmetrical Masonry Arches.....	8vo,	2 50
Treatise on Arches.....	8vo,	4 00
* Hudson's Plate Girder Design.....	8vo,	1 50
* Jacoby's Structural Details, or Elements of Design in Heavy Framing.	8vo,	2 25
Johnson, Bryan and Turneure's Theory and Practice in the Designing of Modern Framed Structures.....	Small 4to,	10 00
* Johnson, Bryan and Turneure's Theory and Practice in the Designing of Modern Framed Structures. New Edition. Part I.....	8vo,	3 00
* Part II. New Edition.....	8vo,	4 00
Merriman and Jacoby's Text-book on Roofs and Bridges:		
Part I. Stresses in Simple Trusses.....	8vo,	2 50
Part II. Graphic Statics.....	8vo,	2 50
Part III. Bridge Design.....	8vo,	2 50
Part IV. Higher Structures.....	8vo,	2 50
Sondericker's Graphic Statics, with Applications to Trusses, Beams, and Arches.....	8vo,	2 00
Waddell's De Pontibus, Pocket-book for Bridge Engineers.....	16mo, mor.	2 00
* Specifications for Steel Bridges.....	12mo,	50
Waddell and Harrington's Bridge Engineering. (In Preparation.)		

HYDRAULICS.

Barnes's Ice Formation.....	8vo,	3 00
Bazin's Experiments upon the Contraction of the Liquid Vein Issuing from an Orifice. (Trautwine.).....	8vo,	2 00
Bovey's Treatise on Hydraulics.....	8vo,	5 00
Church's Diagrams of Mean Velocity of Water in Open Channels. Oblong 4to, paper,		1 50
Hydraulic Motors.....	8vo,	2 00

Coffin's Graphical Solution of Hydraulic Problems.....	16mo, mor	\$2 50
Flather's Dynamometers, and the Measurement of Power.....	12mo,	3 00
Folwell's Water-supply Engineering.....	8vo,	4 00
Frizell's Water-power.....	8vo,	5 00
Fuertes's Water and Public Health.....	12mo,	1 50
Water-filtration Works.....	12mo,	2 50
Ganguillet and Kutter's General Formula for the Uniform Flow of Water in Rivers and Other Channels. (Hering and Trautwine.).....	8vo,	4 00
Hazen's Clean Water and How to Get It.....	Large 12mo,	1 50
Filtration of Public Water-supplies.....	8vo,	3 00
Hazelhurst's Towers and Tanks for Water-works.....	8vo,	2 50
Herschel's 115 Experiments on the Carrying Capacity of Large, Riveted, Metal Conduits.....	8vo,	2 00
Hoyt and Grover's River Discharge.....	8vo,	2 00
Hubbard and Kiersted's Water-works Management and Maintenance.....	8vo,	4 00
* Lyndon's Development and Electrical Distribution of Water Power.....	8vo,	3 00
Mason's Water-supply. (Considered Principally from a Sanitary Stand- point.).....	8vo,	4 00
Merriman's Treatise on Hydraulics.....	8vo,	5 00
* Molitor's Hydraulics of Rivers, Weirs and Sluices.....	8vo,	2 00
* Morrison and Brodie's High Masonry Dam Design.....	8vo,	1 50
* Richards's Laboratory Notes on Industrial Water Analysis.....	8vo,	50
Schuyler's Reservoirs for Irrigation, Water-power, and Domestic Water- supply. Second Edition, Revised and Enlarged.....	Large 8vo,	6 00
* Thomas and Watt's Improvement of Rivers.....	4to,	6 00
Turneure and Russell's Public Water-supplies.....	8vo,	5 00
* Wegmann's Design and Construction of Dams. 6th Ed., enlarged.....	4to,	6 00
Water-Supply of the City of New York from 1658 to 1895.....	4°o,	10 00
Whipple's Value of Pure Water.....	Large 12mo,	1 00
Williams and Hazen's Hydraulic Tables.....	8vo,	1 50
Wilson's Irrigation Engineering.....	8vo,	4 00
Wood's Turbines.....	8vo,	2 50

MATERIALS OF ENGINEERING.

Baker's Roads and Pavements.....	8vo,	5 00
Treatise on Masonry Construction.....	8vo,	5 00
Black's United States Public Works.....	Oblong 4to,	5 00
Blanchard's Bituminous Roads. (In Preparation.)		
Bleiningger's Manufacture of Hydraulic Cement. (In Preparation.)		
* Bovey's Strength of Materials and Theory of Structures.....	8vo,	7 50
Burr's Elasticity and Resistance of the Materials of Engineering.....	8vo,	7 50
Byrne's Highway Construction.....	8vo,	5 00
Inspection of the Materials and Workmanship Employed in Construction.....	16mo,	3 00
Church's Mechanics of Engineering.....	8vo,	6 00
Du Bois's Mechanics of Engineering.....		
Vol. I. Kinematics, Statics, Kinetics.....	Small 4to,	7 50
Vol. II. The Stresses in Framed Structures, Strength of Materials and Theory of Flexures.....	Small 4to,	10 00
* Eckel's Cements, Lines, and Plasters.....	8vo,	6 00
Stone and Clay Products used in Engineering. (In Preparation.)		
Fowler's Ordinary Foundations.....	8vo,	3 50
* Greene's Structural Mechanics.....	8vo,	2 50
* Holley's Lead and Zinc Pigments.....	Large 12mo,	3 00
Holley and Ladd's Analysis of Mixed Paints, Color Pigments and Varnishes.....	Large 12mo,	2 50
* Hubbard's Dust Preventives and Road Binders.....	8vo,	3 00
Johnson's (C. M.) Rapid Methods for the Chemical Analysis of Special Steels, Steel-making Alloys and Graphite.....	Large 12mo,	3 00
Johnson's (J. B.) Materials of Construction.....	Large 8vo,	5 00
Keep's Cast Iron.....	8vo,	2 50
Lanza's Applied Mechanics.....	8vo,	7 50
Lowe's Paints for Steel Structures.....	12mo,	1 00

Maire's Modern Pigments and their Vehicles.....	12mo.	\$2 00
Maurer's Technical Mechanics.....	8vo	4 00
Merrill's Stones for Building and Decoration.....	8vo	5 00
Merriman's Mechanics of Materials.....	8vo.	5 00
* Strength of Materials.....	12mo.	1 00
Metcalf's Steel. A Manual for Steel-users.....	12mo.	2 00
Morrison's Highway Engineering.....	8vo.	2 50
Murdock's Strength of Materials. (In Press.)		
Patton's Practical Treatise on Foundations.....	8vo.	5 00
Rice's Concrete Block Manufacture.....	8vo.	2 00
Richardson's Modern Asphalt Pavement.....	8vo.	3 00
Riehey's Building Foreman's Pocket Book and Ready Reference.....	16mo, mor	5 00
* Cement Workers' and Plasterers' Edition (Building Mechanics' Ready Reference Series).....	16mo, mor.	1 50
Handbook for Superintendents of Construction.....	16mo, mor.	4 00
* Stone and Brick Masons' Edition (Building Mechanics' Ready Reference Series).....	16mo, mor.	1 50
* Ries's Clays: Their Occurrence, Properties, and Uses.....	8vo.	5 00
* Ries and Leighton's History of the Clay-working Industry of the United States.....	8vo.	2 50
Sabin's Industrial and Artistic Technology of Paint and Varnish.....	8vo.	3 00
* Smith's Strength of Material.....	12mo	1 25
Snow's Principal Species of Wood.....	8vo.	3 50
Spalding's Hydraulic Cement.....	12mo.	2 00
Text-book on Roads and Pavements.....	12mo.	2 00
* Taylor and Thompson's Extracts on Reinforced Concrete Design.....	8vo.	2 00
Treatise on Concrete, Plain and Reinforced.....	8vo.	5 00
Thurston's Materials of Engineering. In Three Parts.....	8vo.	8 00
Part I. Non-metallic Materials of Engineering and Metallurgy.....	8vo.	2 00
Part II. Iron and Steel.....	8vo.	3 50
Part III. A Treatise on Brasses, Bronzes, and Other Alloys and their Constituents.....	8vo.	2 50
Tillson's Street Pavements and Paving Materials.....	8vo.	1 00
* Trautwine's Concrete, Plain and Reinforced.....	16mo.	2 00
Turneure and Maurer's Principles of Reinforced Concrete Construction. Second Edition, Revised and Enlarged.....	8vo.	3 50
Waterbury's Cement Laboratory Manual.....	12mo.	1 00
Wood's (De V.) Treatise on the Resistance of Materials, and an Appendix on the Preservation of Timber.....	8vo.	2 00
Wood's (M P) Rustless Coatings: Corrosion and Electrolysis of Iron and Steel.....	8vo.	4 00

RAILWAY ENGINEERING.

Andrews's Handbook for Street Railway Engineers.....	3×5 inches, mor	1 25
Berg's Buildings and Structures of American Railroads.....	4to.	5 00
Brooks's Handbook of Street Railroad Location.....	16mo, mor.	1 50
Burt's Railway Station Service.....	(In Press.) 12mo.	
Butts's Civil Engineer's Field-book.....	16mo, mor.	2 50
Crandall's Railway and Other Earthwork Tables.....	8vo.	1 50
Crandall and Barnes's Railroad Surveying.....	16mo. mor.	2 00
* Crockett's Methods for Earthwork Computations.....	8vo.	1 50
Dredge's History of the Pennsylvania Railroad. (1879).....	Paper.	5 00
Fisher's Table of Cubic Yards.....	Cardboard.	25
Godwin's Railroad Engineers' Field-book and Explorers' Guide.....	16mo, mor.	2 50
Hudson's Tables for Calculating the Cubic Contents of Excavations and Embankments.....	8vo.	1 00
Ives and Hilt's Problems in Surveying, Railroad Surveying and Geodesy.....	16mo, mor.	1 50
Molitor and Beard's Manual for Resident Engineers.....	16mo.	1 00
Nagle's Field Manual for Railroad Engineers.....	16mo, mor.	3 00
* Orrock's Railroad Structures and Estimates.....	8vo.	3 00
Philbrick's Field Manual for Engineer.....	16mo, mor.	3 00
Raymond's Railroad Field Geometry.....	16mo, mor.	2 00
Elements of Railroad Engineering.....	8vo.	3 50
Railroad Engineer's Field Book (In Preparation)		
Roberts' Track Formulæ and Tables.....	16mo, mor.	3 00

Searles's Field Engineering.....	16mo, mor.	\$3 00
Railroad Spiral.....	16mo, mor.	1 50
Taylor's Prismatical Formulae and Earthwork.....	8vo,	1 50
* Tautwine's Field Practice of Laying Out Circular Curves for Railroads.....	12mo, mor.	2 50
* Method of Calculating the Cubic Contents of Excavations and Embankments by the Aid of Diagrams.....	8vo,	2 00
Webb's Economics of Railroad Construction.....	Large 12mo,	2 50
Railroad Construction.....	16mo, mor.	5 00
Wellington's Economic Theory of the Location of Railways.....	Large 12mo,	5 00
Wilson's Elements of Railroad-Track and Construction.....	12mo,	2 00

DRAWING

Barr's Kinematics of Machinery.....	8vo,	2 50
* Bartlett's Mechanical Drawing.....	8vo,	3 00
* " " " " Abridged Ed.....	8vo,	1 50
* Bartlett and Johnson's Engineering Descriptive Geometry.....	8vo,	1 50
Coolidge's Manual of Drawing.....	8vo, paper,	1 00
Coolidge and Freeman's Elements of General Drafting for Mechanical Engineers.....	Oblong 4to,	2 50
Durley's Kinematics of Machines.....	8vo,	4 00
Emch's Introduction to Projective Geometry and its Application.....	8vo,	2 50
Hill's Text-book on Shades and Shadows, and Perspective.....	8vo,	2 00
Jamison's Advanced Mechanical Drawing.....	8vo,	2 00
Elements of Mechanical Drawing.....	8vo,	2 50
Jones's Machine Design:		
Part I. Kinematics of Machinery.....	8vo,	1 50
Part II. Form, Strength, and Proportions of Parts.....	8vo,	3 00
Kaup's Text-book on Machine Shop Practice. (In Press.)		
* Kimball and Barr's Machine Design.....	8vo,	3 00
MacCord's Elements of Descriptive Geometry.....	8vo,	3 00
Kinematics; or, Practical Mechanism.....	8vo,	5 00
Mechanical Drawing.....	4to,	4 00
Velocity Diagrams.....	8vo,	1 50
McLeod's Descriptive Geometry.....	Large 12mo,	1 50
* Mahan's Descriptive Geometry and Stone-cutting.....	8vo,	1 50
Industrial Drawing. (Thompson.).....	8vo,	3 50
Moyer's Descriptive Geometry.....	8vo,	2 00
Reed's Topographical Drawing and Sketching.....	4to,	5 00
* Rei's Mechanical Drawing. (Elementary and Advanced.).....	8vo,	2 00
Text-book of Mechanical Drawing and Elementary Machine Design.....	8vo,	3 00
Robinson's Principles of Mechanism.....	8vo,	3 00
Schwamb and I Merrill's Elements of Mechanism.....	8vo,	3 00
Smith (A. W.) and Marx's Machine Design.....	8vo,	3 00
Smith's (R. S.) Manual of Topographical Drawing. (McMillan.).....	8vo,	2 50
* Titsworth's Elements of Mechanical Drawing.....	Oblong 8vo,	1 25
Tracy and North's Descriptive Geometry. (In Press.)		
Warren's Drafting Instruments and Operations.....	12mo,	1 25
Elements of Descriptive Geometry, Shadows, and Perspective.....	8vo,	3 50
Elements of Machine Construction and Drawing.....	8vo,	7 50
Elements of Plane and Solid Free-hand Geometrical Drawing....	12mo,	1 00
General Problems of Shades and Shadows.....	8vo,	3 00
Manual of Elementary Problems in the Linear Perspective of Forms and Shadow.....	12mo,	1 00
Manual of Elementary Projection Drawing.....	12mo,	1 50
Plane Problems in Elementary Geometry.....	12mo,	1 25
Weisbach's Kinematics and Power of Transmission. (Hermann and Klein.).....	8vo,	5 00
Wilson's (H. M.) Topographic Surveying.....	8vo,	3 50
* Wilson's (V. T.) Descriptive Geometry.....	8vo,	1 50
Free-hand Lettering.....	8vo,	1 00
Free-hand Perspective.....	8vo,	2 50
Woolf's Elementary Course in Descriptive Geometry.....	Large 8vo,	3 00

ELECTRICITY AND PHYSICS.

* Abegg's Theory of Electrolytic Dissociation. (von Ende.).....	12mo,	\$1 25
Andrews's Hand-book for Street Railway Engineering.....	3×5 inches, mor.	1 25
Anthony and Ball's Lecture notes on the Theory of Electrical Measurements.....	12mo,	1 00
Anthony and Brackett's Text-book of Physics. (Magic.)....	Large 12mo,	3 00
Benjamin's History of Electricity.....	8vo,	3 00
Betts's Lead Refining and Electrolysis.....	8vo,	4 00
Classen's Quantitative Chemical Analysis by Electrolysis. (Boltwood.)	8vo,	3 00
* Collins's Manual of Wireless Telegraphy and Telephony.....	12mo,	1 50
Crehore and Squier's Polarizing Photo-chronograph.....	8vo,	3 00
* Dannel's Electrochemistry. (Merriam.).....	12mo,	1 25
Dawson's "Engineering" and Electric Traction Pocket-book....	16mo, mor.	5 00
Dolezalek's Theory of the Lead Accumulator (Storage Battery). (von Ende.)	12mo,	2 50
Duhem's Thermodynamics and Chemistry. (Burgess.).....	8vo,	4 00
Flather's Dynamometers, and the Measurement of Power.....	12mo,	3 00
* Getman's Introduction to Physical Science.....	12mo,	1 50
Gilbert's De Magnete (Mottelay).....	8vo,	2 50
* Manchestt's Alternating Currents.....	12mo,	1 00
Hering's Ready Reference Tables (Conversion Factors).....	16mo, mor.	2 50
* Hobart and Eils's High-speed Dynamo Electric Machinery.....	8vo,	6 00
Holman's Precision of Measurements.....	8vo,	2 00
Telescopic Mirror-scale Method, Adjustments, and Tests....	Large 8vo,	75.
* Hutchinson's High-Efficiency Electrical Illuminants and Illumination.	Large 12mo,	2 50
Karapetoff's Experimental Electrical Engineering:		
* Vol. I.....	8vo,	3 50
* Vol. II.....	8vo,	2 50
Kinzbrunner's Testing of Continuous-current Machines.....	8vo,	2 00
Landauer's Spectrum Analysis. (Tingle.).....	8vo,	3 00
Le Chatelier's High-temperature Measurements. (Boudouard—Burgess.)	12mo,	3 00
Löb's Electrochemistry of Organic Compounds. (Lorenz.).....	8vo,	3 00
* Lyndon's Development and Electrical Distribution of Water Power..	8vo,	3 00
* Lyons's Treatise on Electromagnetic Phenomena. Vols. I and II.	8vo, each,	6 00
* Michie's Elements of Wave Motion Relating to Sound and Light....	8vo	4 00
Morgan's Outline of the Theory of Solution and its Results.....	12mo,	1 00
* Physical Chemistry for Electrical Engineers.....	12mo,	1 50
* Norris's Introduction to the Study of Electrical Engineering.....	8vo,	2 50
Norris and Dennison's Course of Problems on the Electrical Characteristics of Circuits and Machines. (In Press.)		
* Parshall and Hobart's Electric Machine Design.....	4to, half mor,	12 50
Reagan's Locomotives: Simple, Compound, and Electric. New Edition.	Large 12mo,	3 50
* Rosenberg's Electrical Engineering. (Haldane Gee—Kinzbrunner.)..	8vo,	2 00
Ryan, Norris and Hoxie's Electrical Machinery. Vol. I.....	8vo,	2 50
Schapper's Laboratory Guide for Students in Physical Chemistry....	12mo,	1 00
* Tillman's Elementary Lessons in Heat.....	8vo,	1 50
* Timbie's Elements of Electricity.....	Large 12mo,	2 00
Tory and Pitcher's Manual of Laboratory Physics.....	Large 12mo,	2 00
Uike's Modern Electrolytic Copper Refining.....	8vo,	3 00
Waters's Commercial Dynamo Design. (In Press.)		

LAW.

* Brennan's Hand-book of Useful Legal Information for Business Men.	16mo, mor.	5 00
* Davis's Elements of Law.....	8vo,	2 50
* Treatise on the Military Law of United States.....	8vo,	7 00
* Dudley's Military Law and the Procedure of Courts-martial..	Large 12mo,	2 50
Manual for Courts-martial.....	16mo, mor.	1 50
Wait's Engineering and Architectural Jurisprudence.....	8vo,	6 00
	Sheep,	6 50

Wait's Law of Contracts.....	8vo,	\$3 00
Law of Operations Preliminary to Construction in Engineering and Architecture.....	8vo,	5 00
	Sheep,	5 50

MATHEMATICS

Baker's Elliptic Functions.....	8vo,	1 50
Briggs's Elements of Plane Analytic Geometry (Bôcher).....	12mo,	1 00
* Buchanan's Plane and Spherical Trigonometry.....	8vo,	1 00
Byerly's Harmonic Functions.....	8vo	1 00
Chandler's Elements of the Infinitesimal Calculus.....	12mo,	2 00
* Coffin's Vector Analysis.....	12mo,	2 50
Compton's Manual of Logarithmic Computations.....	12mo,	1 50
* Dickson's College Algebra.....	Large 12mo,	1 50
* Introduction to the Theory of Algebraic Equations.....	Large 12mo,	1 25
Emch's Introduction to Projective Geometry and its Application.....	8vo,	2 50
Fiske's Functions of a Complex Variable.....	8vo,	1 00
Halsted's Elementary Synthetic Geometry.....	8vo,	1 50
Elements of Geometry.....	8vo,	1 75
* Rational Geometry.....	12mo	1 50
Synthetic Projective Geometry.....	8vo,	1 00
* Hancock's Lectures on the Theory of Elliptic Functions.....	8vo,	5 00
Hyde's Grassmann's Space Analysis.....	8vo,	1 00
* Johnson's (J. B.) Three-place Logarithmic Tables: Vest-pocket size, paper,		15
* 100 copies,		5 00
* Mounted on heavy cardboard, 8×10 inches,		25
* 10 copies,		2 00
Johnson's (W. W.) Abridged Editions of Differential and Integral Calculus.....	Large 12mo, 1 vol.	2 50
Curve Tracing in Cartesian Co-ordinates.....	12mo,	1 00
Differential Equations.....	8vo,	1 00
Elementary Treatise on Differential Calculus.....	Large 12mo,	1 50
Elementary Treatise on the Integral Calculus.....	Large 12mo,	1 50
* Theoretical Mechanics.....	12mo,	3 00
Theory of Errors and the Method of Least Squares.....	12mo,	1 50
Treatise on Differential Calculus.....	Large 12mo,	3 00
Treatise on the Integral Calculus.....	Large 12mo,	3 00
Treatise on Ordinary and Partial Differential Equations... Large 12mo,		3 50
Karapetoff's Engineering Applications of Higher Mathematics. (In Preparation)		
Laplace's Philosophical Essay on Probabilities. (Truscott and Emory.)	12mo,	2 00
* Ludlow's Logarithmic and Trigonometric Tables.....	8vo,	1 00
* Ludlow and Bass's Elements of Trigonometry and Logarithmic and Other Tables.....	8vo,	3 00
* Trigonometry and Tables published separately.....	Each,	2 00
Macfarlane's Vector Analysis and Quaternions.....	8vo,	1 00
McMahon's Hyperbolic Functions.....	8vo,	1 00
Manning's Irrational Numbers and their Representation by Sequences and Series.....	12mo,	1 25
Mathematical Monographs. Edited by Mansfield Merriman and Robert S. Woodward.....	Octavo, each	1 00
No. 1 History of Modern Mathematics, by David Eugene Smith.		
No. 2 Synthetic Projective Geometry, by George Bruce Halsted.		
No. 3 Determinants, by Lucas Gifford Weld. No. 4. Hyper-		
bolic Functions, by James McMahon. No. 5. Harmonic Func-		
tions, by William E. Byerly. No. 6 Grassmann's Space Analysis,		
by Edward W. Hyde. No. 7. Probability and Theory of Errors,		
by Robert S. Woodward. No. 8. Vector Analysis and Quaternions,		
by Alexander Macfarlane. No. 9. Differential Equations, by		
William Woolsey Johnson. No. 10. The Solution of Equations,		
by Mansfield Merriman. No. 11. Functions of a Complex Variable,		
by Thomas S. Fiske.		
Maurer's Technical Mechanics.....	8vo	4 00
Merriman's Method of Least Squares.....	8vo	2 00
Solution of Equations.....	8vo	1 00

* Moritz's Elements of Plane Trigonometry.....	8vo,	\$2 00
Rice and Johnson's Differential and Integral Calculus. 2 vols. in one.		
	Large 12mo,	1 50
Elementary Treatise on the Differential Calculus.	Large 12mo,	3 00
Smith's History of Modern Mathematics.	8vo,	1 00
* Veblen and Lennes's Introduction to the Real Infinitesimal Analysis of One Variable.	8vo,	2 00
* Waterbury's Vest Pocket Hand-book of Mathematics for Engineers.		
	2 $\frac{7}{8}$ × 5 $\frac{3}{8}$ inches, mor.	1 00
* Enlarged Edition, Including Tables	mor.	1 50
Weld's Determinants.	8vo,	1 00
Wood's Elements of Co-ordinate Geometry.	8vo,	2 00
Woodward's Probability and Theory of Errors.	8vo,	1 00

MECHANICAL ENGINEERING.

MATERIALS OF ENGINEERING. STEAM-ENGINES AND BOILERS.

Bacon's Forge Practice.	12mo,	1 50
Baldwin's Steam Heating for Buildings.	12mo,	2 50
Barr's Kinematics of Machinery.	8vo,	2 50
* Bartlett's Mechanical Drawing.	8vo,	3 00
* " " " Abridged Ed.	8vo,	1 50
* Bartlett and Johnson's Engineering Descriptive Geometry.	8vo,	1 50
* Burr's Ancient and Modern Engineering and the Isthmian Canal.	8vo,	3 50
Carpenter's Experimental Engineering.	8vo,	6 00
Heating and Ventilating Buildings.	8vo,	4 00
* Clerk's The Gas, Petrol and Oil Engine.	8vo,	4 00
Compton's First Lessons in Metal Working.	12mo,	1 50
Compton and De Groodt's Speed Lathe.	12mo,	1 50
Coolidge's Manual of Drawing.	8vo, paper,	1 00
Coolidge and Freeman's Elements of General Drafting for Mechanical Engineers.	Oblong 4to,	2 50
Cromwell's Treatise on Belts and Pulleys.	12mo,	1 50
Treatise on Toothed Gearing.	12mo,	1 50
Dingey's Machinery Pattern Making.	12mo,	2 00
Durley's Kinematics of Machines.	8vo,	4 00
Flanders's Gear-cutting Machinery.	Large 12mo,	3 00
Flather's Dynamometers and the Measurement of Power.	12mo,	3 00
Rope Driving.	12mo,	2 00
Gill's Gas and Fuel Analysis for Engineers.	12mo,	1 25
Goss's Locomotive Sparks.	8vo,	2 00
Greene's Pumping Machinery. (In Press.)		
Hering's Ready Reference Tables (Conversion Factors).	16mo, mor.	2 50
* Hobart and Ellis's High Speed Dynamo Electric Machinery.	8vo,	6 00
Hutton's Gas Engine.	8vo,	5 00
Jamison's Advanced Mechanical Drawing.	8vo,	2 00
Elements of Mechanical Drawing.	8vo,	2 50
Jones's Gas Engine.	8vo,	4 00
Machine Design:		
Part I. Kinematics of Machinery.	8vo,	1 50
Part II. Form, Strength, and Proportions of Parts.	8vo,	3 00
Kaup's Text-book on Machine Shop Practice. (In Press.)		
* Kent's Mechanical Engineer's Pocket-Book	16mo, mor.	5 00
Kerr's Power and Power Transmission.	8vo,	2 00
* Kinball and Barr's Machine Design.	8vo,	3 00
Leonard's Machine Shop Tools and Methods.	8vo,	4 00
* Levin's Gas Engine.	8vo,	1 00
* Lorenz's Modern Refrigerating Machinery. (Pope, Haven, and Dean).	8vo,	4 00
MacCord's Kinematics; or Practical Mechanism.	8vo,	5 00
Mechanical Drawing.	4to,	4 00
Velocity Diagrams.	8vo,	1 50
MacFarland's Standard Reduction Factors for Gases.	8vo,	1 50
Mahar's Industrial Drawing. (Thompson).	8vo,	3 50

Mehrtens's Gas Engine Theory and Design.....	Large 12mo.	\$2 50
Oberg's Handbook of Small Tools.....	Large 12mo.	2 50
* Parshall and Hobart's Electric Machine Design. Small 4to, half leather,	12 50	
* Peele's Compressed Air Plant for Mines. Second Edition, Revised and En-		
larged.....	8vo,	3 50
Poole's Calorific Power of Fuels.....	8vo,	3 00
* Porter's Engineering Reminiscences, 1855 to 1882.....	8vo,	3 00
* Reid's Mechanical Drawing. (Elementary and Advanced.).....	8vo,	2 00
Text-book of Mechanical Drawing and Elementary Machine Design.....	8vo,	3 00
Richards's Compressed Air.....	12mo,	1 50
Robinson's Principles of Mechanism.....	8vo,	3 00
Schwamb and Merrill's Elements of Mechanism.....	8vo,	3 00
Smith (A. W.) and Marx's Machine Design.....	8vo,	3 00
Smith's (O.) Press-working of Metals.....	8vo,	3 00
Sorel's Carbureting and Combustion in Alcohol Engines. (Woodward and		
Preston.).....	Large 12mo,	3 00
Stone's Practical Testing of Gas and Gas Meters.....	8vo,	3 50
Thurston's Animal as a Machine and Prime Motor, and the Laws of Energetics.		
	12mo,	1 00
Treatise on Friction and Lost Work in Machinery and Mill Work.....	8vo,	3 00
* Tillson's Complete Automobile Instructor.....	16mo,	1 50
* Titworth's Elements of Mechanical Drawing.....	Oblong 8vo,	1 25
Warren's Elements of Machine Construction and Drawing.....	8vo,	7 50
* Waterbury's Vest Pocket Hand-book of Mathematics for Engineers.		
	2½ × 5½ inches, mor.	1 00
* Enlarged Edition, Including Tables.....	mor.	1 50
Weisbach's Kinematics and the Power of Transmission. (Herrmann—		
Klein.).....	8vo,	5 00
Machinery of Transmission and Governors. (Herrmann—Klein.).....	8vo,	5 00
Wood's Turbines.....	8vo,	2 50

MATERIALS OF ENGINEERING.

* Bovey's Strength of Materials and Theory of Structures.....	8vo,	7 50
Burr's Elasticity and Resistance of the Materials of Engineering.....	8vo,	7 50
Church's Mechanics of Engineering.....	8vo,	6 00
* Greene's Structural Mechanics.....	8vo,	2 50
* Holley's Lead and Zinc Pigments.....	Large 12mo	3 00
Holley and Ladd's Analysis of Mixed Paints, Color Pigments, and Varnishes.		
	Large 12mo,	2 50
Johnson's (C. M.) Rapid Methods for the Chemical Analysis of Special		
Steels, Steel-Making Alloys and Graphite.....	Large 12mo,	3 00
Johnson's (J. B.) Materials of Construction.....	8vo,	6 00
Keep's Cast Iron.....	8vo,	2 50
Lanza's Applied Mechanics.....	8vo,	7 50
Lowe's Paints for Steel Structures.....	12mo,	1 00
Maire's Modern Pigments and their Vehicles.....	12mo,	2 00
Maurer's Technical Mechanics.....	8vo,	4 00
Merriman's Mechanics of Materials.....	8vo,	5 00
* Strength of Materials.....	12mo,	1 00
Metcalf's Steel. A Manual for Steel-users.....	12mo,	2 00
Sabin's Industrial and Artistic Technology of Paint and Varnish.....	8vo,	3 00
Smith's (A. W.) Materials of Machines.....	12mo,	1 00
* Smith's (H. E.) Strength of Material.....	12mo,	1 25
Thurston's Materials of Engineering.....	3 vols., 8vo,	8 00
Part I. Non-metallic Materials of Engineering.....	8vo,	2 00
Part II. Iron and Steel.....	8vo,	3 50
Part III. A Treatise on Brasses, Bronzes, and Other Alloys and their		
Constituents.....	8vo,	2 50
Wood's (De V.) Elements of Analytical Mechanics.....	8vo,	3 00
Treatise on the Resistance of Materials and an Appendix on the		
Preservation of Timber.....	8vo,	2 00
Wood's (M. P.) Rustless Coatings: Corrosion and Electrolysis of Iron and		
Steel.....	8vo,	4 00

STEAM-ENGINES AND BOILERS.

Berry's Temperature-entropy Diagram. Third Edition Revised and Enlarged.	12mo,	\$2 50
Carnot's Reflections on the Motive Power of Heat. (Thurston)	12mo,	1 50
Chase's Art of Pattern Making.	12mo,	2 50
Creighton's Steam-engine and other Heat Motors.	8vo,	5 00
Dawson's "Engineering" and Electric Traction Pocket-book.	16mo, mor,	5 00
* Gebhardt's Steam Power Plant Engineering.	8vo,	6 00
Goss's Locomotive Performance.	8vo,	5 00
Hemenway's Indicator Practice and Steam-engine Economy.	12mo,	2 00
Hutton's Heat and Heat-engines.	8vo,	5 00
Mechanical Engineering of Power Plants.	8vo,	5 00
Kent's Steam Boiler Economy.	8vo,	4 00
Kneass's Practice and Theory of the Injector.	8vo,	1 50
MacCord's Slide-valves.	8vo,	2 00
Meyer's Modern Locomotive Construction.	4to,	10 00
Moyer's Steam Turbine.	8vo,	4 00
Peabody's Manual of the Steam-engine Indicator.	12mo,	1 50
Tables of the Properties of Steam and Other Vapors and Temperature-Entropy Table.	8vo,	1 00
Thermodynamics of the Steam-engine and Other Heat-engines.	8vo,	5 00
Valve-gears for Steam-engines.	8vo,	2 50
Peabody and Miller's Steam-boilers.	8vo,	4 00
Pupin's Thermodynamics of Reversible Cycles in Gases and Saturated Vapors. (Osterberg.)	12mo,	1 25
Reagan's Locomotives: Simple, Compound, and Electric. New Edition.	Large 12mo,	3 50
Sinclair's Locomotive Engine Running and Management.	12mo,	2 00
Smart's Handbook of Engineering Laboratory Practice.	12mo,	2 50
Snow's Steam-boiler Practice.	8vo,	3 00
Spangler's Notes on Thermodynamics.	12mo,	1 00
Valve-gears.	8vo,	2 50
Spangler, Greene, and Marshall's Elements of Steam-engineing.	8vo,	3 00
Thomas's Steam-turbines.	8vo,	4 00
Thurston's Handbook of Engine and Boiler Trials, and the Use of the Indicator and the Prony Brake.	8vo,	5 00
Handy Tables.	8vo,	1 50
Manual of Steam-boilers, their Designs, Construction, and Operation	8vo,	5 00
Manual of the Steam-engine.	2 vols., 8vo,	10 00
Part I. History, Structure, and Theory	8vo,	6 00
Part II. Design, Construction, and Operation	8vo,	6 00
Wehrenfennig's Analysis and Softening of Boiler Feed-water. (Patterson)	8vo,	4 00
Weisbach's Heat, Steam, and Steam-engines. (Du Bois.)	8vo,	5 00
Whitham's Steam-engine Design	8vo,	5 00
Wood's Thermodynamics, Heat Motors, and Refrigerating Machines.	8vo,	4 00

MECHANICS PURE AND APPLIED.

Church's Mechanics of Engineering.	8vo,	6 00
* Mechanics of Internal Works.	8vo,	1 50
Notes and Examples in Mechanics.	8vo,	2 00
Dana's Text-book of Elementary Mechanics for Colleges and Schools	12mo,	1 50
Du Bois's Elementary Principles of Mechanics:		
Vol. I. Kinematics.	8vo,	3 50
Vol. II. Statics.	8vo,	4 00
Mechanics of Engineering. Vol. I.	Small 4to,	7 50
Vol. II.	Small 4to,	10 00
* Greene's Structural Mechanics.	8vo,	2 50
* Hartmann's Elementary Mechanics for Engineering Students.	12mo,	1 25
James's Kinematics of a Point and the Rational Mechanics of a Particle.	Large 12mo,	2 00
* Johnson's (W. W.) Theoretical Mechanics.	12mo,	3 00
Lanza's Applied Mechanics.	8vo,	7 50

Smith's Materials of Machine.....	12mo.	\$1 00
Tate and Stone's Foundry Practice.....	12mo.	2 00
Thurston's Materials of Engineering. In Three Parts.	8vo.	8 00
Part I. Non metallic Materials of Engineering, see Civil Engineering, page 9.		
Part II. Iron and Steel.....	8vo.	3 50
Part III. A Treatise on Brasses, Bronzes, and Other Alloys and their Constituents.....	8vo.	2 50
Ulke's Modern Electrolytic Copper Refining.....	8vo.	3 00
West's American Foundry Practice.....	12mo.	2 50
Moulders' Text Book.....	12mo.	2 50

MINERALOGY.

Baskerville's Chemical Elements. (In Preparation.).....	8vo.	1 50
* Browning's Introduction to the Rarer Elements.....	8vo.	4 00
Bruhl's Manual of Determinative Mineralogy. (Penfield.)....	16mo, mor.	3 00
Butler's Pocket Hand-book of Minerals.....	8vo, paper,	1 00
Cloth,		1 25
* Crane's Gold and Silver.....	8vo.	5 00
Dana's First Appendix to Dana's New "System of Mineralogy".....	Large 8vo.	1 00
Dana's Second Appendix to Dana's New "System of Mineralogy."		
Large 8vo.		1 50
Manual of Mineralogy and Petrography.....	12mo.	2 00
Minerals and How to Study Them.....	12mo.	1 50
System of Mineralogy.....	Large 8vo, half leather,	12 50
Text-book of Mineralogy.....	8vo.	4 00
Douglas's Untechnical Addresses on Technical Subjects.....	12mo.	1 00
Eakle's Mineral Tables.....	8vo.	1 25
Eckel's Stone and Clay Products Used in Engineering. (In Preparation.)		
Goessel's Minerals and Metals: A Reference Book.....	16mo, mor.	3 00
* Groth's The Optical Properties of Crystals. (Jackson.).....	8vo.	3 50
Groth's Introduction to Chemical Crystallography (Marshall).....	12mo.	1 25
* Hayes's Handbook for Field Geologists.....	16mo, mor.	1 50
Iddings's Igneous Rocks.....	8vo.	5 00
Rock Minerals.....	8vo.	5 00
Johannsen's Determination of Rock-forming Minerals in Thin Sections.	8vo.	
With Thumb Index		5 00
* Martin's Laboratory Guide to Qualitative Analysis with the Blow- pipe.....	12mo.	60
Merrill's Non-metallic Minerals: Their Occurrence and Uses.....	8vo.	4 00
Stones for Building and Decoration.....	8vo.	5 00
* Penfield's Notes on Determinative Mineralogy and Record of Mineral Tests.	8vo, paper,	50
Tables of Minerals, Including the Use of Minerals and Statistics of Domestic Production.....	8vo.	1 00
* Pirsson's Rocks and Rock Minerals.....	12mo.	2 50
* Richards's Synopsis of Mineral Characters.....	12mo, mor.	1 25
* Ries's Clays: Their Occurrence, Properties and Uses.....	8vo.	5 00
* Ries and Leighton's History of the Clay-working Industry of the United States.....	8vo.	2 50
* Rowe's Practical Mineralogy Simplified.....	12mo.	1 25
* Tillman's Text-book of Important Minerals and Rocks.....	8vo.	2 00
Washington's Manual of the Chemical Analysis of Rocks.....	8vo.	2 00

MINING.

* Beard's Mine Gases and Explosions.....	Large 12mo.	3 00
* Crane's Gold and Silver.....	8vo.	5 00
* Index of Mining Engineering Literature.....	8vo.	4 00
* Ore Mining Methods.....	* 8vo, mor.	5 00
Dana and Saunders's Rock Drilling. (In Press.)	8vo.	3 00
Douglas's Untechnical Addresses on Technical Subjects.....	12mo.	1 00
Eissler's Modern High Explosives.....	8vo.	4 00

Goesel's Minerals and Metals: A Reference Book.....	16mo, mor.	\$3 00
Ihlseng's Manual of Mining.	8vo, 5 00	
* Iles's Lead Smelting.	12mo, 2 00	
Peele's Compressed Air Plant for Mines.	8vo, 3 00	
Riemer's Shaft Sinking Under Difficult Conditions. (Corning and Peele.)	8vo, 3 00	
* Weaver's Military Explosives	8vo, 3 00	
Wilson's Hydraulic and Placer Mining. 2d edition, rewritten.....	12mo, 2 50	
Treatise on Practical and Theoretical Mine Ventilation	12mo, 1 25	

SANITARY SCIENCE.

Association of State and National Food and Dairy Departments, Hartford Meeting, 1906.	8vo, 3 00	
Jamestown Meeting, 1907.	8vo, 3 00	
* Bashore's Outlines of Practical Sanitation.	12mo, 1 25	
Sanitation of a Country House.	12mo, 1 00	
Sanitation of Recreation Camps and Parks.	12mo, 1 00	
* Chapin's The Sources and Modes of Infection.....	Large 12mo, 3 00	
Folwell's Sewerage. (Designing, Construction, and Maintenance.)....	8vo, 3 00	
Water-supply Engineering.	8vo, 4 00	
Fowler's Sewage Works Analyses.	12mo, 2 00	
Fuertes's Water-filtration Works.	12mo, 2 50	
Water and Public Health.	12mo, 1 50	
Gerhard's Guide to Sanitary Inspections.....	12mo, 1 50	
* Modern Baths and Bath Houses.....	8vo, 3 00	
Sanitation of Public Buildings.....	12mo, 1 50	
* The Water Supply, Sewerage, and Plumbing of Modern City Buildings.	8vo, 4 00	
Hazen's Clean Water and How to Get It.	Large 12mo, 1 50	
Filtration of Public Water-supplies.	8vo, 3 00	
* Kinnicutt, Winslow and Pratt's Sewage Disposal.....	8vo, 3 00	
Leach's Inspection and Analysis of Food with Special Reference to State Control.	8vo, 7 50	
Mason's Examination of Water. (Chemical and Bacteriological)....	12mo, 1 25	
Water-supply. (Considered principally from a Sanitary Standpoint).	8vo, 4 00	
* Mast's Light and the Behavior of Organisms.....	Large 12mo, 2 50	
* Merriman's Elements of Sanitary Engineering.	8vo, 2 00	
Ogden's Sewer Construction	8vo, 3 00	
Sewer Design.....	12mo, 2 00	
Parsons's Disposal of Municipal Refuse.....	8vo, 2 00	
Prescott and Winslow's Elements of Water Bacteriology, with Special Reference to Sanitary Water Analysis.	12mo, 1 50	
* Price's Handbook on Sanitation.	12mo, 1 50	
Richards's Conservation by Sanitation.....	8vo, 2 50	
Cost of Cleanness.	12mo, 1 00	
Cost of Food. A Study in Diets.	12mo, 1 00	
Cost of Living as Modified by Sanitary Science.	12mo, 1 00	
Cost of Shelter.	12mo, 1 00	
* Richards and Williams's Dietary Computer.	8vo, 1 50	
Richards and Woodman's Air, Water, and Food from a Sanitary Standpoint.	8vo, 2 00	
* Richey's Plumbers', Steam-fitters', and Tinnerns' Edition (Building Mechanics' Ready Reference Series).	16mo, mor. 1 50	
Rideal's Disinfection and the Preservation of Food.	8vo, 4 00	
Sewage and Bacterial Purification of Sewage.	8vo, 4 00	
Soper's Air and Ventilation of Subways.....	12mo, 2 50	
Turneaure and Russell's Public Water-supplies.....	8vo, 5 00	
Venable's Garbage Crematories in America.	8vo, 2 00	
Method and Devices for Bacterial Treatment of Sewage.....	8vo, 3 00	
Ward and Whipple's Freshwater Biology. (In Press.)		
Whipple's Microscopy of Drinking-water.....	8vo, 3 50	
* Typhoid Fever.	Large 12mo, 3 00	
Value of Pure Water.....	Large 12mo, 1 00	
Winslow's Systematic Relationship of the Coccaceae.....	Large 12mo, 2 50	

MISCELLANEOUS.

* Chapin's How to Enamel	12mo.	\$1 00
Emmons's Geological Guide-book of the Rocky Mountain Excursion of the International Congress of Geologists.	Large 8vo.	1 50
Ferrel's Popular Treatise on the Winds.	8vo.	4 00
Fitzgerald's Boston Machinist.	18mo.	1 00
Gannett's Statistical Abstract of the World.	24mo.	75
Haines's American Railway Management.	12mo.	2 50
Hanausek's The Microscopy of Technical Products. (Winton).	8vo.	5 00
Jacobs's Betterment Briefs. A Collection of Published Papers on Or- ganized Industrial Efficiency.	8vo.	3 50
Metcalf's Cost of Manufactures, and the Administration of Workshops. 8vo.		5 00
Putnam's Nautical Charts.	8vo.	2 00
Ricketts's History of Rensselaer Polytechnic Institute 1824-1894.	Large 12mo.	3 00
Rotherham's Emphasised New Testament.	Large 8vo.	2 00
Rust's Ex-Meridian Altitude, Azimuth and Star-finding Tables.	8vo.	5 00
Standage's Decoration of Wood, Glass, Metal, etc.	12mo.	2 00
Thome's Structural and Physiological Botany. (Bennett).	16mo.	2 25
Westermaier's Compendium of General Botany. (Schneider).	8vo.	2 00
Winslow's Elements of Applied Microscopy.	12mo.	1 50

HEBREW AND CHALDEE TEXT-BOOKS.

Gesenius's Hebrew and Chaldee Lexicon to the Old Testament Scriptures. (Tregelles.)	Small 4to, half mor.	5 00
Green's Elementary Hebrew Grammar.	12mo.	1 25

